



Reprint of: Synthesis of a tricyclic core of rameswaralide [☆]

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ABSTRACT

A tricyclic core containing a 5,7-fused bicyclic unit of rameswaralide was prepared starting from a 1,6-enyne. The synthetic sequence involved (i) ruthenium-catalyzed [5+2]-cycloaddition of 1,6-enyne, (ii) an acyl radical based approach to construct the lactone, and (iii) a regioselective installation of the conjugated double bond by a concomitant sulfenylation–dehydrosulfenylation sequence.

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Rameswaralide (**1**), a novel diterpene with a tetracyclic ring skeleton, was isolated from the soft coral *Sinularia dissecta* near the Mandapam coast of India in 1998 (Figure 1).¹ Its relative stereochemistry was determined by analysis of ¹H NMR coupling constants and NOESY correlations. The structure of **1** was further confirmed by selective reduction of the enolic group with NaBH₄ to form the corresponding dihydromameswaralide (**2**). It was reported that rameswaralide (**1**), and its derivatives, could function as effective anti-inflammatory agents for the treatment of a variety of inflammatory disorders, including arthritis, psoriasis, and inflammatory bowel disease.²

Rameswaralide (**1**) is a highly oxygenated natural product containing a variety of functional groups, including a tertiary hydroxyl group, a carbonyl moiety, a fully enolized β-ketoester, a γ-lactone, and an isopropylidene group (Fig. 1). Furthermore, it contains *cis*-fusions at the AB- and BC-ring junctions. Due to its complex features and biological importance, rameswaralide (**1**) is a good target for total synthesis. To date, the total synthesis of **1** has not been reported, but several reported approaches have appeared.³ Herein, we report the synthesis of the tricyclic core **3** containing a 5,7-fused bicyclic (AB-ring) unit of rameswaralide (**1**) (Fig. 2).

Retrosynthetically, the tricyclic compound **3** (Fig. 2) could be accessed from chemoselective epoxidation of vinylsilane **4**. In turn,

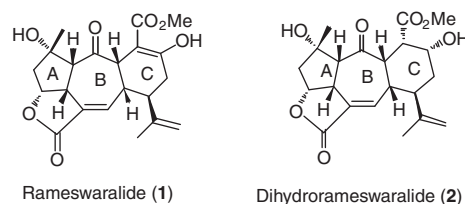


Figure 1. Structure of rameswaralide (**1**) and its derivative (**2**).

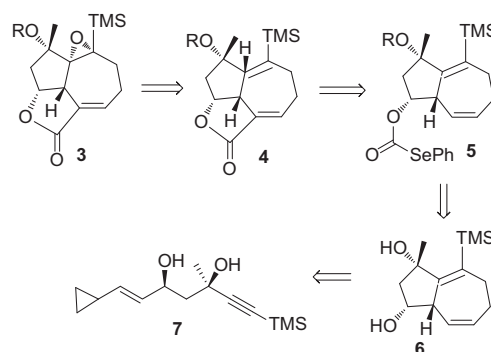


Figure 2. Retrosynthetic analysis of tricyclic core **3**.

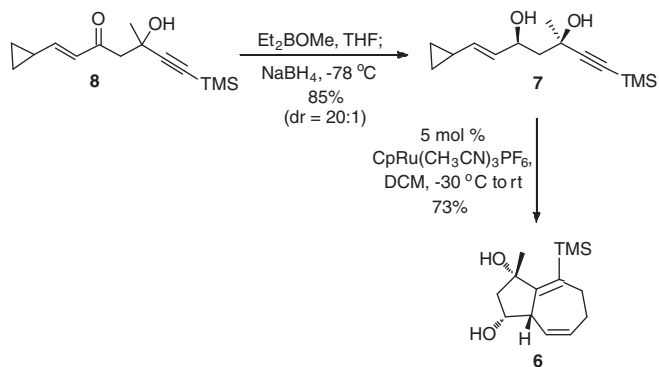
the five-membered γ-lactone in **4** could be derived from the alkoxycarbonyl radical cyclization of acylselenium **5** onto the exocyclic *cis*-disubstituted olefin with retention of the alkene

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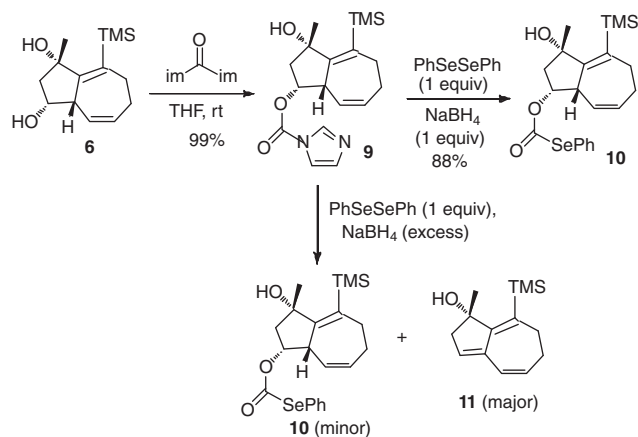


Scheme 1. Formation of the 5,7-fused bicyclic compound **6**.

functionality.⁴ The 5,7-fused bicyclic skeleton **6** could be formed by ruthenium-catalyzed [5+2] intramolecular cycloaddition of the 1,6-enyne **7** developed in these laboratories.^{5,6}

Synthesis of the 5,7-fused bicyclic compound **6** commenced with a known β -hydroxy ketone **8** (Scheme 1).^{5c,d} Stereoselective reduction of **8** to the corresponding *syn* 1,3-diol **7** was realized using Et_2BOMe as a chelating reagent and NaBH_4 as a hydride source.⁷ The diol **7** was obtained in 85% yield and with excellent diastereoselectivity (*dr* = 20:1). The relative stereochemistry of compound **7** was assigned by using Rychnovsky's method for determining the stereochemistry of the 1,3-diol acetonides.⁸ Subjection of 1,6-enyne **7**–5 mol % of $\text{CpRu}(\text{CH}_3\text{CN})_3\text{PF}_6$ in CH_2Cl_2 for 3 h provided the desired hydroazulene product **6** (Scheme 1) in 73% yield as a single diastereomer. The stereochemistry of compound **6** has been determined by X-ray crystallography.^{5c,d}

With 5,7-bicyclic compound **6** in hand, our attention turned to development of the conditions for forming the five-membered α,β -unsaturated lactone found within the tricyclic skeleton **4** (Fig. 2). One possible approach to forming the γ -lactone while retaining the alkene functionality involves an atom transfer radical cyclization of acylselenium precursor **10** (Scheme 2) onto the disubstituted *cis*-olefin.⁴ Accordingly, chemoselective acylation of diol **6** with 1,1'-carbonyl-diimidazole provided acylimidazole **9** in 99% yield. Treatment of **9** with a reagent combination of PhSeSePh and NaBH_4 afforded acylselenium **10** in 88% yield (Scheme 2). It is important that an equimolar amount of PhSeSePh and NaBH_4 must be used in the reaction to ensure high conversion in forming the desired product **10**. If an excess amount of NaBH_4 was employed in the reaction, an undesired elimination product **11** (Scheme 2) was formed as the major product.



Scheme 2. Synthesis of acylselenium **10**.

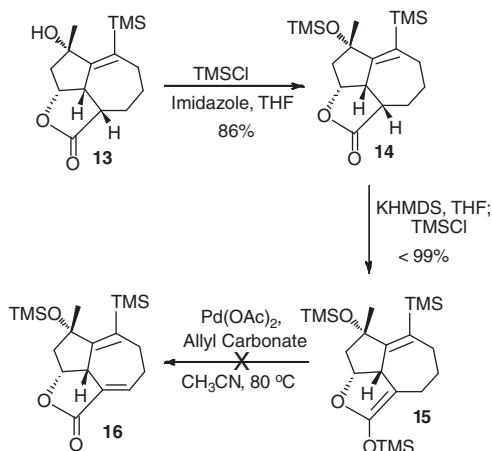
Table 1
Atom transfer/reductive radical cyclization

Entry	Conditions	Results
1	Et_3B , O_2 , PhH , rt, 24 h	No reaction
2	Et_3B , O_2 , $\text{Yb}(\text{OTf})_3$, PhH	Decomposition
3	$\text{Bu}_3\text{SnSnBu}_3$, <i>h</i> ν , PhH , 80 °C	No reaction
4	$\text{Me}_3\text{SnSnMe}_3$, <i>h</i> ν , PhH , 80 °C	No reaction
5	40 mol % Bu_3SnH , 20 mol % AIBN PhH , 80 °C, 24 h	12 (30% yield)
6	120 mol % Bu_3SnH , 20 mol % AIBN PhH , 80 °C, 12 h	12 (86% yield)

Having successfully synthesized radical precursor **10**, a variety of atom transfer radical cyclization conditions were attempted (Table 1, entries 1–4).⁹ Unfortunately, these conditions did not provide the desired tricyclic product **12**, and only resulted in decomposition or recovery of the starting material **10**. On the other hand, when a substoichiometric amount of Bu_3SnH (40 mol %) was employed in the reaction, a reductive radical product **13** was isolated in 30% yield (entry 5). Use of stoichiometric amount of Bu_3SnH (1.2 equiv) and AIBN (0.2 equiv) as the radical initiator provided tricyclic product **13** in 88% yield as a single diastereomer (entry 6).

To prepare the required five-membered α,β -unsaturated lactone, the reported Tsuji method of reacting a silyl enol ether with allyl methyl carbonate was explored.¹⁰ To avoid any difficulties associated with the formation of α,β -unsaturated lactone, the tertiary hydroxyl group within **13** was protected as silyl ether **14** (Scheme 3). Subsequent treatment of **14** with KHMDS at -78 °C, followed by trapping the enolate intermediate with TMSCl , provided silyl enol ether **15**. Reaction of **15** with $\text{Pd}(\text{OAc})_2$ and allyl methyl carbonate in THF at 80 °C for 18 h did not yield the desired α,β -unsaturated lactone **16**.

An unsuccessful attempt at transforming the silyl enol ether **15** into the corresponding γ -lactone **16** prompted us to explore the introduction of the conjugated double bond by a concomitant sulfenylation–dehydrosulfenylation approach (Scheme 4).¹¹ When PhSSO_2Ph was employed as the electrophilic reagent, sulfide **17** was obtained in 66% yield along with the undesired epimer **18** (Scheme 4). Chemoselective oxidation of the phenyl sulfide functionality in **17** was investigated next. The use of *m*CBPA (1.1 equiv)



Scheme 3. Attempted formation of α,β -unsaturated lactone.

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