



Technologically promising, room temperature luminescent columnar liquid crystals derived from *s*-triazine core: molecular design, synthesis and characterization

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ABSTRACT

Three fold Horner-Wadsworth-Emmons (HWS) reaction of triphosphonate with achiral/chiral 3,4,5-tris(alkoxy)benzaldehydes yield novel achiral/chiral star-shaped liquid crystals (LCs), in which three fluorophore arms such as 1,2,3-tris(alkoxy)-5-styrylbenzenes are tethered to a central *s*-triazine core. These LCs, the first of their kind, display a columnar liquid crystalline phase over a wide temperature range existing between well below and above the room temperature, which is evidenced by optical and calorimetric studies. Besides, they show photoluminescence both in solution and mesomorphic states. Thus, given their self-organization into fluid one-dimensional columnar array coupled with light generating capability, these organic ordered-fluids can be regarded as novel media for advanced technological applications.

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Since 1977 when Chandasekhar et al.¹ discovered columnar (Col) liquid crystal (LC) phases in benzene-hexa-*n*-alkanoates, there has been a phenomenal growth in research activity relating to the rational molecular design and synthesis of structurally diverse organic molecules capable of exhibiting such mesophase(s).² This is because Col phases, generally formed by the spontaneous self-assembly of shape-anisotropic π -conjugated molecules into fluid columnar stacks, are highly anisotropic and ordered structures with extensive π -orbital overlap leading to a high degree of uniaxial charge-carrier mobilities. Besides, they can be readily and repeatedly processed from the isotropic melt of mesogens to eliminate structural/electronic defects when compared to single-crystals or inorganic semiconductors used in technological applications.^{2,3} They have been recognized as potential media for hole- and electron-transport given the fact that the measured charge-carrier mobility value in some Col LCs exceeds $1 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$.⁴ Thus, Col phases hold immense promise in electronic devices such as photovoltaic cells,³ field effect transistors⁵ and organic light emitting devices (OLEDs).⁶

Their usage in fabricating OLEDs is especially significant if an additional parameter such as the transport of excitation energy, the luminescence behavior, is incorporated in the constituent mesogens; in fact, the intrinsic photoluminescence property enables

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the Col phases to serve as emissive layers as well in OLEDs. Thus, in recent years there has been a growing interest in developing luminescent organic molecules exhibiting columnar behavior where a large variety of shape-anisometric motifs bearing fluorophores have been reported.⁷ Among these, 1,3,5-triazine-based LCs⁸ appear to be suitable substances in view of the fact that 1,3,5-triazine core possesses strong electron affinity when compared to other aromatic rings.⁹ Although a large number of triazine-based compounds have been prepared,⁸ the efforts in design and synthesis of such functional materials capable of exhibiting luminescent Col phase existing between well below, and above the ambient temperature, a vital feature required for device performance, are absolutely necessary.

In order to achieve these goals, C_3 -symmetric three-armed star-shaped compounds **1a** and **1a-b** have been rationally designed, synthesized, and probed for their physical properties. Scheme 1 depicts the general molecular structure of these systems. As can be seen, they contain three 1,2,3-tris(alkoxy)-5-styrylbenzenes, as fluorophore arms, linked covalently to a central electron-deficient 1,3,5-triazine core. Here, styryl arms not only accounted for the effective conjugation but also serve as electron-rich cores as they possess electron-donating branched alkoxy tails. These branched alkoxy chains have been especially chosen given the experimental observation that they reduce melting (crystal-to-Col phase) transition temperatures and thus, widen the thermal range of the required LC phase. They are also known to lower the clearing (Col-to-isotropic liquid phase) transition temperatures.^{2,10} In fact,

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