Tetrahedron Letters 57 (2016) 317-320

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

An expedient synthesis of *N*,*N*-dialkylamino-dihydroxanthenepyrylium conjugated near-infrared fluorescent dyes



Anthony Romieu^{b,c}, Jean-Alexandre Richard^{a,*}

^a Organic Chemistry, Institute of Chemical and Engineering Sciences (ICES), Agency for Science, Technology and Research (A*STAR), 8 Biomedical Grove, Neuros, #07-01, Singapore 138665, Singapore

^b ICMUB, UMR 6302, CNRS, Univ. Bourgogne Franche-Comté, 9, Avenue Alain Savary, 21078 Dijon cedex, France ^c Institut Universitaire de France, 103 Boulevard Saint-Michel, 75005 Paris, France

ARTICLE INFO

Article history: Received 19 September 2015 Revised 29 November 2015 Accepted 2 December 2015 Available online 7 December 2015

Keywords: Cascade reactions NIR dyes (Nor)dihydroxanthenes Pyrylium dyes

ABSTRACT

An expedient synthesis of amino-(nor)dihydroxanthene-based fluorophores is reported, proceeding in two steps from commercially available salicylic aldehydes. The synthesis relies on the one-pot construction of the (nor)dihydroxanthene scaffolds through an oxa-Michael/retro-Michael/vinylogous aldol/dehydration cascade sequence. Further extension of the π -conjugated systems through a condensation reaction with a 2,4,6-trisubstituted pyrylium salt led to a novel class of near-infrared (NIR) fluorescent dyes with absorption/emission maxima in polar media in the ranges of 705–778 and 785–828 nm, respectively.

© 2015 Elsevier Ltd. All rights reserved.

Introduction

The synthesis of near-infrared (NIR) absorbing/emitting small molecules¹ is of tremendous interest because of the increasing number of applications in optical bioimaging,² photoacoustic tomography³ and, more broadly, in medical diagnostics⁴ and theranostics.⁵ NIR-emitting dyes offer important advantages in vivo in terms of sensitivity (e.g., deeper tissue penetration, low autofluorescence and background noise compared to the visible range), safety (e.g., noninvasiveness due to the lack of exposure to ionizing radiations), user-friendliness and affordable cost.⁶ While the more widely used fluorophores⁷ emitting in the therapeutic window (700–900 nm)⁸ of the electromagnetic spectrum contain heptamethine cyanine $(Cy7)^{9,10}$ π -extended BODIPY and aza-BODIPY dyes,^{11,12} we have noticed the recent emergence of modified xanthene-based fluorophores which, in our view, would beneficially complete the repertoire of NIR fluorescent dyes currently available. The dihydroxanthene scaffold was first reported by Grummt and Czerney¹³ in 1996 and displayed an interesting emission above 730 nm. The initially developed chemistry was revisited in 2012 by Lin and co-workers who described analogs of rhodamine B which combined the stability of rhodamines (e.g., marked photoand chemical-stability, especially under harsh conditions of pH and temperature) and the easy modulation of their fluorescence properties through a spirocyclic/open-ring switching mechanism¹⁴ with the far-emission usually observed for Cy7 dyes (Fig. 1A).¹⁵ During the last few years, phenolic dihydroxanthene dyes have been extensively used in the design of reaction-based fluorescent probes for chemoselective bioimaging by taking advantage of their optically tunable hydroxyl group.¹⁶ Conversely, less attention has been paid to the synthesis and bioanalytical applications of their *N*,*N*-dialkylamino counterparts.^{16h,17} A cyanine-degradationbased approach was recently reported by Lin and co-workers¹⁸ but a short, high yielding *de novo* access to the *N*,*N*-dialkylamino dihydroxanthene scaffold avoiding the destruction of valuable NIR cyanine dyes is desirable in order to provide more flexibility to access this promising fluorescent scaffold.¹⁹

We have previously reported a streamlined synthesis of phenolic dihydroxanthene fluorophores thanks to the construction of the dye skeleton through an oxa-Michael/retro-Michael/vinylogous aldol/dehydration cascade sequence.²⁰ The newly developed *de novo* synthesis allowed access to phenolic near-infrared emitting dyes in three steps and paved the way for an expedient synthesis of fluorogenic probes derived from this 'smart' fluorescent phenol. We then wondered whether the modularity of this approach could be used to rapidly synthesize *N,N*-dialkylamino-dihydroxanthene dyes, aimed at assessing their photophysical properties in the NIR spectral range.

Herein, we report the convenient preparation of π -extended pyrylium-*N*,*N*-dialkylamino-dihydroxanthene chimera along with





^{*} Corresponding author. Tel.: +65 6799 8521; fax: +65 6464 2102. *E-mail address: jean_alexandre@ices.a-star.edu.sg* (J.-A. Richard).





Figure 1. (A) Rhodamine B analog featuring an extended π -conjugated system; (B) (nor)dihydroxanthene scaffolds and (C) retrosynthetic analysis.

the previously unreported nordihydroxanthene scaffold (Fig. 1B). The spectral properties of these long-wavelength xanthene-like fluorophores were evaluated in polar media.

Results and discussion

The desired *N*,*N*-dialkylamino (nor)dihydroxanthene dyes I featuring an electron acceptor moiety at one end of the π -conjugated system could be obtained by a Knoevenagel condensation of key intermediate aldehydes II (Fig. 1C). We hypothesized the latter would be derived from readily available starting materials salicylic aldehydes **V** and β -bromoenal IV where the size of the ring could be varied in order to assess its effect on the fluorescence properties. Mechanistically, we previously proposed,²⁰ in the presence of base, salicylic aldehydes **V** would form a phenolate moiety that undergoes an oxa-Michael/retro Michael reaction leading to aryl ether III. Subsequent vinylogous aldol reaction followed by a final dehydration would close the central ring leading to the desired aldehyde II featuring the nordihydroxanthene (*n* = 1) and dihydroxanthene (*n* = 2) skeletons.

In order to test that hypothesis, we used commercially available 9-formyl-8-hydroxyjulolidine **1** and 4-(*N*,*N*-diethylamino)salicylaldehyde **2** as *ortho*-hydroxy benzaldehydes **V**, reacting with 5membered and 6-membered β -bromoenals **3** and **4**. Applying the optimized conditions previously found for the synthesis of the phenolic dihydroxanthene scaffold (3 equiv of Cs₂CO₃ in DMF) we were pleased to observe that nordihydroxanthene aldehydes **5a** and **5b** could be isolated in 65% and 50% yield. Similar results were obtained using 6-membered-ring β -bromoenal **4**, and aldehydes **5c** and **5d** were synthesized in 50% and 62% yield, respectively (Scheme 1).

With these key intermediates in hand, the stage was set for their condensation to positively charged heterocycles bearing an



Scheme 1. Synthesis of (nor)dihydroxanthene aldehyde intermediates **5a–5d** through the oxa-Michael/retro-Michael/vinylogous aldol/dehydration cascade.

activated methyl aimed at extending their π -conjugated system and providing NIR fluorophores. Furthermore, the presence of a donor–acceptor pair (i.e., *N*,*N*-diethylamino and heteroonium) within the same molecular scaffold should promote internal charge transfer (ICT) and a dramatic red-shift in the absorption/ emission maxima. Due to their valuable chemical reactivity and photophysical properties, pyrylium salts (including 4-methyl derivatives) are suitable synthetic building blocks for achieving this goal.²² Thus, condensation of aldehydes **5a–5d** with commercial 2,6-di-*tert*-butyl-4-methylpyrilium triflate **6** in the presence of K₂CO₃ in Ac₂O at room temperature successfully provided dyes **7a– 7d** with yields in the range 45–82% (Fig. 2A). The conversion of the aldehyde precursors **5a–5d** to the desired dyes appeared to be high by TLC, and we believed that the moderate isolated yields obtained

A. Synthesis of far-emitting amino dihydroxanthene pyrylium fluorophores



Figure 2. (A) Synthesis of *N*,*N*-dialkylamino (nor)dihydroxanthene pyrylium dyes **7a–7d** from aldehydes **5a–5d** and (B) known far-emitting fluorophores from Lin and co-workers.

Download English Version:

https://daneshyari.com/en/article/5261310

Download Persian Version:

https://daneshyari.com/article/5261310

Daneshyari.com