



N-Arylation of a hindered indoline as a route to 2-(2-methyl-1-(4-oxo-3,4-dihydrophthalazin-1-yl)-1H-indol-3-yl)acetic acid derivatives

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ABSTRACT

A convenient synthesis of 2-(2-methyl-1-(4-oxo-3,4-dihydrophthalazin-1-yl)-1H-indol-3-yl)acetic acid derivatives is described using a microwave-promoted multi-step S_NAr reaction. The desired products were found to exhibit atropisomerism.

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Prostaglandin D₂ (PGD₂) is involved in a wide variety of inflammatory conditions. Two receptors have been identified for PGD₂, DP₁, and more recently DP₂ (CRTh2).^{1–3} The recent discovery of CRTh2 has prompted an enormous effort both in industry as well as in academia for the identification of CRTh2 antagonists.⁴ Indole acetic acids have found great prominence in this area of research.⁵ Our efforts in this particular area have led us to investigate structural motifs like **1** (Fig. 1) as antagonists for CRTh2.

Our initial retrosynthetic analysis of **1** led to the obvious disconnection between the two biaryl rings. We envisaged constructing

this bond using a metal mediated coupling of an appropriately functionalized substrate or a base promoted S_NAr reaction.⁶ Preliminary experiments showed that 3,6-dichloropyridazine **5** could be coupled to the desired indole **4**, suggesting that 1,4-dichlorophthalazine **2** could also be a suitable coupling partner as well (Scheme 2). Additionally, phthalazinones **3** could be used in the event that 1,4-dichlorophthalazine failed to couple to **4**. Strategy **A** (Scheme 1) was preferred as structural diversity may be installed at a later stage in the synthesis. Unfortunately, extensive efforts to achieve this bond construction using both **2** as well as a variety of

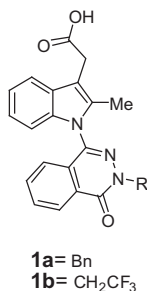
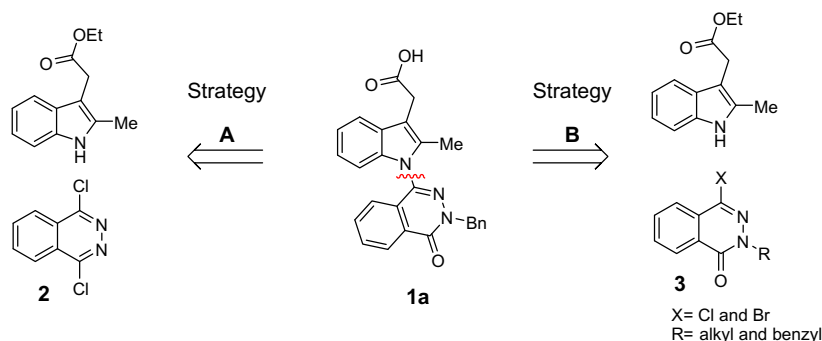


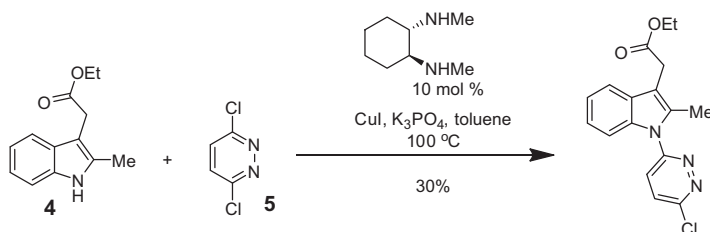
Figure 1. Compounds with the indole acetic acid scaffold.

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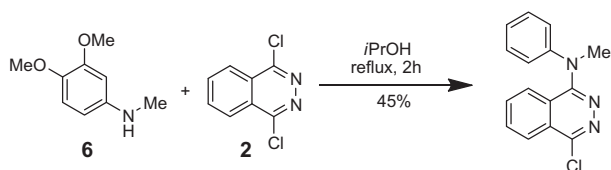
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Scheme 1. Initial synthetic strategy for the construction of indole acetic acids.



Scheme 2. Precedent for Buchwald coupling of **4**.



Scheme 3. Precedent for the synthetic strategy.

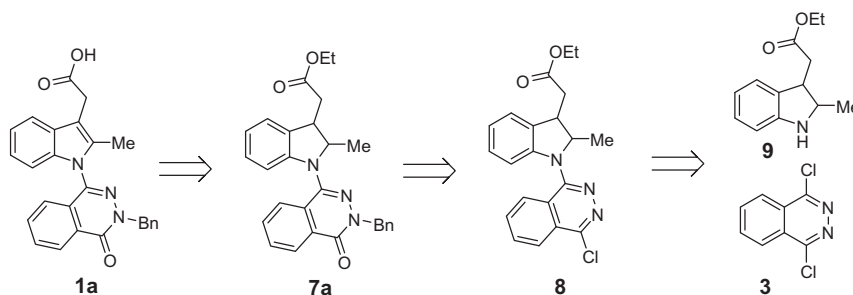
functionalized phthalazinones **3** were unsuccessful. This required us to reevaluate our approach.

We surmised that both the substitution at the 2-position of the indole as well as the electron withdrawing nature of the ester served to decrease the nucleophilicity of the indole nitrogen. De-aromatization of the indole should, in principle, increase the nucleophilicity of the nitrogen while retaining the desired molecular framework. Indeed, such a strategy has been previously utilized.⁷ However, details regarding the yields and efficiency of these reactions were not disclosed. Additionally, indoline intermediates could be valuable compounds in our medicinal chemistry endeavor. Since we had a large amount of intermediate **4** available and literature precedent for using indolines, this strategy was determined to be a worthy pursuit.⁸

We envisaged the desired product arising from an oxidation followed by ester hydrolysis. N-Arylated indoline **7** would come from hydrolysis of **8** followed by N-alkylation. The functionalized indoline **8** would be constructed from an S_NAr reaction and the indoline starting material **9** would be derived from reduction of the parent indole **4** (Scheme 4).

The synthesis commenced with the known reduction of commercially available ethyl 2-(2-methyl-1H-indol-3-yl)acetate **4** using a mixture of Et₃SiH and TFA. The reduction proceeded as reported to afford indoline **9** as an 8:1 cis:trans mixture.⁹ With indoline **9** in hand, we were curious to try the S_NAr reaction. We selected a modified version of the conditions for the coupling of aniline **7** with 1,4-dichlorophthalazine (Scheme 3). As a starting point, we decided to use microwave heating as these conditions would allow for rapid evaluation of the reaction at a larger range of temperatures. To our delight, the initial experiment revealed that microwave heating of indoline **9** along with 1,4-dichlorophthalazine **2** in iPrOH at 110 °C produced the desired functionalized product **8** in 90% yield. Treatment of **8** with NaOH in AcOH at 70 °C for 2 h afforded **10** in 61% yield (Scheme 5).

Upon evaluating the by-products of the microwave promoted S_NAr reaction, we found a small amount of **10** present in the crude reaction mixture. We pondered whether this finding could be exploited to drive the reaction directly to **10** obviating the need



Scheme 4. Retrosynthetic analysis of **1a**.

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