



Synthetic studies on isoschizogamine: construction of [3.3.1] bicyclic aminal core by using oxidative skeletal rearrangement



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This article is dedicated to memory of the late Professor Emeritus Keiichiro Fukumoto

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ABSTRACT

The tetracyclic core structure of isoschizogamine containing aminal functionality was constructed by oxidative skeletal rearrangement of a 1,2-diaminoethene derivative. The 1,2-diaminoethane was prepared by palladium-catalyzed allylation at the 4a position of a 1,2,3,4-tetrahydro- β -carboline derivative and subsequent lactam formation. After the oxidative skeletal rearrangement using dimethyldioxirane, the allyl group was removed by a three-step sequence to provide the tetracyclic core skeleton of isoschizogamine with aminal functionality.

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Biologically active polycyclic alkaloids containing aminal functionality have often been found in nature. Among them, isoschizogamine (**1**), isolated from the shrub of *Schizogygia coffaeoides*,¹ is a member of the schizogygane indole alkaloid (Fig. 1). Since schizogygane family compounds typically possess a hexacyclic ring system including an indoline skeleton, the structure of isoschizogamine was initially assigned as compound **1'**, an epimer of schizogamine (**5**) at the C-7 position (Fig. 2).² Later, Hajicek et al. revised the structure **1'** to **1** with a densely fused tetrahydroisoquinoline core containing four consecutive stereogenic centers and aminal functionality.³

The densely fused aminal structures in **1** have posed a formidable synthetic challenge and inspired significant interest from the synthetic community. The first racemic total synthesis of **1** was achieved by Heathcock and co-workers in 1999,⁴ which was followed by the recently completed Fukuyama's first asymmetric total synthesis.⁵ In addition, three synthetic studies have been reported to date.⁶

During the course of our research program on synthetic studies on dimeric indole alkaloid, (+)-haplophytine (**11**), we developed a strategy for the construction of [3.3.1] bicyclic aminal framework from indole derivative **7** by a sequential oxidative skeletal rearrangement (Scheme 1).⁷ The plausible reaction mechanism of the

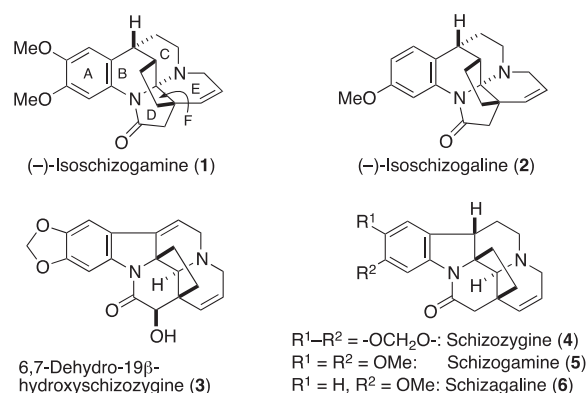


Figure 1. Schizogamine alkaloids.

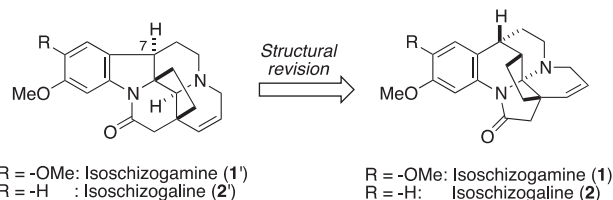
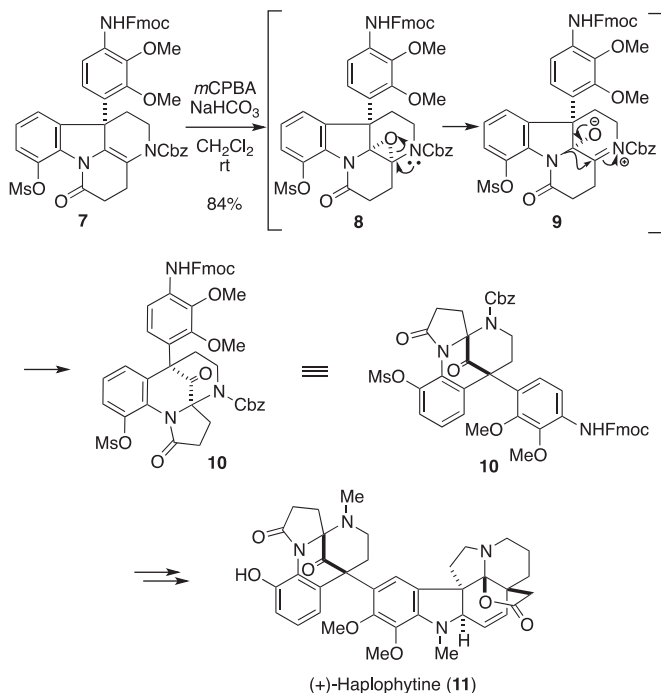


Figure 2. Structural revision of isoschizogamine (**1**).

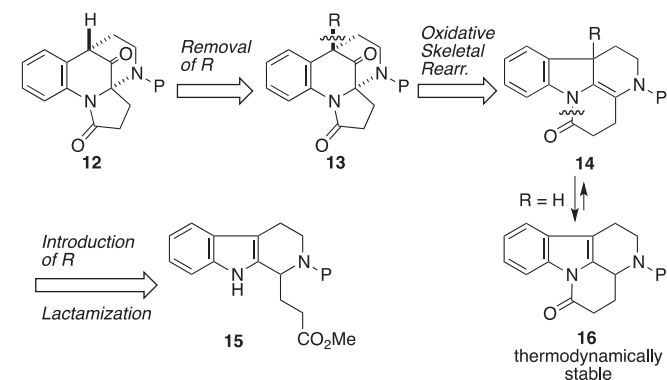
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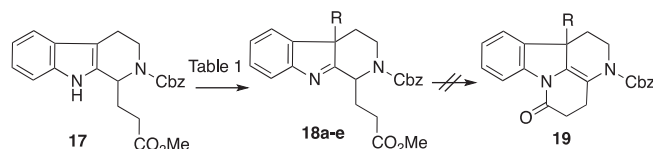
Scheme 1. Development of oxidative skeletal rearrangement in total synthesis of (+)-haplophytine (**11**).

sequence is epoxidation of the 1,2-diaminoethene moiety and spontaneous ring opening of epoxide by nitrogen of the Cbz carbamate to afford a zwitterionic species **9**. Then, 1,2-shift of the C–N



Scheme 2. Synthetic strategy for tetracyclic core of isoschizogamine (**1**).

Table 1
Introduction of substituents at C-7 position



Entry	Reagents (equiv)	Solvent	Temp	Time (min)	R	Yield
1	Selectfluor (1.5)	CH ₃ CN	rt	10	F	Complex mixture
2	<i>t</i> -BuOCl (1.1)	CH ₂ Cl ₂	0 °C	10	Cl	Complex mixture
3	NIS (1.5)	CH ₂ Cl ₂	rt	5	I	18c , 89%
4	Pb(OAc) ₄ (1.5)	CH ₂ Cl ₂	rt	5	OAc	18d , <64%
5	CSA (2.0), <i>N</i> -PSP (3.0)	CH ₂ Cl ₂	rt	5	SePh	Complex mixture

bond by a way of semi-pinacol type rearrangement⁸ gives the [3.3.1] bicyclic aminal system. We considered that this strategy should serve as a powerful means for construction of a broad range of polycyclic alkaloids containing densely fused aminal from easily accessible substrates. Herein, we describe stereoselective construction of the [3.3.1] bicyclic aminal core of **1** by utilizing the oxidative skeletal rearrangement developed in our group.

Our synthetic strategy for **12**, a possible intermediate for isoschizogamine (**1**), is shown in Scheme 2. For construction of the [3.3.1] bicyclic aminal skeleton, we planned to conduct the oxidative skeletal rearrangement of **14**. Compound **14** would be readily prepared from tetrahydro-β-carboline **15** by introduction of the substituent at the C-7 position and lactamization of the resultant indolenine derivative with migration of the double bond. Introduction of a substituent at the C-7 position is necessary to prevent tautomerization from the 1,2-diaminoethene form **14** to the thermodynamically more stable indole form **16**. After construction of the aminal structure, the substituent should be removed to obtain **12**. Tetrahydro-β-carboline **15** would be prepared from tryptamine via Bischler–Napieralski reaction.^{7c,9}

We initiated research by exploration of a suitable substituent at the C-7 position. First, tetrahydro-β-carboline derivative **17**, which was prepared from tryptamine derivative by a three-step sequence,¹⁰ was subjected to halogenation conditions (Table 1). Disappointingly, fluorination¹¹ and chlorination¹² conditions resulted in formation of complex mixtures containing trace amounts of haloindolenines (entries 1 and 2). On the other hand, iodination^{7a,b} by using *N*-iodosuccinimide (NIS) proceeded smoothly to give the corresponding iodoindolenine **18c** in a high yield (entry 3). In addition, installation of the acetoxy group took place nicely by treatment of **17** with Pb(OAc)₄¹³ to give **18d** in <64% yield (entry 4).

On the other hand, treatment with *N*-(phenylseleno)phthalimide (*N*-PSP)¹⁴ in the presence of 10-camphorsulfonic acid (CSA) did not provide the desired product **18e**. Although we have succeeded in the introduction of iodo and acetoxy groups at the C-7 position, these compounds could not be converted to the desired lactam **19** under a variety of conditions due to elimination of substituent.

With these unsuccessful results, we then examined the introduction of a more robust substituent, an allyl group, and the subsequent lactamization (Scheme 3). The desired allylation at the C-7 position proceeded smoothly when tetrahydro-β-carboline derivative **17** was subjected to palladium catalyzed allylation conditions reported by Rawal and co-workers¹⁵ to furnish allyl indolenine **20** as a 1.7:1 mixture of diastereomers.

Unexpectedly, the next lactamization reaction was strongly dependent on the relative stereochemistry between the allyl group and the ester side chain. After chromatographic separation, two

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