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Synthesis of 2,9-dialkylated phenanthro[1,2-*b*:8,7-*b'*]dithiophenes via cross-coupling reactions and sequential Lewis acid-catalyzed regioselective cycloaromatization of epoxide

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ABSTRACT

Phenanthro[1,2-*b*:8,7-*b'*]dithiophene (PDT) was prepared via Suzuki–Miyaura or Negishi cross-coupling of a 2-thienylboron or -zinc compound with 1,4-dibromobenzene, followed by Lewis acid-catalyzed regioselective cycloaromatization of the epoxide. A series of 2,9-dialkylated phenanthro[1,2-*b*:8,7-*b'*]dithiophene (PDT) derivatives could also be synthesized in good yields by Suzuki–Miyaura coupling of the brominated PDT with alkylboranes by introducing linear alkyl substituents.

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Organic field-effect transistors (OFETs) have attracted considerable interest as key components in future ubiquitous electronics due to advantages such as flexibility, light weight, and ease of design.¹ One particular acene-type molecule, pentacene,² has served as the active semiconducting layer in OFETs owing to the high field-effect mobility (μ) of $5.5 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ that it has shown in a thin-film transistor,³ and its state-of-the-art value of $40 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ in single crystals.⁴ However, pentacene is unstable under atmospheric conditions, and readily photodegrades owing to its relatively high HOMO energy (-5.0 eV), which arises from its extended π -conjugation.⁵ Recently, a phenacene-type molecule, picene, incorporating the same number of benzene rings as pentacene, has become the focus of considerable interest because it becomes superconductive⁶ with alkali-metal doping, and also shows high field-effect mobility in a transistor ($1.1 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ in a thin-film OFET).⁷ Moreover, a picene-based FET is stable in air because picene has a larger energy bandgap ($E_g = 3.3 \text{ eV}$) and a lower HOMO energy (-5.5 eV) than pentacene.⁸ The potential utility of phenacene-type molecules makes the development of more efficient synthetic methods⁹ and further improvement in their OFET properties^{10–12} matters of some importance.

We have recently reported the synthesis of picenes¹³ and fulminene¹⁴ by the palladium-catalyzed Suzuki–Miyaura coupling

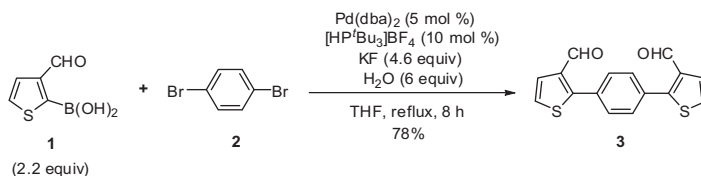
of (*Z*)-alkenylboronates with polyhalobenzene and sequential intramolecular double cyclization via C–H activation. This protocol is also applicable to the synthesis of phenanthro[1,2-*b*:8,7-*b'*]dithiophene (PDT) by replacing the terminal phenyl rings in picene with thiophene rings, aiming at increased intermolecular π – π interactions due to the large atomic radius of sulfur, which may enhance its performance in OFETs.¹⁵ Results showed that OFET devices fabricated with thin films of PDT formed by thermal deposition exhibited carrier mobility as large as $1.1 \times 10^{-1} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, and suggested that fabrication using a solution process might be possible owing to PDT's high solubility in common organic solvents.

However, we found that this synthetic strategy is not suitable for the large-scale synthesis of PDT because in some cases a mixture of stereoisomers of the coupled products is formed through (*E*)/(*Z*) isomerization upon Suzuki–Miyaura coupling, leading to a lower yield of the desired products. To produce derivatives of PDT for use as organic semiconductors, a more efficient synthetic method is highly desirable. Here we report a new synthetic route to the PDT core structure using cross-coupling reactions. Furthermore, the solubility of PDT might be improved by introducing long alkyl chains, as this may induce a self-assembly process by the 'fastener effect', leading to high crystallinity in thin films. Some examples of this have already been reported in alkyl-substituted picene¹⁶ and alkylated thienoacene.^{17,18}

To explore a new synthetic route to PDT, we investigated the palladium-catalyzed Suzuki–Miyaura coupling of commercially

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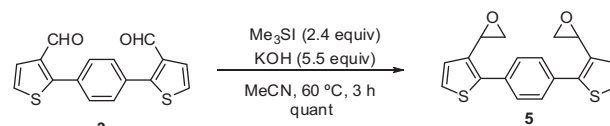
E-mail address: ynishih@okayama-u.ac.jp (Y. Nishihara).



Scheme 1.

available 3-formyl-2-thiopheneboronic acid (**1**) with 1,4-dibromobenzene (**2**) affording the corresponding coupled product **3** in 78% yield (Scheme 1).¹⁹

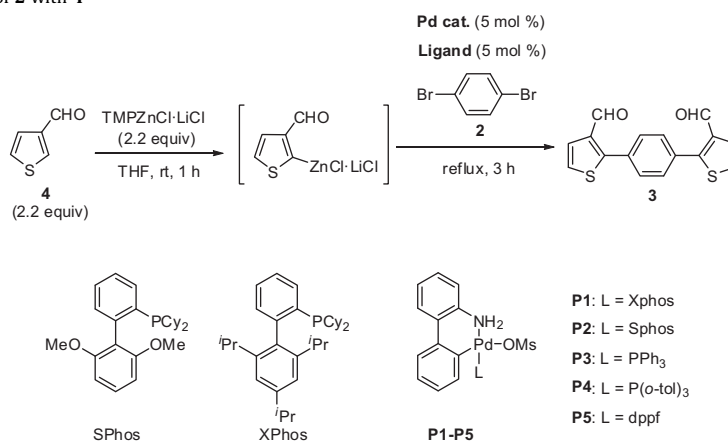
We screened various reaction conditions to develop an efficient synthetic route to **3**. The results are summarized in Table 1. To our surprise, we found that chemoselective C–H zincation across 3-formylthiophene (**4**) occurred at the 2-position (adjacent to the formyl group) of the thiophene ring, using TMPZnCl·LiCl.²⁰ Negishi coupling of the in situ generated 2-thienylzinc reagent with 1,4-dibromobenzene (**2**) using a catalyst system of Pd(dba)₂ (dba = dibenzylideneacetone) with a phosphonium salt, [HP^tBu₃]BF₄ used as a precursor of the phosphine ligand proceeded at reflux to furnish **3** in 65% yield (entry 1).²¹ However, other palladium precursors and phosphine-based ligands including biarylphosphine (Sphos)²² and PdCl₂(dppf)·C₆H₆²³ (dppf = 1,1'-bis(diphenylphosphino) ferrocene) were found to be inferior (entries 2–4). PEPPSI-IPr ((PEPPSI = pyridine-enhanced precatalyst preparation stabilization and initiation, IPr = 1,3-diisopropylimidazol-2-ylidene),²⁴ recently introduced by Organ, displayed a modest catalytic activity to afford **3** in 52% yield as determined by NMR (entry 5). With palladacycle precatalysts²⁵ utilized in the sp²–sp² Negishi couplings, compound **3** was obtained in lower yields (entries 6–10).



Scheme 2.

Following this, sequential epoxidation of **3** gave the desired product **5** quantitatively (Scheme 2).^{26,27} We next screened the reaction conditions of acid-mediated²⁸ and -catalyzed²⁹ Friedel–Crafts-type cycloaromatization of **5** and the results are summarized in Table 2. Attempted reactions with MeSO₃H and BF₃·OEt₂ did not proceed, even with the addition of excess reagent. With a stoichiometric amount of Sc(OTf)₃, PDT (**6**) was obtained in 32% yield (entry 1). However, the catalytic variant of Sc(OTf)₃ was not effective (entry 2). We then explored catalytic reactions with other Lewis acids M(OTf)_n, but yields of **6** were insufficient (entries 3–6). To our delight, 10 mol % of InCl₃ was found to give better results and afforded **6** in 46% yield (entry 7). Increasing the amount of InCl₃ to 20 mol % improved the yield to 50% (entry 8). Varying the concentrations, we found that lower concentrations gave

Table 1
Optimization of Negishi cross-coupling of **2** with **4**



Entry	Pd cat.	Ligand	Yield ^a (%)
1 ^b	Pd(dba) ₂	[HP ^t Bu ₃]BF ₄	65
2 ^b	Pd(OAc) ₂	Sphos	46
3 ^b	Pd(dba) ₂	P(<i>o</i> -tol) ₃	9
4	PdCl ₂ (dppf)·C ₆ H ₆	—	55
5	PEPPSI-IPr	—	52
6	P1	—	41
7	P2	—	36
8	P3	—	15
9	P4	—	3
10	P5	—	4

^a NMR yields based on **2**.

^b 10 mol % of phosphine ligand was used.

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