



Efficient and enantioselective total syntheses of heliannuols A and K

Makoto Kanematsu, Kana Soga, Yuki Manabe, Sachie Morimoto, Masahiro Yoshida, Koza Shishido*

Graduate School of Pharmaceutical Sciences, The University of Tokushima, 1-78-1 Sho-machi, Tokushima 770-8505, Japan

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ABSTRACT

The second-generation enantioselective synthesis of heliannuol A and the first enantioselective total synthesis of heliannuol K (via two routes) have both been accomplished efficiently; (heliannuol A, nine steps and 25% yield; heliannuol K, seven steps and 47% yield). Highlights of our synthetic strategy include a substrate-controlled chirality transfer in the Lewis acid mediated Claisen rearrangement of the allyl aryl ether for the key construction of a tertiary stereogenic center at the benzylic position followed by, for heliannuol A, ring-closing metathesis, diastereoselective epoxidation, and regioselective cleavage of the epoxide; and for heliannuol K, ring-closing metathesis and conjugate reduction of the eight-membered enone.

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1. Introduction

Heliannuol A (**1**),¹ the first member of a class of bioactive sesquiterpenes, and heliannuol K (**2**)² were isolated from the moderately polar bioactive fractions of the fresh leaf aqueous extracts of the cultivar *Helianthus annuus* L. var. SH-222 in 1993 and in 1999, respectively, by Macías et al. These natural products have been reported to exhibit significant allelopathic activity against several dicotyledon (*Lactuca sativa* and *Lepidium sativum*) and monocotyledon species (*Hordeum vulgare* and *Triticum aestivum*) at concentrations of 10^{-4} – 10^{-9} M.³ The structure of **1**, including the relative stereochemistry, was determined by single crystal X-ray diffraction analysis. This intriguing compound contains an aryl ring fused to the eight-membered oxygen heterocycle with two tertiary stereogenic centers at the C7 and C10 positions. Its absolute structure was established to be (7*R*,10*S*) by our enantioselective total synthesis of the unnatural enantiomer of (+)-heliannuol A,^{4a} while the structure of heliannuol K (**2**) was elucidated mainly by ¹H NMR and found to have the same carbon framework as heliannuol A except for the presence of a C10 carbonyl function. Although the absolute stereochemistry at C7 has never been established, it can be deduced to be *R* based on the biogenetic parallelism with heliannuol A. Because of the interesting structural features and phytotoxic properties of these compounds, several reports on the total synthesis of the heliannuols A (**1**) and K (**2**) have been published.⁵ During the course of our studies directed toward the enantioselective synthesis of helianane-type terpenoids with allelopathic activity,⁶ we reported the first enantioselective total

synthesis of the natural enantiomer (–)-heliannuol A (**1**),^{4b} which was completed in seventeen steps from 2,5-dimethoxy-4-methyl-iodobenzene with an overall yield of 5%. The obvious challenge was to improve both the low yield and the many reaction steps. Herein we describe the efficient and enantioselective total syntheses of heliannuols A (**1**) and K (**2**) (Fig. 1).

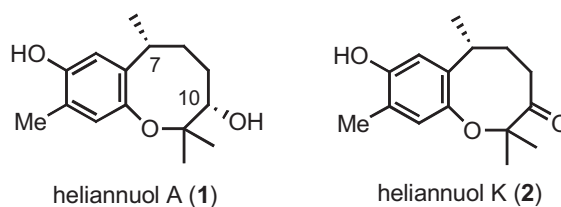


Fig. 1. Structures of heliannuols A and K.

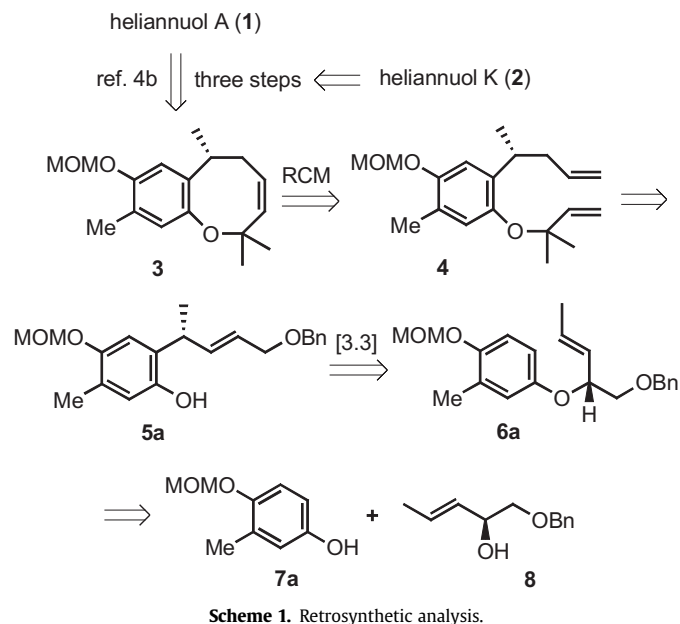
2. Results and discussion

2.1. The second-generation synthesis of (–)-heliannuol A (**1**) and the first enantioselective synthesis of (–)-heliannuol K (**2**)

Our retrosynthetic analysis of **1** is shown in Scheme 1. For the synthesis of **1**, we chose the dihydrobenzoxocine **3**⁷ with a stereogenic center at C7 as the key compound, because it has been converted efficiently to **1** by sequential diastereoselective epoxidation, regioselective cleavage of the epoxide, and deprotection of the phenolic hydroxyl function in good overall yield.^{4b} The MOM protected heliannuol A, the penultimate intermediate of **1**, would be transformed to heliannuol K (**2**) by a simple two-step sequence; oxidation followed by deprotection. The eight-membered heterocycle fused to the aryl ring can be assembled by ring-closing

* Corresponding author. Tel.: +81 88 6337287; fax: +81 88 6339575; e-mail address: shishido@ph.tokushima-u.ac.jp (K. Shishido).

metathesis of the diene **4**, which would be prepared from the phenol **5a** via Pd-catalyzed dimethylallyl etherification.^{4b,8} For the key construction of the tertiary stereogenic center at the benzylic position, we planned to use a substrate-controlled chirality transfer in the Claisen rearrangement^{9,10} of the allyl aryl ether **6a** prepared from the phenol **7a** and *S*-(*E*)-1-(benzyloxy)pent-3-en-2-ol (**8**)¹¹ by the Mitsunobu coupling¹² (Scheme 1).



The Mitsunobu reaction between 4-(methoxymethoxy)-3-methylphenol (**7a**), prepared from 3-hydroxy-4-methylacetophenone in two steps, and the allylic alcohol **8**, derived from *S*-(+)-benzyl glycidyl ether in two steps, in the presence of 1,1-(azodicarbonyl)dipiperidine (ADDP)¹³ and ⁿBu₃P provided a chromatographically separable mixture of the requisite ether **6a** (>99% ee; by HPLC analysis using a Chiralcel OD column) and the regioisomer **9a**, which was derived via the S_N2' process, in 81% and 7% yield, respectively. The key Claisen rearrangement of **6a** was examined and the results are shown in Table 1. Treatment of

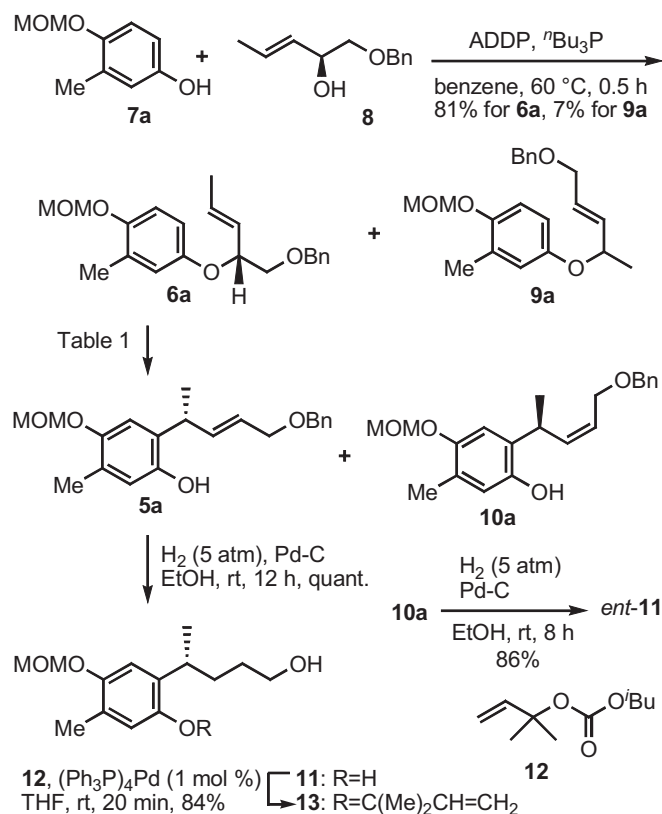
Table 1
Claisen rearrangement of **6a**

Entry	Conditions	Yield, %		(% ee) ^a
		5a	10a	
1	Eu(fod) ₃ (8 mol %) ClCH ₂ CH ₂ Cl, 90 °C, 24 h	92 (90)	7	
2	Et ₂ AlCl (1.5 equiv), hexane 0 °C, 4 h	27 (>99)	5	
3	Me ₃ Al (3 equiv), hexane rt, 0.5 h	85 (>99)	4	

^a HPLC (Chiralcel AD column).

a solution of **6a** in dichloroethane with 8 mol % of Eu(fod)₃^{9d} at 90 °C for 24 h gave a separable mixture of *R*-**5a** and the *Z*-isomer **10a** with the *S* configuration in 92% and 7% yield, respectively. However, the enantiomeric excess of **5a** was 90% (determined by HPLC analysis using a Chiralcel AD column) (entry 1). When the reaction was conducted with 1.5 equiv of Et₂AlCl in hexane at 0 °C for 4 h, the enantiomerically pure **5a** was obtained in only 27% yield along with 5% of **10a** (entry 2). The optimized reaction conditions call for 3 equiv of Me₃Al¹⁴ in hexane at room temperature for 0.5 h; the requisite **5a** was obtained in 85% yield (>99% ee), together with **10a** (4%) (entry 3). On exposure of **5a** to hydrogenation conditions, the phenolic alcohol **11** {[α]_D²⁷ –17.3 (c 2.65, CHCl₃)} was obtained

quantitatively. As for the absolute configuration of the stereogenic center in **10a**, it was determined to be *S* by hydrogenation providing *ent*-**11** {[α]_D²⁷ +18.0 (c 1.48, CHCl₃)} in 86% yield. Treatment of **11** with the mixed carbonate **12** in the presence of catalytic (Ph₃P)₄Pd (1 mol %)⁸ provided the alkenyl alcohol **13** in 84% yield (Scheme 2).



The alcohol **13** was exposed to the dehydration protocol of Nishizawa–Grieco¹⁵ to give the diene **4** (R=MOM), which was identical with the authentic material prepared previously, in 76% yield for the two steps. It was then treated with the Grubbs' second-generation catalyst **14** (0.5 mol %)¹⁶ in refluxing methylene chloride to give the dihydrobenzoxocine **3** in 93% yield. Attempts at a substrate-controlled diastereoselective epoxidation of **3** using *m*CPBA provided a lower yield (73%) of the product **15**. However reaction of **3** with methyl(trifluoromethyl)dioxirane¹⁷ generated in situ from methyl trifluoromethyl ketone and Oxone[®] in the presence of Na₂-EDTA·2H₂O and sodium hydrogen carbonate in acetonitrile provided only the epoxide **15** in 83% yield. The stereochemistry was confirmed by the observation of a distinct NOE correlation between the Ha (δ 3.11) and Hb (δ 2.55) protons. LiAlH₄ reduction of the epoxide occurred at the sterically less congested carbon (C9) regioselectively to give the alcohol **16** in 91% yield as a single product. Finally, acidic hydrolysis of the MOM ether produced heliannuol A (**1**), whose spectroscopic (¹H and ¹³C NMR) properties as well as optical rotation, {[α]_D²⁶ –78.0 (c 2.4, MeOH) [lit.^{1a} [α]_D –55.4 (c 0.3, MeOH)]}, were identical with those of the natural product. Thus, the second-generation enantioselective total synthesis of heliannuol A (**1**) has been accomplished in a longest linear sequence of nine steps in 25% yield from the phenol **7a**. For the synthesis of heliannuol K (**2**), the penultimate intermediate **16** was oxidized with Dess–Martin periodinane to give the ketone **17**, which was exposed to the acidic hydrolysis conditions producing **2** in 99% yield for the two steps. The spectroscopic properties (¹H and

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