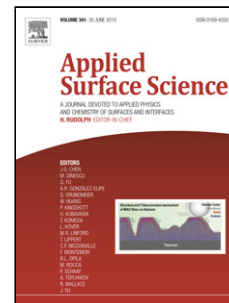


## Accepted Manuscript

Title: H<sub>2</sub>O<sub>2</sub>-assisted photocatalysis on flower-like rutile TiO<sub>2</sub> nanostructures: Rapid dye degradation and inactivation of bacteria

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Hierarchically assembled rutile TiO<sub>2</sub> was synthesized at room temperature. Hydrothermal treatment enhanced the crystallinity, while morphology was maintained. Hydrothermal treatment also led to larger crystallites and a lower surface area. Effective *K. pneumoniae* killing and MO degradation were achieved with the use of H<sub>2</sub>O<sub>2</sub>. Higher crystallinity enhanced the reaction rate in the presence of H<sub>2</sub>O<sub>2</sub>.

## Equation Chapter 1 Section 1 H<sub>2</sub>O<sub>2</sub>-assisted photocatalysis on flower-like rutile TiO<sub>2</sub> nanostructures: Rapid dye degradation and inactivation of bacteria

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### Abstract

Hierarchically assembled flower-like rutile TiO<sub>2</sub> (FLH-R-TiO<sub>2</sub>) nanostructures were successfully synthesized from TiCl<sub>4</sub> at room temperature without the use of surfactants or templates. An initial sol-gel synthesis at room temperature allowed long-term hydrolysis and condensation of the precursors. The resulting FLH-R-TiO<sub>2</sub> possessed relatively high crystallinity (85 wt%) and consisted of rod-shaped subunits assembling cauliflower-like superstructures. Hydrothermal evolution of FLH-R-TiO<sub>2</sub> nanostructures at different temperatures (150, 200 and 250 °C) was followed by means of X-ray diffraction, transmission and scanning electron microscopy.

These FLH-R-TiO<sub>2</sub> nanostructures were tested as photocatalysts under simulated daylight (full-spectrum lighting) in the degradation of methyl orange and in the inactivation of a multiresistant bacterium, *Klebsiella pneumoniae*. The effects of hydrothermal treatment on the structure, photocatalytic behavior and antibacterial activity of FLH-R-TiO<sub>2</sub> are discussed.

### Keywords

Hierarchical rutile TiO<sub>2</sub>; surface morphology; band gap energy; X-ray photoelectron spectroscopy; photocatalytic activity; antibacterial activity

## 1. Introduction

By virtue of its unique physicochemical properties, TiO<sub>2</sub> is one of the most intensively studied semiconductor oxides. It has been widely used in photocatalysis and photovoltaics and for the production of self-cleaning surfaces [1]. TiO<sub>2</sub> has four naturally occurring polymorphs: anatase, brookite, TiO<sub>2</sub>(B) and the most stable form, rutile. Anatase is the most widely studied photocatalyst since it exhibits high photoactivity, whereas generally lower photocatalytic activity is noticed for rutile [2].

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