



# Michael Initiated Ring Closure (MIRC) reaction on in situ generated benzylidenecyclohexane-1,3-diones for the construction of chromeno[3,4-*b*]quinoline derivatives

Abu T. Khan\*, Deb Kumar Das

Department of Chemistry, Indian Institute of Technology Guwahati, Guwahati 781 039, India

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## ABSTRACT

One-pot synthesis of chromeno[3,4-*b*]quinoline derivatives have been achieved in good yields through Michael Initiated Ring Closure (MIRC) by employing three-component condensation of aromatic aldehydes, 3-aminocoumarins, and cyclic 1,3-diketones in the presence of catalytic amount of *p*-toluenesulfonic (*p*-TSA) acid in ethanol under reflux condition. The salient features of this protocol are: simple reaction procedure, shorter reaction time, good yields, avoidance of aqueous work-up, and column-chromatographic separation. The merit of this process is highlighted by its high bond efficiency of producing three new bonds and one stereocenter in a single operation.

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The Michael Initiated Ring Closure (MIRC) reaction represents an elegant approach, which has been used extensively for the construction of cyclopropane rings,<sup>1</sup> carbocyclic compounds,<sup>2</sup> and small/medium sized nitrogen<sup>3</sup> or oxygen<sup>4</sup> containing heterocyclic compounds. The MIRC reaction strategy can also be cleverly achieved through one-pot multicomponent reaction, which is gaining interest to the synthetic organic chemists in recent times.<sup>5</sup> Multi-component reactions (MCRs) play an important role in the modern synthetic organic chemistry as they generally occur in a single pot and exhibit high atom-economy and selectivity.<sup>6</sup> They also provide a powerful synthetic tool for the synthesis of diverse and complex molecules as well as small and drug-like heterocycles.<sup>7</sup> We have perceived that cyclic 1,3-diketones may react with various aromatic aldehydes in the presence of a suitable acid catalyst to generate benzylidenecyclohexane-1,3-dione derivatives, which might be reacted instantly with carbon nucleophile such as 3-aminocoumarin through Michael type reaction followed by ring closure reaction leading to chromeno[3,4-*b*]quinoline derivatives. The similar synthetic strategy has also been demonstrated by others for the construction of 4-aza-2,3-didehydropodophyllotoxin<sup>8a</sup> and tricyclic dihydropyridine analogues,<sup>8b</sup> and pharmacological properties of these compounds have also been studied. Compounds containing 3-aminocoumarin framework are found in many natural products and some of them are used as antibiotic and antiviral

agent.<sup>9,10</sup> For example, novobiocine is a 3-aminocoumarin derived antibiotic which acts as an ATP-competitive inhibitor of the gyrase B subunit, blocking the negative super-coiling of relaxed DNA.<sup>9d,10</sup> On the other hand, the pyrido[2,3-*c*]coumarin skeleton constitutes the backbone of santiagonamine (**B**), an alkaloid (Fig. 1).<sup>11</sup> As a result, there is a continuing effort to prepare this class of compounds for biological studies. *p*-Toluenesulfonic acid (*p*-TSA) is a readily available chemical which has been used extensively in place of mineral acids. In recent times, it was also used as an efficient acid catalyst for the synthesis of 4(3*H*)-quinazolinones,<sup>12</sup> the regioselective nitration of phenols<sup>13</sup> and the carbonylation of formaldehyde.<sup>14</sup> As a part of our ongoing efforts to develop multi-component reactions to access potentially bioactive scaffolds,<sup>15</sup> we would like to report one-pot three-component reaction for the synthesis of chromeno[3,4-*b*]quinoline derivatives through MIRC reaction using condensation of aromatic aldehydes, cyclic 1,3-dicarbonyl compounds and 3-aminocoumarins under reflux condition in ethanol using *p*-TSA catalyst as shown in Scheme 1.

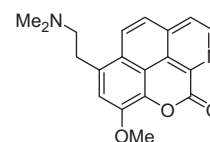
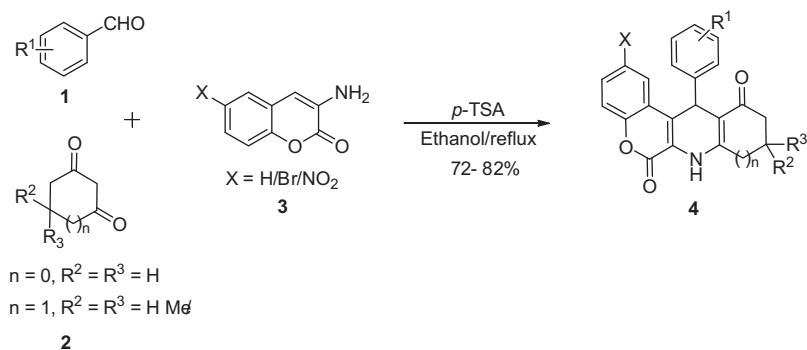


Figure 1. The naturally occurring biologically active alkaloid santiagonamine (**B**).

\* Corresponding author. Tel.: +91 361 2582305; fax: +91 361 2582349.

E-mail address: [atk@iitg.ernet.in](mailto:atk@iitg.ernet.in) (A.T. Khan).



**Scheme 1.** One-pot three-component condensation reaction for the synthesis of chromeno[3,4-*b*]quinoline derivatives.

**Table 1**  
Optimization of reaction conditions for the synthesis of chromeno[3,4-*b*]quinoline derivative **4a**<sup>a</sup>

Entry	Catalyst	Solvent	Catalyst (mol %)	Reaction condition	Time (h)	Yield <sup>b</sup> (%)
1	<i>p</i> -TSA	EtOH	5	Reflux	12	54
2	<i>p</i> -TSA	EtOH	10	Reflux	12	68
3	<i>p</i> -TSA	EtOH	20	Reflux	7	77
4	<i>p</i> -TSA	EtOH	30	Reflux	7	78
5	<i>p</i> -TSA	MeOH	20	Reflux	12	62
6	<i>p</i> -TSA	CH <sub>3</sub> CN	20	Reflux	12	66
7	<i>p</i> -TSA	H <sub>2</sub> O	20	Reflux	10	42
8	ZnCl <sub>2</sub>	EtOH	20	Reflux	12	38
9	SiO <sub>2</sub>	EtOH	20	Reflux	12	28
10	Et <sub>3</sub> N	EtOH	20	Reflux	12	24
11	Piperidine	EtOH	20	Reflux	12	26
12	Acetic acid	EtOH	20	Reflux	24	22
13	HCl	EtOH	20	Reflux	24	00
14	No catalyst	<i>n</i> -BuOH	0	Reflux	24	00
15	<i>p</i> -TSA	<i>n</i> -BuOH	20	Reflux	4.5	75

<sup>a</sup> Reaction conditions: benzaldehyde, dimedone and 3-aminocoumarin were taken in 1:1:1 ratio.

<sup>b</sup> Isolated yields.

**Table 2**  
Scope of various substituted chromeno[3,4-*b*]quinoline derivatives<sup>a</sup>

Entry	Aldehyde	1,3-Cyclic diketones	3-Aminocoumarin	Product	Time (h)	Yield <sup>b</sup>
1				 <b>4a</b>	7	77
2				 <b>4b</b>	8	78
3				 <b>4c</b>	8	82

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