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Synthesis of an all-cis intermediate of ticagrelor



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ABSTRACT

A six step conversion of the common carbocyclic nucleoside precursor **8** into the all-*cis* key intermediate for the synthesis of ticagrelor analogs is reported. The method involves two oxidation/stereoselective reduction sequences for both the C—O and C—N bonds. Inversion of stereochemistry was confirmed by analysis of spin couplings between the hydrogens at the junction of the 1,3-dioxolane and cyclopentane rings.

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Introduction

Carbocyclic nucleosides (CNs) are compounds possessing important antiviral, antitumor, and antibiotic activities. Naturally occurring CNs—aristeromycin $(1)^2$ and neplanocin A $(2)^3$ served as an inspiration for the design and synthesis of many unnatural analogs. Some of these, entecavir (3), carbovir (4), abacavir (5), and ticagrelor (6), have found application as drugs (Fig. 1).

The stereochemistry of many CNs analogs is similar to that of the natural nucleosides. However, this similarity is not required to effect biological activities and there are examples of bioactive nucleoside analogs in which one stereogenic center has been inverted. 4b-d.f Interestingly, all-cis CNs, in which two stereocenters are inverted, have not been broadly investigated. However, these compounds have occasionally been reported as part of general synthetic approaches to CNs (Scheme 1).9

These examples show that all-cis CNs can be formed, at least as minor products, during the synthesis of APIs (active pharmaceutical ingredients) in the pharmaceutical industry. Therefore, access to this scaffold is particularly important in drug development processes in order to determine their levels in APIs.

During the course of our research aimed at the synthesis of ticagrelor impurities, we developed a synthesis of all-*cis* **7** from previously reported carbocycle **8**. All-*cis* **7**, where the C-1 and C-4 stereocenters are inverted, is an analog of **9**, a key intermediate

previously used in the synthesis of ticagrelor. We considered that with all-*cis* **7** in hand, all-*cis* ticagrelor may also be prepared (Scheme 2).

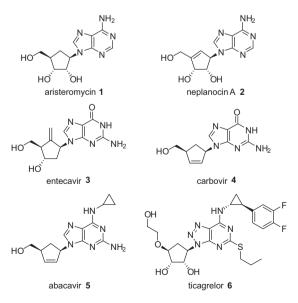


Figure 1. Selected carbocyclic nucleoside natural products and analogs.

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Scheme 1. Studies toward all-cis CNs.

Scheme 2. An outline of the study.

Results and discussion

Our strategy for inversion of the configuration at C-1 and C-4 involved oxidation of the C—O and C—N bonds to the ketone and imine, respectively, followed by reduction with metal hydrides. We envisioned that steric hindrance induced by the 1,3-dioxolane ring should preferentially favor exposure of one face of the cyclopentane ring to hydride attack (Fig. 2).

Our synthesis began with compound **8**, which was obtained from p-ribose (**10**) in 7 steps according to a previously published route. The C-4 stereochemistry was inverted in two high yielding steps. First, the secondary alcohol was oxidized to ketone **11** under

Figure 2. Proposed stereoselectivity for reduction of the C=O and C=N bonds.

Scheme 3. Synthesis of all-cis **7**.

Swern conditions before reduction with NaBH₄ to exclusively afford alcohol **12**.

Before transformation of the C—N bond to the C=N bond, the secondary alcohol was needed to be protected. The CH₂CO₂Et group was selected because it could be converted into the HOCH₂-CH₂ unit present in ticagrelor, during reduction of the C=N bond. Williamson etherification using ethyl bromoacetate proved to be a difficult transformation, giving **13** in only 34% yield. Presumably, both the steric hindrance of the secondary alcohol and the relatively acidic proton of the NHCbz group made this reaction difficult.

The Cbz protecting group was removed under standard conditions providing amine **14** in 93% yield which was smoothly converted, in a tungsten-catalyzed process¹⁰ with hydrogen peroxide as the external oxidant, to oxime **15** in 90% yield. Reduction of both oxime and esters moieties in **15** with LiAlH₄ provided amine **7** in 43% yield (Scheme 3).

The stereochemistry of compounds **12** and **7** was assigned according to their ¹H and 2D NMR spectra and by comparison with the spectra of uninverted **8** and **9** (Fig. 3).

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