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Grignard allylic substitution reaction catalyzed by 1,2,3-triazol-5-ylidene magnesium complexes

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ABSTRACT

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Controlling the reactivities and selectivities of nucleophilic organometallic compounds, RmMXn, is key to the success of selective bond formation. The electronic and/or steric nature of these compounds can be altered by the coordination of Lewis-basic solvents or chelating agents and/or by complexation with other inorganic or organometallic materials such as LiCl and Li(acac).¹ Such reactivity control is accepted as a versatile means of developing and tuning these reactions. However, most of these schemes require stoichiometric or solvent quantities of the controlling reagent.² Catalytic use of Lewis-basic reagents is most efficient for controlling the regiochemical and stereochemical course of a reaction, as exemplified by asymmetric catalyses developed for the Lewis-base-activated aldol³ and allylation⁴ reactions of carbonyl compounds.

Recently, Lewis-base-catalyzed, Cu-free allylic substitution reactions with organomagnesium and organozinc reagents have been reported. Lee and Hoveyda reported that in asymmetric allylic alkylation reactions with Grignard reagents using chiral N-heterocyclic carbenes⁵ (NHCs) as Lewis-base catalysts,⁶ magnesium 4,5-dihydroimidazol-2-ylidene alkoxide was postulated as an active catalyst. This was based on the observation that the lack of a phenolic OH in the side-chain of the precursor imidazolium salt caused low reactivity and selectivity. Kondo et al. reported that the allylic substitution of 4-chloro-2-alkylbutenoates with dialkyl-zinc reagents in the presence of catalytic phosphazene base (*t*-Bu-P4 base) and LiCl selectively yields S_N2' products.⁷ However, these

investigations have been restricted to the reaction of 4-chloro-2alkylbutenoates (tiglic acid esters). Alexakis and co-workers have reported an improved, efficient asymmetric allylic substitution procedure based on chiral imidazol-2-ylidenes with a side-chain bearing a phenolic OH as an N-substituent.⁸ These findings clearly demonstrate the effects of Lewis-base coordination in altering the reactivity of RMgX and R₂Zn reagents. Recently, we have developed γ -selective Grignard allylic substitution reactions of γ -alkyl- or aryl-substituted allylic chlorides and phosphates, **1** catalyzed by imidazol-2-ylidenes (**imd-2** carbenes) as well as imidazole-5-ylidenes (**imd-5** carbenes) (Eq. 1), which can be possibly generated in situ from the corresponding imidazolium salts such as **4a-d** and **5a-e** (Scheme 1).⁹

Allylic chlorides and phosphates reacted with alkyl-Grignard reagent in an S_N2'-selective manner in the

presence of a catalytic amount of 1,2,3-triazol-5-ylidenes to provide α -branched alkenes.

It has been observed that the reactions with imidazol-2-ylidene (**imd-2**) were generally slower and more selective while those with imidazol-5-ylidene (**imd-5**) were faster and less selective.

Herein we report the γ -selective Grignard substitution reactions of allylic chlorides and phosphates **1** catalyzed by 1,2,3-triazol-5-ylidenes (**taz-5** carbenes),¹⁰ which may be generated in situ from the corresponding triazolium salts **6a–d** (Scheme 1). It was found that, in comparison to catalysis with **imd-2** or **imd-5**, reactions with **taz-5** were reasonably rapid with a higher S_N2'-(γ -) selectivity, which enables the construction of a quaternary carbon center from 3,3-disubstituted allylic chlorides. It has been reported that 1,2,3-triazol-5-ylidenes were coordinated to various metals and their donor properties were estimated to be compatible with the most basic 2-imidazolylidenes, albeit being slightly less basic than other abnormal carbenes.¹¹ This report presents the first example of catalytic use of 1,2,3-triazol-5-ylidene magnesium





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Scheme 1. N-heterocyclic carbine-catalyzed allylic Grignard substitution.

complexes in organic transformation and comparison of their reactivity with that of imidazol-2-ylidene and imidazole-5-ylidene complexes.

Trisubstituted 1,2,3-triazolium salts **6a–d** were synthesized by the copper-catalyzed azide-alkyne [3+2] cycloaddition (Hüisgen reaction)¹² followed by inter- or intra-molecular N-alkylation reaction (Scheme 2).¹³

Initially, we investigated the reaction of trans-1-chloro-3-nonene (1a) with *n*-BuMgCl in the presence of various imidazolium salts, 4a,b or 5a,b or triazolium salts 6a-d to compare their catalytic activity and selectivity (Scheme 3). The reactions were quenched after 6 h and the results are shown in Table 1. The reaction in the absence of an imidazolium or triazolium salt was very slow and yielded only a trace amount of the $S_N 2$ product **2a** (data not provided). As shown in the table, all imidazolium and triazolium salts cleanly catalyzed the substitution reaction to yield a mixture of 2a and 3a. In all cases, the S_N2' product 3a was obtained with high selectivity. The 1,2,3-trisubstituted imidazolium salts 5a,b and 1,3,4-trisubstituted 1,2,3-triazolium salts 6a-d, which may be deprotonated by the reaction with a Grignard reagent to respectively generate imd-5 or taz-5 carbenes, effectively catalyzed the reaction to complete in <6 h at room temperature (runs 3-8). Meanwhile, the reactions with 1,3-disubstituted imidazolium salts 4a and 4b – precursors of the corresponding imd-2 carbenes-were relatively slow (runs 1 and 2), when the reaction for 6 h did not complete and over 55% of the substrate 1a was recovered. Since 4a (or 4b) and 6b have the same substituents (mesityl) at the carbene–carbon β , the observed difference of reactivity may



(a) CuSO₄-5H₂O (5 mol%), sodium ascorbate (15 mol%), CH₃CN-H₂O, 80 °C; (b) MeI, CH₃CN, 60 °C; (d) *i*-Pr₂NEt, Tf₂O, CH₂Cl₂

Scheme 2. Preparation of 1,2,3-triazolium salts 6a-d.



Scheme 3. Reaction of 1a with *n*-BuMgCl.

Table 1

Reaction of **1a** with *n*-BuMgCl in the presence of 5 mol % of an imidazolium salt, **4a**-**d** or **5a**-**e** or triazolium salt **6a**-**d**^a

Run	Catalyst (5 mol %)	Conversion ^b	2a:3a ^b (α : γ)
1 ^c	4a	43	1:99
2 ^c	4b	36	2:98
3 ^c	5a	99	4:96
4 ^c	5b	96	6:94
5	6a	96	1:99
6	6b	99	2:98
7	6c	93	1:99
8	6d	98	5:95

^a Compound **1a** (1.0 mmol), *n*-BuMgCl (1.5 mmol, 0.90 M in THF), **4**, **5**, or **6** (0.05 mmol), THF (5 mL).

^b Determined by ¹H NMR analysis of the crude mixture using an internal standard.

^c Previous results (see Ref. 9).

mainly attribute to an electronic reason (runs 1, 2, and 6). The TLC analyses of the reaction mixtures indicated that the reaction with **5a** was faster than that with **6b**. Overall, 1,2,3-triazol-5-ylidenes (**taz-5** carbenes) was able to combine both high reactivity and S_N2' -selectivity.

With these results in hand, we further studied the reaction of various allylic substrates **1** illustrated in Figure 1 and several Grignard reagents in the presence of 5 mol % of the representative salts of **4a** (or **4b**), **5a**, or **6b** (Table 2).¹⁴ For comparison, **4a**, **4b**, and **6b** having the same substituents at the carbene–carbon β were chosen.

For the reaction of a secondary Grignard reagent (i-PrMgCl) (runs 1-3), only taz-5 carbene from 6b gave satisfactory yield and selectivity (run 3). The 4b- and 5b-catalyzed reactions showed much lower selectivity. Meanwhile, the reaction with the aryl-Grignard reagent (PhMgCl) resulted in a low selectivity to predominantly give the α -substituted product (runs 4 and 5). The 6b-catalyzed reactions with silyloxymethyl- and secondary alkylsubstituted substrates, 1b and 1c, also proceeded very selectively (runs 6 and 7). The imd-2 or taz-5 carbene from 4a or 6b, respectively, selectively catalyzed the reaction of cinnamyl chloride (1d) and phosphate 1e to afford the corresponding $S_N 2'$ product 3d in excellent yield (runs 8, 10, 11, and 13) but the reactions with imd-5 carbene from 5a showed relatively low selectivity (runs 9 and 12). All 4b-, 5a-, 6b- and IMes-CuCl-catalyzed reactions of the conjugated diene substrate **1f** produced a mixture of $S_N 2'$ $(\alpha$ -) and S_N2' $(\gamma$ -) products **2f** and **3f**, but the ε -substituted



Figure 1. Allylic substrates.

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