



Stereoselective synthesis of two potential metabolites of *cis*-metconazole



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ABSTRACT

The stereoselective synthesis of compounds **5** and **6**, potential hydroxylated metabolites of the agricultural fungicide *cis*-metconazole, is reported. In a key step of the initially surveyed synthetic route, hydrodechlorination of **12** was competitive with hydrogenation of the trisubstituted olefin. Application of a Miyaura borylation/hydrogenation/boron-to-halogen exchange reaction sequence solved the chemoselectivity issue.

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Introduction

Metconazole (*cis*-**1** and *trans*-**1**, Fig. 1A), belonging to the 1,2,4-triazole-containing fungicide family, is extensively used in agriculture to protect cereals, vegetables, and fruits against fungal infections.^{1–6} In fungi, **1** inhibits lanosterol 14 α -demethylase (CYP51A1), a cytochrome P450 enzyme engaged in the biosynthesis of ergosterol, a constituent of cell membranes which is essential for proper fungal growth.⁷ Numerous 1,2,4-triazoles have found applications as drugs and agrochemicals, e.g. tebuconazole (**2**), fluconazole (**3**), and voriconazole (**4**) (Fig. 1B).

Cytochrome P450 enzymes (CYPs) also play important roles in humans, including the synthesis of steroids and participation in the metabolism of xenobiotics. Similarly to other triazoles, **1** may inhibit CYPs, resulting in adverse effects such as hormonal and metabolic disorders.^{8–11} Moreover, long-term exposure to fungicides may be connected with the development of drug resistance for pathogenic fungi.^{12–15} Concerns regarding the impact of widespread agricultural applications of **1** on human health have arisen, and in order to address these questions, the biological activity of **1** as well as its metabolites must be studied more closely.¹⁶

Although, the synthesis of metabolites and metabolite-like derivatives of biologically active compounds have attracted significant attention,^{17–21} the preparation of metconazole metabolites

have not been reported; only the *in vitro* monooxygenation of **1** catalyzed by human CYP3A4, providing two hydroxylated products with undetermined structures, has been studied.²² Among the twenty C–H bonds in **1**, one tertiary and two benzylic bonds are the most electronically prone toward monooxygenation, although other bonds may also be reactive with specific CYPs since electronic factors are not the only determinants of enzymatic oxidation selectivity.²³

Herein, we disclose a concise synthesis of a pair of diastereomeric hydroxymetconazoles **5** and **6** (Fig. 1C), representing potential metabolites of *cis*-**1**, which is the more abundant metconazole isomer in the manufactured product.

Results and discussion

Our general strategy to access **5** and **6** is depicted in Scheme 1. We proposed that both compounds could be derived from commercially available **7** and the remaining structural motifs could be introduced using well-established chemistry. We also envisioned the relative stereochemistry between the cyclopentane ring substituents could be controlled by a bulky protecting group on C (3)-OH.

First, precursor **7** was alkylated with MeI to afford **8** in 36% yield.²⁴ Then, a single carbonyl group in **8** was selectively reduced using a substoichiometric amount of NaBH₄, affording alcohol **9** in 94% yield.²⁵ Cross-aldol condensation of alcohol **9** and 4-chlorobenzaldehyde followed by silylation of the hydroxyl group, furnished **11** in good overall yield. The Corey-Chaykovsky

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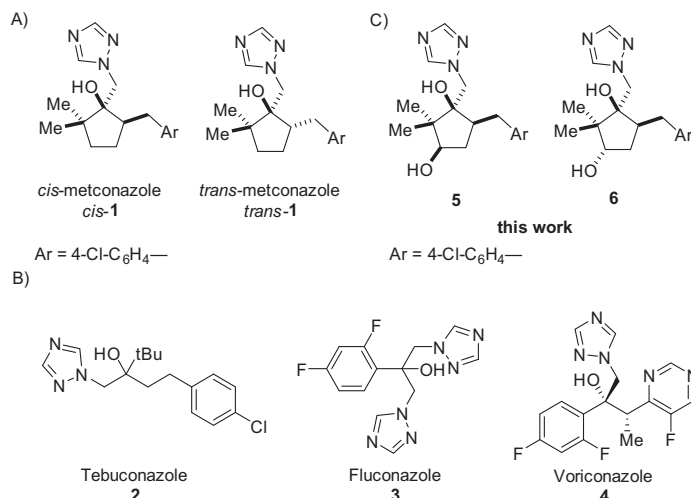
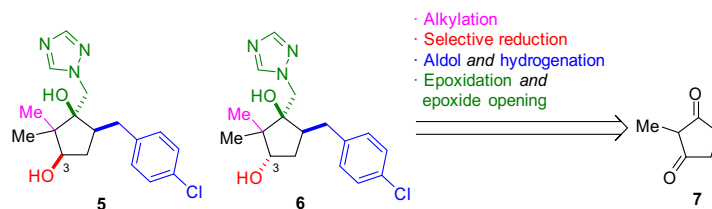


Fig. 1. (A) *cis*- and *trans*-metconazole; (B) Selected important triazole fungicides; (C) Potential metabolites of *cis*-metconazole.

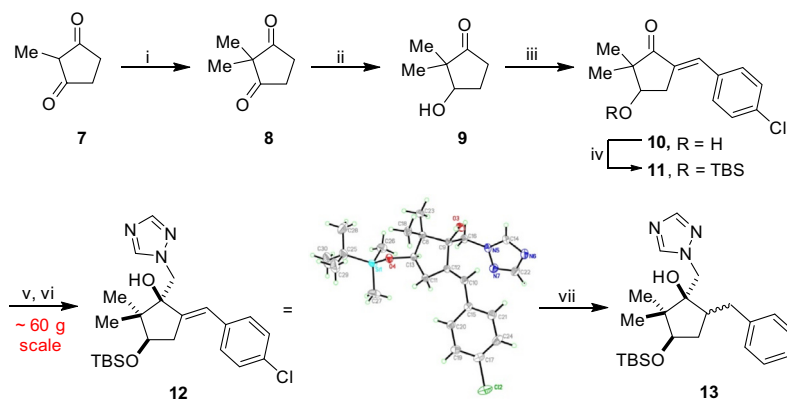


Scheme 1. Retrosynthetic analysis of compounds **5** and **6**.

epoxidation of **11**, performed under Danishefsky conditions,²⁶ gave the corresponding epoxide which was directly subjected to the next step due to its instability. Treatment of the epoxide with 1,2,4-triazole/K₂CO₃ in DMF at 85 °C afforded **12** in 50% yield as a single diastereomer; the relative stereochemistry was confirmed by X-ray crystallography (Scheme 2). The stereochemical outcome of the epoxidation reaction can be explained by shielding the top face of the cyclopentane ring by the TBS-protecting group, which forces attack of the sulfonium ylide on the C=O bond from the opposite side (see ESI, Fig. S1 for details).

Hydrogenation of the trisubstituted alkene bond in **12** was challenging. Under typical conditions (Pd/C, 1 atm H₂, EtOH or EtOAc, rt;

Pd/C, 1 atm H₂, EtOAc, 65 °C; Pd/C, 100 atm H₂, EtOH) compound **12** reacted sluggishly with low conversions, leading to dechlorinated **13** as the main product. Attempts to modify the catalyst activity by using Ph₂S^{27,28} as a catalyst poison failed. With low Ph₂S loadings, no changes in the chemoselectivity were observed; however, higher loadings stopped the reaction completely. Remarkably, in a weakly basic environment with ammonium formate as the hydrogen source, **12** was reduced smoothly, providing **13** in 75% yield as a mixture of diastereomers. Cationic reduction of **12** was also attempted, however, this approach led to partial deprotection of the TBS-protecting group (Et₃SiH/TFA) or to complete decomposition of the starting material (Et₃SiH/TfOH).



Scheme 2. Synthesis and attempted hydrogenation of key intermediate **12**. Reagents and conditions: i. a) NaOH (1 eq), H₂O, rt, 15 min; b) MeI (1.5 eq), DMF, rt, overnight; c) HCl(aq), 1 h, reflux, 36% (3 steps) ii. NaBH₄ (0.27 eq), MeOH/H₂O (4:1), 0 °C → rt, 30 min, 94%; iii. 4-chlorobenzaldehyde (1 eq), NaOH (1 eq), MeOH/H₂O (4:1), rt, 1 h, 89%; iv. TBSCl (1.5 eq), imidazole (3.0 eq), DMF, rt, overnight, 91%; v. trimethylsulfonium iodide (1.2 eq), KHMDS (1.1 eq), THF, 0 °C, then **11**, 0 °C, 1.5 h; vi. 1,2,4-triazole (1.5 eq), K₂CO₃, (1.5 eq), DMF, 60 °C, overnight, 50% (2 steps); vii. 10% Pd/C (16% w/w), HCO₂NH₄ (10 eq), EtOH, reflux, 1 h, 75%, dr = 88:12.

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