

Novel biotransformation of pentacyclic triterpenoid acids by *Nocardia* sp. NRRL 5646

Jian Zhang,^a Zhi-Hong Cheng,^a Bo-Yang Yu,^{a,*} Geoffrey A. Cordell^b and Samuel X. Qiu^{c,*}

^aDepartment of Complex Prescription of TCM, China Pharmaceutical University, Nanjing 210038, People's Republic of China

^bDepartment of Medicinal Chemistry and Pharmacognosy, College of Pharmacy, University of Illinois at Chicago, 833 South Wood Street, Chicago, IL 60612, USA

^cDepartment of Chemistry, PO Box 1134, Washington University, One Brookings Drive, St. Louis, MO 63130, USA

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Abstract—Six pentacyclic triterpene acids, ursolic acid, oleanolic acid, betulinic acid, 23-hydroxybetulinic acid, glycyrrhetic acid, and senegenin, were metabolized by the microbe *Nocardia* sp. NRRL 5646 to selectively furnish their corresponding 28-methyl esters. Notably, ursolic acid (**1**) was converted to oleanolic acid methyl ester (**4**) via two intermediates, oleanolic acid (**2**), and ursolic acid methyl ester (**3**), which are formed by participation of 'retro-biosynthetic' methyl migration from C-19 to C-20. Senegenin (**11**) was selectively converted to a nortriterpene methyl ester, senegenic acid methyl ester (**12**), via an unprecedented C–C bond cleavage. The stereochemical assignments of compounds **11** and **12** were made unambiguously for the first time using 2D NMR spectroscopy. © 2005 Elsevier Ltd. All rights reserved.

Natural products (NPs) are an incredibly diverse group of small (usually molecular weight less than 1500 Da) organic compounds isolated from a variety of natural sources, principally plants. The reason that NPs capture the imagination of organic chemists and pharmaceutical scientists is because of their well documented and wide ranging biological activities and their skeletal diversity and intriguing functional group characteristics, which render them as indispensable leads for probing biological system status and for drug discovery with new bioassay systems. Recently, we demonstrated that microbial transformation could be a fruitful tool to enhance the structural diversity of natural triterpene glycosides.¹ As a continuation of these efforts to expand the structural diversity of natural products using microbial transformation, a group of pentacyclic triterpene acids (TAs) were chosen for biotransformation study. TAs are very widely distributed in the plant kingdom and have been shown to possess a wide range of biological activities. For example, ursolic acid (**1**) and betulinic acid (**5**) pos-

sess anti-cancer and anti-HIV activities.^{2,3} Betulinic acid is in preclinical development. It is envisioned that biotransformation of these TAs may provide analogs, which could be utilized for screening for new activities or for studying SAR. A panel of microbes was utilized to screen for their ability to transform the TAs. *Nocardia* sp. NRRL 5646 was found to convert these substrates efficiently to less polar metabolites. This organism is widely used for transformation because of its broad spectrum of enzymatic capability, including carboxylic acid and aldehyde reduction, phenol methylation, and the noteworthy skeleton rearrangement of quinovic acid glycosides via methyl migration.¹ In this report, the biotransformation results of six triterpenoid acids, ursolic acid (**1**), oleanolic acid (**2**), betulinic acid (**5**), 23-hydroxybetulinic acid (**7**), glycyrrhetic acid (**9**), and senegenin (**11**), by *Nocardia* sp. are described. Each of the six triterpene acids was incubated on a preparative scale with the *Nocardia* sp. according to the standard, two-stage fermentation protocol.^{1,4} The work-up and isolation of the metabolites were carried out as described previously.¹

Keywords: Biotransformation; *Nocardia* sp. NRRL 5646; Pentacyclic triterpenoid acids; 'Retro-synthetic' methyl migration; Senegenin; C–C bond cleavage; 2D NMR spectroscopy; Natural products.

* Corresponding authors. Tel.: +1 636 346 4322; fax: +1 636 536 4294 (S.X.Q.); tel.: +86 25 8530 2355; fax: +86 25 8331 3080 (B.-Y.Y.); e-mail: sqiu@wustl.edu

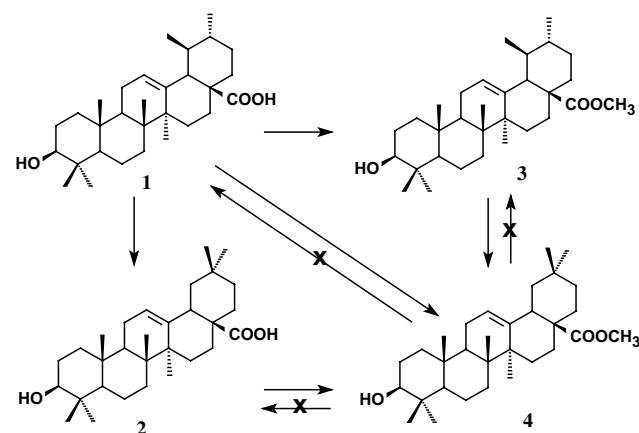
A 200 mg sample of **1** (C₃₀H₄₈O₃, *M_r* = 456) was used for the scale-up incubation and three metabolites, coded MT-1 (**2**), MT-2 (**3**), and MT-3 (**4**), were formed from the biotransformation, as evident from the HPTLC analysis. Silica gel chromatographic separation of the

transformation residue resulted in the isolation of compounds **2** (21%), **3** (9%), and **4** (13% yield).

MT-1 (**2**) was obtained as a white powder. The positive ion ESI-MS showed a quasimolecular ion at m/z $[M+H]^+$ 457.2, and a molecular formula $C_{30}H_{48}O_3$ was deduced from the HR-ESI-MS, indicating that **2** has the same molecular formula as that of the parent compound **1**. The structure of **2** was assigned as oleanolic acid by comparison of its 1H and ^{13}C NMR spectra with literature data,⁵ as well as co-TLC comparison with an authentic sample.

MT-2 (**3**) and MT-3 (**4**) were isolated as colorless powders, and their structures were readily assigned to ursolic acid methyl ester and oleanolic acid methyl ester, respectively, by comparison of their 1H and ^{13}C NMR data with the literature.^{6–9} Thus methylation at the C-28 carboxylic acid of ursolic acid (**1**) occurred through catalysis by the enzyme system of *Nocardia* sp. to form the corresponding ester **3**. It is also notable (Scheme 1) that two, skeleton-rearranged metabolites, oleanolic acid (**2**) and oleanolic acid methyl ester (**4**), were also obtained from this biotransformation involving a methyl migration. Biogenetically, the ursane and oleanane skeletons share a common intermediate, and the ursane nucleus is biosynthesized from an oleanane precursor via a methyl migration from C-20 to C-19.¹⁰ The current observation confirms the unique ability of this *Nocardia* sp. to catalyze a 'retro-biosynthetic' conversion from the ursane to the oleanane skeleton. This is the second example of a 'retro-biosynthetic' transformation with this organism. A similar conversion was observed in the biotransformation utilizing quinovic acid glycosides as substrates.¹

To investigate the mechanism of this biotransformation, several experiments were performed. Both oleanolic acid (**2**) and ursolic acid methyl ester (**3**) were found to be converted to oleanolic acid methyl ester (**4**) when incubated with *Nocardia* sp. under the same conditions. On the other hand, no appreciable reaction could be detected for substrate **4** after extended incubation for four days. This implies that ursolic acid (**1**) is converted step-



Scheme 1. Biotransformation of ursolic acid (**1**), oleanolic acid (**2**) and their corresponding methyl esters (**3** and **4**) by *Nocardia* sp.

wise to the final product oleanolic acid methyl ester (**4**) via two intermediates, namely, oleanolic acid (**2**), and ursolic acid methyl ester (**3**), which are further converted to oleanolic acid methyl ester (**4**) through extended incubation.

The source of the methyl group in the enzyme system of *Nocardia* sp. (inter- or intramolecular) has yet been defined. In subsequent experiments, simple alcohols such as ethanol, glycerol, isopropanol, and butanol showed no effect on the biotransformation, implying that methylation may not be catalyzed by the general lipases or esterases, which might be inhibited by exogenous alcohols. Thus, it would be suggested that methyl ester formation could most likely result from an S-adenosylmethionine-dependent methyltransferase, a member of a large family of enzymes with broad biological activity including transmethylation.¹¹

Betulinic acid (**5**) and 23-hydroxybetulinic acid (**7**) (Fig. 1) are members of the so-called 'lupane-type' pentacyclic triterpenes, which were shown to exhibit potent apoptotic and anti-HIV activities.^{12,13} Incubation of **5** and **7** with *Nocardia* sp. afforded their corresponding methyl esters, betulinic acid methyl ester (**6**) and 23-hydroxybetulinic acid methyl ester (anemosapogenin methyl ester, **8**), respectively. The structures were identified on the basis of their almost super-imposable NMR data (Table 1) with literature values for betulinic acid methyl ester,^{9,14} and 23-hydroxybetulinic acid.¹⁵

Glycyrrhetic acid (**9**), a potent and selective inhibitor of 11β -hydroxysteroid dehydrogenases,¹⁶ is the aglycone of the saponins from the well-known Traditional Chinese Medicine licorice (*Glycyrrhiza glabra*). Similarly, **9** was converted to its methyl ester, glycyrrhetic acid methyl ester (methyl glycyrrhetinate, **10**) by incubation

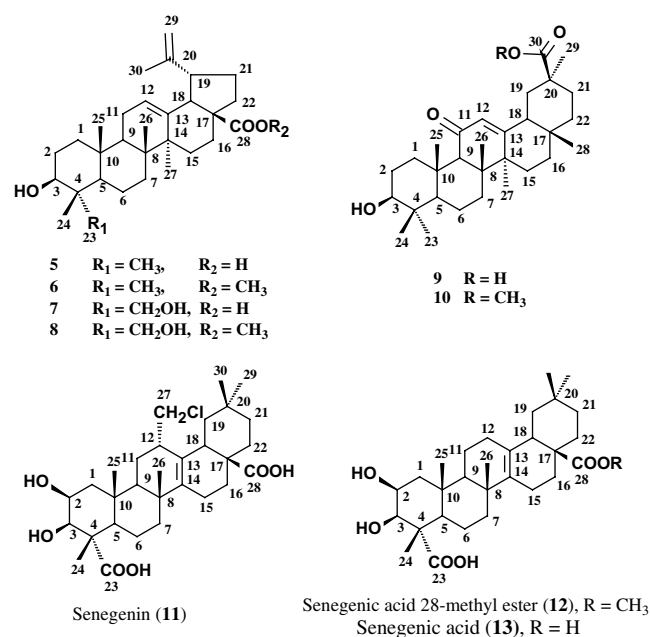


Figure 1. Structures of pentacyclic triterpene acids biotransformed by *Nocardia* sp.

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