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# Ultrathin cobalt-alloyed barrier layers for copper metallization by a new seeding and electroless-deposition process

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#### 1. Introduction

Electroless deposition has been integrated into commercialized electroplating tools for almost a decade to directly form (or repair sputter deposited) copper-seeding layers, and recently has received intensive attention due to its potential for the sequential deposition of a nanostructured cobalt-based diffusion barrier and copper wire into sub-100-nm trenches in a conformal and in-line manner [1].

A catalyst, typically palladium, has to be deposited on dielectric layers prior to electroless deposition of the diffusion barrier layer. Conventionally, the catalyst is produced as colloidal particles by either sensitization-activation or displacement-activation in an aqueous chemical solution that contains palladium species [2,3]. However, the palladium particles tend to agglomerate into clusters of sizes typically exceeding tens to hundreds of nm [4,5], and thus are inappropriate to initiate electroless deposition of the copper metallization layer for the sub-60-nm integrated circuits. Obviously, a perquisite for employing electroless deposition to grow a barrier layer with a thickness of less than 10 nm on dielectric layers is to develop a seeding method, which has the capacity to grow catalytic seeds with a size of only a few nm without agglomeration.

We had previously developed a seeding process that grew 3-nm-sized catalytic particles (e.g., Co or Ni) on SiO<sub>2</sub> and HOSP<sup>TM</sup>

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#### ABSTRACT

Pioneering activation-seeding processes grow catalytic particles with sizes exceeding 10 nm due to agglomeration, and thus are unable to act as a template for electroless deposition of a barrier layer with a thickness of 10 nm or less, which is desperately needed for the incoming ULSI copper interconnecting technology. In this work, the capacity of a seeding process to grow a continuous Co-P barrier layer of 8-nm thickness on thermally oxidized SiO<sub>2</sub> layers using electroless deposition will be demonstrated. The Co-P barrier layer works effectively in retarding (a) Cu agglomeration and (b) Cu diffusion into the dielectric layer subjected to thermal annealing. Evidently, thermal stability of the Cu film on SiO<sub>2</sub> is markedly strengthened by interposing the 8-nm-thick barrier layer. The mechanism of the interposed barrier layer in enhancing thermal stability of the metallization layer is currently under investigation.

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dielectric layers [6]. It involved a sequence of wet-processing steps, including chemical-solution surface functionalization and seedadsorption treatments. A vacuum-plasma surface pre-treatment was late incorporated into the wet-processing steps, dramatically increasing the seed density by a magnitude of over 20 times [7]. The purpose of this work is using the refined seeding process to sequentially initiate electroless deposition of an ultrathin (<10 nm) Co-P barrier and copper metallization layer on the Si/SiO<sub>2</sub> underlying substrates. Enhancement of thermal stability of the metallization layer by interposing the barrier layer will be demonstrated, and the mechanism of the seeding refinement will be investigated by surface-bonding analysis.

#### 2. Experimental procedure

(100)-oriented, p-type silicon wafers with a 100-nm-thick thermally oxidized SiO<sub>2</sub> layer were used as substrates. First, cobalt catalytic particles were grown on one type of the SiO<sub>2</sub> dielectric layers without surface plasma pre-treatment by the processing steps presently previously [6], referred as the reference samples. Briefly, the dielectric samples were immersed into an ammonia-hydrogen peroxide aqueous solution, the so-called SC-1, to modify the surfaces into a negatively charged state, which had the capacity to attract metallic cations (Co<sup>2+</sup>) from a metal-salt (CoSO<sub>4</sub>) aqueous solution. The adsorbed metallic cations were reduced to neutral species (Co) in a reducing agent and used as a catalyst for electroless deposition. Then, the catalyst-bearing dielectric samples were immersed into an electroless-deposition solution (80 °C and pH

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8.5) to trigger the growth of Co-P deposits; the times of deposition ranged from 10 s to a few minutes.

For another type of the dielectric samples, the as-prepared dielectric layers were pre-treated by N<sub>2</sub>/H<sub>2</sub> vacuum plasma at a total pressure of 50 Pa. The change in surface-bonding states was elucidated by X-ray photoelectron spectroscopy (XPS). It is worth noting that the dielectric layers have to be treated by the plasma with a proper  $N_2/H_2$  flow ratio and dosage so that their surfaces are effectively activated to increase the seeding density. The dielectric layers after the plasma pre-treatments were then subjected to the seeding and electroless-deposition treatments using the bath and deposition conditions that were identical to those of the reference samples. Then, the differences in the seeding and progress of the Co-P deposit-plating, ultimately leading to the formation of a continuous Co-P barrier layers, between the two types of the dielectric layers were compared by field-emission scanning electron microscopic (SEM) top-view imaging. Distribution of the seed particles was analyzed by transmission electron microscopy (TEM); thickness of the barrier layers was measured by cross-sectional SEM, reconfirming by a step profilometer ( $\pm 10\%$  accuracy).

The Co-P barrier layers were then used as catalysts to promote electroless deposition of copper layers using a cupric sulfate-formaldehyde based solution. Thermal stability of the Si/SiO<sub>2</sub>/Co-P (8 nm)/Cu (40 nm) sample layers after isothermal annealing ( $\geq$ 400 °C) at argon atmosphere for 30 min, using Si/SiO<sub>2</sub>/Cu (40 nm) layered structures as the competing samples, was evaluated by SEM, X-ray diffractometry (XRD), and the changes of overall sheet resistances of the metallization layers.

#### 3. Results and discussion

Plan-view TEM and high-resolution SEM images were used to characterize the distributions and surface morphologies of the seed (Co) particles and Co-P deposits on the two types of the catalystbearing SiO<sub>2</sub> dielectric layers. It can be seen from the plan-view TEM micrograph inserted in Fig. 1(a) that the Co seeds, with a tiny size of typically 3 nm, were sparsely dispersed on the reference SiO<sub>2</sub> dielectric layers which were free of the plasma pre-treatment. Importantly, the size of 3 nm obtained herein is yet to be achieved by both the ordinary and other refined activation processes [4,5,8]. The seeds presented in the inset of Fig. 1(a) provided a platform for electroless deposition of Co-P barrier deposits, evolving from equiaxed islands finally into a continuous film structure [6]. However, after electroless deposition for 10 s, the scarcity of the seeds, as observed in the inset of Fig. 1(a), catalyzed a scattering distribution of the Co-P islands with an average size of  ${\sim}25\,\text{nm}$  [see Fig. 1(a)]. It could be referred from the number of the isolated particles in Fig. 1(a) that the density of the catalytic particles was approximately  $1.2 \times 10^{14} \text{ m}^{-2}$ . Therefore, prolonged deposition for duration greater than 60s was required for the Co-P islands to coalesce into a continuous barrier layer, whose grain size and thickness were both greater than 40 nm.

Fig. 1(b) displays an SEM micrograph of the Co-P deposits, formed on another type of the catalyst-bearing SiO<sub>2</sub> dielectric layers by electroless deposition for 10 s. Prior to the seeding treatment which was identical to that of the reference SiO<sub>2</sub>, the dielectric surfaces were pre-treated optimally by the N<sub>2</sub> (80%)/H<sub>2</sub> (20%) vacuum plasma. This high-magnification SEM micrograph demonstrates that the plasma surface pre-treatment is a key factor for the considerable increase in the seed density to  $5.0 \times 10^{15}$  m<sup>-2</sup>, as compared to that of the reference samples  $(1.2 \times 10^{14} \text{ m}^{-2})$ ; the increment is by more than 40 times. Cross-sectional SEM imaging, as presented later in Fig. 5(b) indicated that the refinement of the seeding process ultimately led to the achievement of a continuous barrier layer with a thickness of 8 nm (±10%).



**Fig. 1.** (a) Top-view SEM micrograph showing the morphology of Co-P deposits on SiO<sub>2</sub> layers without plasma pre-treatment after 10 s of electroless deposition catalyzed by the Co seeds shown in the inserted TEM micrograph; (b) the corresponding Co-P deposits on plasma pre-treated SiO<sub>2</sub> layers.

The N<sub>2</sub>/H<sub>2</sub> plasma surface pre-treatment is the crucial experimental step to cause a significant increase of the seed density. To clarify the mechanism of the seeding improvement by the plasma pre-treatment, surface-bonding structures of the SC-1 treated SiO<sub>2</sub> layers without and with the plasma pre-treatments were investigated by examining the oxygen (1s) and nitrogen (1s) XPS core-level spectra. The spectrum (dotted line) in Fig. 2 shows that the reference SiO<sub>2</sub> layers only subjected to the SC-1 solution treatment clearly exhibited a symmetry peak at 533.0 eV which corresponded to the O 1s (O–Si–O) bonds purely from  $SiO_2$  [9]. As expected from the dotted plot inserted in Fig. 2, the N (1s) signals associated with the reference samples were at the background level. The spectrum (solid line) in Fig. 2 presents that the plasma pre-treated SiO<sub>2</sub> layers after the SC-1 solution treatment yielded an asymmetric peak with the tail toward the low-energy side, implying changes in surface bonding structures. The significant chemical shift in the O(1s) peak is due to the breakdown of the O-Si-O intrinsic bonds (533.0 eV) by the  $N_2/H_2$  plasma treatment. Deconvolution of the newly formed asymmetric peak (see the deconvoluted solid lines in Fig. 2) revealed that the surfaces contained



Fig. 2. O 1s and N 1s (insert) XPS spectra for the SiO<sub>2</sub> layers after the solution treatment without (dotted lines) and with (solid lines) the  $N_2/H_2$  plasma pre-treatment.

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