



Organocatalytic activity of α,α -dipeptide derivatives of (S)-proline in the asymmetric aldol reaction in absence of solvent. Evidence for non-covalent π – π interactions in the transition state



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ABSTRACT

The trend in the magnitude of stereoselectivities observed in the asymmetric aldol reaction between cyclohexanone and aromatic aldehydes with varying electron densities, catalyzed by dipeptidic organocatalyst (S,S)-**1d**, which contains an aromatic naphthyl substituent in the acyclic amino acid residue, is in line with expectation based on π – π stacking interactions between the aromatic ring on the organocatalyst and the aromatic aldehydes. Such non-covalent interactions are apparently enhanced under solvent-free mechanochemical activation in a ball mill.

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Introduction

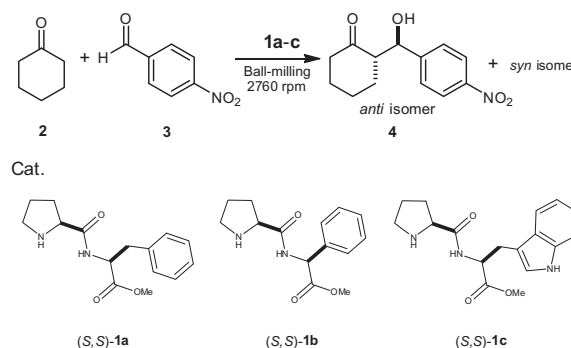
The area of organocatalysis has grown exponentially since the realization that small organic molecules can catalyze organic reactions with efficiency comparable to that exhibited by enzymes.¹ In particular, proline and proline derivatives have been widely studied as chiral organocatalysts in asymmetric organic transformations.² Application of these organocatalysts in asymmetric aldol reactions is of great importance both in academics and in the pharmaceutical industry.³ Several research groups have recently reported the successful application of dipeptide and tripeptide derivatives containing the proline residue as chiral organocatalysts in asymmetric reactions.⁴

In this context, we recently examined the aldol reaction between representative ketones and various aromatic aldehydes in the presence of organocatalysts (S,S)-**1a–c**.⁵ The reaction was carried out at -20°C in a ball mill at 2760 rpm, in the presence of 1.1 equiv H_2O , and PhCO_2H (5 mol %) as additive. After 6 h of milling, all six catalysts **1a–c** afforded the aldol product in good yields, high *anti* diastereoselectivity and high enantioselectivity (Table 1).

Best results were generated with dipeptide (S,S)-**1c** as catalyst, providing the aldol product in 88% yield, a diastereomeric ratio

Table 1

Direct asymmetric aldol reaction of cyclohexanone with 4-nitrobenzaldehyde catalyzed by dipeptides (S,S)-**1a–c**^a



Entry	Cat.	Yield ^b (%)	dr (<i>anti</i> / <i>syn</i>) ^c	ee ^d (%)
1	1a	89	93:7	94
2	1b	79	91:9	85
3	1c	88	92:8	>98

^a Reaction conditions: cyclohexanone (**2**, 0.22 mmol), 4-nitrobenzaldehyde (**3**, 0.20 mmol), catalyst **1a–c** (3 mol %), -20°C , 1.1 equiv H_2O , PhCO_2H (5 mol %), 6.0 h. Best values are highlighted in bold.

^b Isolated yield.

^c Determined by ^1H NMR of the crude product.

^d Determined by chiral HPLC.

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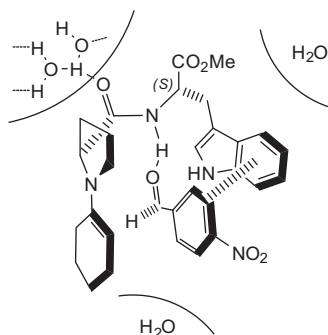
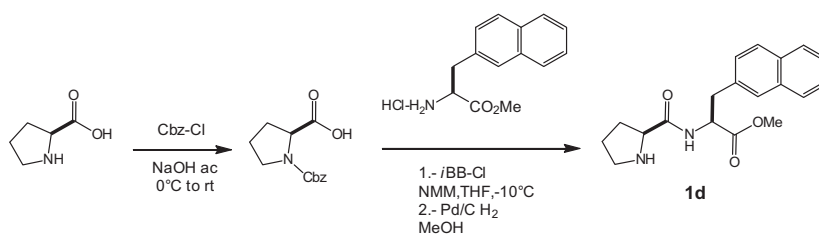


Figure 1. Proposed transition state model of the aldol reaction catalyzed by (S,S)-**1c**.

of 92:8 in favor of the *anti* isomer and higher than 98% ee of aldol (2*S*,1'*R*)-**4**.

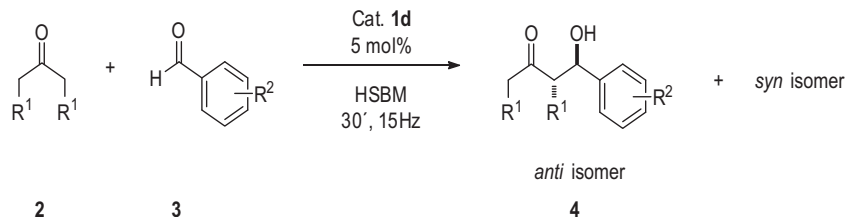
To account for the observed stereoselectivity, the transition state presented in Figure 1 was suggested, which takes into account previously advanced mechanistic proposals where catalysis by dipeptides and prolinamides is operative.⁶ In particular, the pyrrolidine functionality in (S,S)-**1c** activates the ketone through the formation of a chiral enamine intermediate, while the aldehyde is activated through the formation of a strong hydrogen bond involving the amide N–H hydrogen atom.⁷ In addition, it was proposed that a non-covalent π – π interaction between aromatic rings of the catalyst and the aldehydes probably gives rise to a rigid transition state that induces higher stereoselectivity in the aldol reaction.⁸



Scheme 1. Synthesis of organocatalyst (S,S)-**1d**.

Table 2

Efficiency of organocatalyst (S,S)-**1d** in the asymmetric aldol reaction of cyclohexanone as representative ketone with aromatic aldehydes of different electron densities^a



Entry	Product	Yield ^b (%)	dr (<i>anti</i> / <i>syn</i>) ^c	ee ^d (%)
1	 4a	99	99:1	99:1
2	 4b	94	96:4	95:5
3	 4c	98	95:5	89:11
4	 4d	98	98:2	95:5

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