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# FeCl<sub>3</sub>-catalyzed tandem condensation/intramolecular nucleophilic addition/C–C bond cleavage: a concise synthesis of 2-substituted quinazolinones from 2-aminobenzamides and 1,3-diketones in aqueous media

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## ABSTRACT

A concise approach for the synthesis of 2-substituted quinazolinones using an iron-catalyzed tandem reaction of 2-aminobenzamides with acyclic or cyclic 1,3-diketones via condensation, intramolecular nucleophilic addition, C–C bond cleavage in an aqueous solution of poly(ethylene glycol) under oxidant-free conditions has been developed.

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Nitrogen heterocycles have aroused considerable interest because of their presence in many therapeutically and biologically active compounds. In particular, quinazolinones are building blocks for approximately 150 naturally occurring alkaloids and many marketed drugs.<sup>1</sup> Such compounds have a wide range of useful biological and therapeutic activities,<sup>2</sup> such as antimalarial, anti-hypertensive, antimicrobial, anti-inflammatory, and anticancer activities.

Because of their remarkable importance, many efforts have been made to prepare such compounds from a variety of starting materials, of which 2-aminobenzamides are probably the most typical.<sup>3</sup> In general, the quinazolinones were constructed by the reaction of 2-aminobenzamides with carboxylic acids,<sup>4</sup> carbonyls,<sup>5</sup>  $\beta$ -ketoesters,<sup>6</sup> benzyl alcohol,<sup>5,7</sup> methylenes,<sup>8</sup> or carbonylative conditions.<sup>9</sup> Notably, most of the reported synthetic routes require excess amounts of oxidants or bases. Suitable ligands or microwave irradiation conditions are necessary in some cases. Therefore, eco-friendly and practical methods to access valuable 2-substituted quinazolinones are highly desirable. To continue our research interest in iron catalysis<sup>10</sup> and catalytic reactions in green media,<sup>11</sup> we have reported a simple, eco-friendly, and practical iron-catalyzed strategy for synthesis of 2-substituted quinazoli-

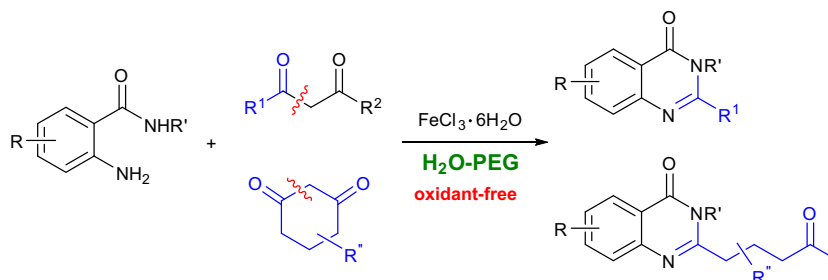
none derivatives through tandem condensation, intramolecular nucleophilic addition, C–C bond cleavage of 2-aminobenzamides, and acyclic or cyclic 1,3-diketones in an aqueous solution of poly(ethylene glycol) under oxidant-free conditions (Scheme 1).

Initially, 2-aminobenzamide (**1a**) and 2,4-pentanedione (**2a**) were used as the model substrates to optimize reaction conditions including iron catalyst, solvents, and catalyst loading. As shown in Table 1, eight iron catalysts (10 mol %) were tested in poly(ethylene glycol) (PEG-400) as the solvent at 100 °C for 24 h (entries 1–8). Iron trichloride hexahydrate (FeCl<sub>3</sub>·6H<sub>2</sub>O) provided the desired product **3aA** in highest yield (entry 8), which was superior to the anhydrous counterpart (entry 2). We supposed that water may enhance the reactivity of this transformation, so the reaction was attempted in pure water. However, the yield was not increased due to the poor solubility of the reactants (entry 9). Then, a mixture of PEG-400/H<sub>2</sub>O was used as solvent. As expected, the desired product was obtained in higher yield (entries 10–12), and PEG-400/H<sub>2</sub>O (1:9) was the optimal choice (entry 12; 80% yield). The yield of **3aA** could be increased from 80% to 91% when the catalyst loading was doubled (entry 13).

Then, a wide range of 1,3-diketones were employed in the reactions with 2-aminobenzamide (**1a**) under the optimized conditions (Table 2). On comparing with **2a**, heptane-3,5-dione (**2b**) gave a lower yield of **3aB** under identical conditions (**3aA**: 91%; **3aB**: 83%). In contrast, the product **3aC** was obtained in moderate yield

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Scheme 1. A green synthesis of 2-substituted quinazolinones.

**Table 1**  
Optimization of the reaction conditions for **3aA**<sup>a</sup>

Entry	[Fe] (mol %)	Solvent	Yield <sup>b</sup> (%)
1	FeCl <sub>2</sub> (10)	PEG-400	26
2	FeCl <sub>3</sub> (10)	PEG-400	42
3	FeBr <sub>3</sub> (10)	PEG-400	48
4	Fe(OAc) <sub>2</sub> (10)	PEG-400	<5
5	Fe(OTf) <sub>2</sub> (10)	PEG-400	50
6	Fe(BF <sub>4</sub> ) <sub>2</sub> ·6H <sub>2</sub> O (10)	PEG-400	32
7	Fe(ClO <sub>4</sub> ) <sub>3</sub> ·6H <sub>2</sub> O (10)	PEG-400	46
8	FeCl <sub>3</sub> ·6H <sub>2</sub> O (10)	PEG-400	60
9	FeCl <sub>3</sub> ·6H <sub>2</sub> O (10)	H <sub>2</sub> O	41
10	FeCl <sub>3</sub> ·6H <sub>2</sub> O (10)	PEG-400/H <sub>2</sub> O (1:1)	70
11	FeCl <sub>3</sub> ·6H <sub>2</sub> O (10)	PEG-400/H <sub>2</sub> O (1:4)	74
12	FeCl <sub>3</sub> ·6H <sub>2</sub> O (10)	PEG-400/H <sub>2</sub> O (1:9)	80
13	FeCl <sub>3</sub> ·6H <sub>2</sub> O (20)	PEG-400/H <sub>2</sub> O (1:9)	91

<sup>a</sup> Reaction conditions: 2-aminobenzamide (**1a**; 0.2 mmol), 2,4-pentanedione (**2A**; 0.3 mmol), solvent (1.0 mL), 100 °C, 24 h.

<sup>b</sup> Isolated yield.

**Table 2**  
The scope of 1,3-diketones<sup>a</sup>

Entry	2	3	Yield <sup>b</sup> (%)
1			91
2			83
3			36
4			40

when sterically hindered 2,6-dimethylheptane-3,5-dione (**2C**) was used as the reaction partner (entry 3; **3aC**: 36%). The reaction of amide **1a** and 1,3-diphenylpropane-1,3-dione (**2D**) could take place smoothly to provide the product 2-phenylquinazolin-4(3H)-one (**3aD**) (entry 4). Next, we attempted the reactions of **1a** with the unsymmetrical 1,3-diketones, 1-phenylbutane-1,3-dione (**2E**), and 1,1,1-trifluoropentane-2,4-dione (**2F**). Interestingly, the same product **3aA** was observed through selective C–C bond cleavage (entries 5 and 6). It may be ascribed to the higher reactivity of the acetyl group than benzoyl and trifluoroacetyl groups. Finally, the cyclic 1,3-diketones cyclohexane-1,3-dione (**2G**) and 2-methylcyclohexane-1,3-dione (**2H**) were also subjected to this transformation. The corresponding products (**3aG**) and (**3aH**) were generated in 52% and 68% yields, respectively (entries 7 and 8).

As shown in Scheme 2, to further evaluate the scope of this novel strategy for the synthesis of 2-substituted quinazolinones, various *N*-substituted 2-aminobenzamides **1a–k** were examined. For example, the reaction of 2-amino-*N*-methylbenzamide (**1b**) and pentane-2,4-dione (**2A**) gave the desired product 2,3-dimethylquinazolin-4(3H)-one (**3bA**) in 84% yield. Notably, *N*-aryl-2-aminobenzamides bearing electron-donating groups (**1d**; 4-Me, **1e**; 2-Me, and **1f**; 4-MeO) or electron-withdrawing groups (**1g**; 3-Cl, **1h**; 4-Cl, and **1i**; 3,4-Cl<sub>2</sub>) on the benzene ring also

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