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Total synthesis of calothrixins and their analogues via formal [3+2] cycloaddition of arynes and 2-aminophenanthridinedione



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ABSTRACT

Bioactive indolo[3,2-j]phenanthridine alkaloids, calothrixin A, B, and their analogues have been synthesized via formal cycloaddition of arynes and 2-aminophenanthridinedione as the key step, which proceeds through successive C-C/C-N bond formation and subsequent oxidation under transition-metal-free and mild conditions in the final stage of the synthesis.

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Calothrixin A (1a) and B (2a) are quinone-based natural products isolated from cell extracts of *Calothrix* cyanobacterium in 1999. They contain indolo[3,2-j]phenanthridine framework with a unique assembly of highly important pharmacophores including indole, quinoline, and quinone (Fig. 1). These compounds act as human DNA topoisomerase I poisons, inhibiting in vitro growth of chloroquine-resistant strain of the human parasite *Plasmodium falciparum* and also exhibiting lethal effects on human cancer cell lines at low nano-molar concentrations.²

Owing to their unique indolo[3,2-j]phenanthridine scaffold and promise as lead compounds for drug discovery, great efforts have been made for the total synthesis of calothrixins. Several approaches have been reported for their synthesis,³ utilizing strategies such as ortho-lithiation, 4 Pd, or Cu-catalyzed coupling, electrocyclization,⁶ Friedel-Crafts acylation/alkylation,⁷ and radical reactions.⁸ These strategies rely upon functionalized indole, quinoline, or carbazole as synthetic precursors, but a greater challenge lies in developing a practical and flexible synthetic strategy for generating a library of calothrixin analogues, which would be invaluable for structure-activity relationship (SAR) studies. Cascade reactions based on arynes is a powerful tool for organic synthesis and have been used in many total synthesises of natural products. Especially, the introduction of 2-(trimethylsilyl)-aryl triflates as aryne precursors allows the reaction to proceed at transition-metal-free and mild condition.¹⁰ Herein, we report a simple and practical approach for the synthesis of calothrixins based on cascade reaction of arynes, which has the flexibility to modulate the functional groups on ring A at the final stage of their synthesis.

As a part of our research program on the synthesis of bioactive heterocycles and their analogues employing aryne chemistry, ¹¹ calothrixins were identified as important synthetic targets. Based on the knowledge of reaction between aryne and 2-aminoquinone, ¹¹ their retrosynthetic analysis is outlined in Scheme 1. It was hypothesized that ring A of calothrixin B (**2a**) could be prepared via a formal [3+2] cycloaddition reaction of aryne **3a**′ with 2-aminophenanthridinedione **4**, and variable aryne precursors could also be used in the final stage to provide diverse analogues of calothrixin B. Intermediate amine **4** could be obtained by oxidation of compound **5**, which in turn could be achieved by the reductive amination of aldehyde **7** with *o*-iodoaniline (**6**) followed by palladium-catalyzed intramolecular coupling.

Our investigation commenced with the synthesis of key intermediate **4** as shown in Scheme 2. Regioselective bromination of commercially available 2,5-dimethoxyaniline (**8**) followed by the protection of primary amine with (Boc)₂O provided compound **10** in good yields. Bromide **10** readily underwent lithium halogen exchange, and quenching of the resulting anion with DMF gave the corresponding aldehyde **7**. Reductive amination of aldehyde **7** with *o*-iodoaniline (**6**) using NaCNBH₃ and acetic acid provided amine **11** in 82% yield. Be Acetylation of compound **11** with Ac₂O in the presence of catalytic amount of DMAP, followed by intramolecular palladium-catalyzed cyclization afforded compound **5** in 85% yield. Compound **5** on treatment with CAN in CH₃CN/H₂O (2:1) furnished phenanthridinedione **13** in 64%

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Figure 1. Structures of calothrixins.

Scheme 1. Retrosynthetic analysis of calothrixin B.

yield. ¹² Deprotection of Boc with TFA provided 2-aminophenan-thridinedione **4** in quantitative yield.

With the required key intermediate **4** and aryne precursors **3a**-**3d** (Fig. 2) (synthesized following the reported procedures)¹³ in hand, the formal [3+2] cycloaddition for constructing ring A of calothrixin B was explored. We systematically examined the effect of solvent (THF, Et₂O, CH₃CN, CH₂Cl₂, and TolH), fluoride source (KF, CsF, TBAF, and TBAT), temperature (0 °C, 25 °C, and 60 °C) and additive (18-crown-6) (see Supporting Information, Table S1). When TBAT was used as fluoride source and THF as reaction solvent at room temperature,¹¹ calothrixin B (**2a**) was obtained in 26% yield (43% brsm), along with *N*-arylated **14** (5%),

Figure 2. Aryne precursors 3a-3d.

Scheme 3. Preliminary results of formal [3+2] cycloaddition. Conditions: **3a** (0.64 mmol), **4** (0.4 mmol), TBAT (1.28 mmol), THF (8.0 mL), N₂, 25 °C.

uncyclized C-arylated 15 (4%), and other several unidentifiable by-products (Scheme 3). Based on the literature precedents, the nitrogen atom in quinoline moiety could induce side reaction by reacting with aryne,14 which might account for the low yield of the desired product. It is noteworthy that other fluoride sources (KF and CsF) led to decrease in desired product 2a and significant increase of side products such as 14 and 15. Oxidation of calothrixin B with mCPBA in CH₂Cl₂ afforded calothrixin A in 71% yield (Table 1). Utilizing the same strategy, the synthesis of calothrixin analogues was conducted. Under the optimized reaction condition, key intermediate 4 was subjected to transition-metal-free cascade reaction with symmetric and unsymmetrical aryne precursors (3b-3d) to provide calothrixin B analogues (2b-2d) in moderate yields. Calothrixin A analogues (1b-1d) were obtained in good yield using the same oxidizing procedure with 1a, and the results are summarized in Table 1.

When unsymmetrical aryne precursor **3c** was used as the aryne source, calothrixin analogue **2c** was obtained regioselectively. To establish the regioselectivity of the reaction, compound **1c** was treated with CH₃I/NaH, followed by quenching the reaction mixture with methanol, leading to the formation of derivative **17** in 88% yield, presumably through intermediate **16**. Structure of compound **17** was confirmed by NOE correlations (Scheme 4).

Two plausible mechanisms for this cascade reaction are depicted in Scheme 5. In Mechanism I, quinone amine nitrogen

Scheme 2. Synthesis of aminophenanthridinedione 4.

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