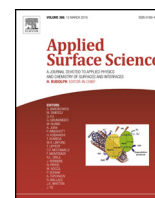




Contents lists available at ScienceDirect

Applied Surface Science

journal homepage: www.elsevier.com/locate/apsusc

Optimisation of anatase TiO₂ thin film growth on LaAlO₃(0 0 1) using pulsed laser deposition

K. Krupski^{a,*}, A.M. Sanchez^a, A. Krupski^{a,b}, C.F. McConville^a^a Department of Physics, University of Warwick, Coventry CV4 7AL, UK^b Faculty of Science, SEES, University of Portsmouth, Portsmouth PO1 3QL, UK

ARTICLE INFO

Article history:

Received 15 January 2016
Received in revised form 23 February 2016
Accepted 24 February 2016
Available online xxx

Keywords:

Anatase
Titanium dioxide
Lanthanum aluminate
Pulsed laser deposition (PLD)
Reflection high-energy electron diffraction (RHEED)
Low-energy electron diffraction (LEED)
X-ray diffraction (XRD)
Atomic force microscopy (AFM)
Scanning tunneling microscopy (STM)
Scanning transmission electron microscopy (STEM)
Growth
Thin film growth

ABSTRACT

Optimisation of epitaxial anatase TiO₂ thin films grown on LaAlO₃(0 0 1) substrates was performed using ultra-high vacuum based pulsed laser deposition (PLD) and studied by in-situ reflection high-energy electron diffraction (RHEED). In addition, ex-situ X-ray diffraction (XRD), atomic force microscopy (AFM), and scanning transmission electron microscopy (STEM) were performed to characterise the bulk properties of these thin films. The deposited TiO₂ thin film is demonstrated to have anatase phase and bonded directly to the LaAlO₃(0 0 1) substrate. In a separate ultra-high vacuum system low-energy electron diffraction (LEED) and scanning tunneling microscopy (STM) measurements were performed and a well-ordered two-domain (1 × 4) and (4 × 1) reconstruction of anatase surface was observed. Analysis of the STM measurements indicates the coexistence of atomic steps of both 2.5 Å and 5.0 Å, confirming the existence of two TiO₂ domains. The atomic resolution STEM images reveal that the TiO₂/LaAlO₃ interface to be terminated with LaO layer and that the anatase-TiO₂ reconstruction was found to be stable during the film growth.

© 2016 Published by Elsevier B.V.

1. Introduction

Titania, TiO₂, has a wide range of applications and a large number of extremely interesting properties. A key 'low-tech' application that stems not only from its optical properties, but also its non-toxicity, is use as a whitening agent in paints and paper. However, applications that more directly stem from its surface properties, as well as the bulk, arise in heterogeneous catalysis including photo-catalysis [1,2], the photovoltaic effect [3,4], and collar cells [5], and it is these applications that are at least part of the reason why TiO₂ is almost certainly the most studied of all oxides surface [6]. Until now, the majority of studies of TiO₂ have been performed on the rutile phase [7–9]. Rutile is the thermodynamically equilibrium phase of TiO₂ at ambient pressures, but two other isomorphs, anatase and brookite also occur naturally. Crucially

anatase appears to be the equilibrium phase for small particles with dimension less than 11 nm [2]. It is therefore generally believed that anatase is the active component in many titania based heterogeneous catalysts [4,10] and in current solar cell applications based on nano-crystalline material. As such, there is a clear need to gain a better understanding of the anatase surface structure and the role of the TiO₂ growth conditions (e.g., substrate temperature, oxygen pressure—during growth and annealing, etc.). Thin films of TiO₂ can be formed on a wide variety of substrates including oxide surfaces such as: MgO [11], SrTiO₃ [11,12], and LaAlO₃ [11–13]. Different deposition methods such as reactive sputter deposition [14], oxygen plasma assisted molecular beam epitaxy (PAMBE) [12,15] or pulsed laser deposition (PLD) [13,16–22] have all been used to fabricate the anatase phase. Until now of the substrates examined, LaAlO₃ (LAO) gives the best coherency owing to its relatively small lattice mismatch with anatase. In the bulk phase, anatase TiO₂ has a tetragonal structure with a lattice parameters $a = 0.3776$ nm and $c = 0.9486$ nm, while LaAlO₃ can be described as a pseudo-cubic perovskite with a lattice parameter $a = 0.3792$ nm, leading to a mismatch of only 0.4% when TiO₂ is grown epitaxially on the (0 0 1)

* Corresponding author.

E-mail addresses: k.j.krupski@warwick.ac.uk, k.j.krupski@gmail.com (K. Krupski).

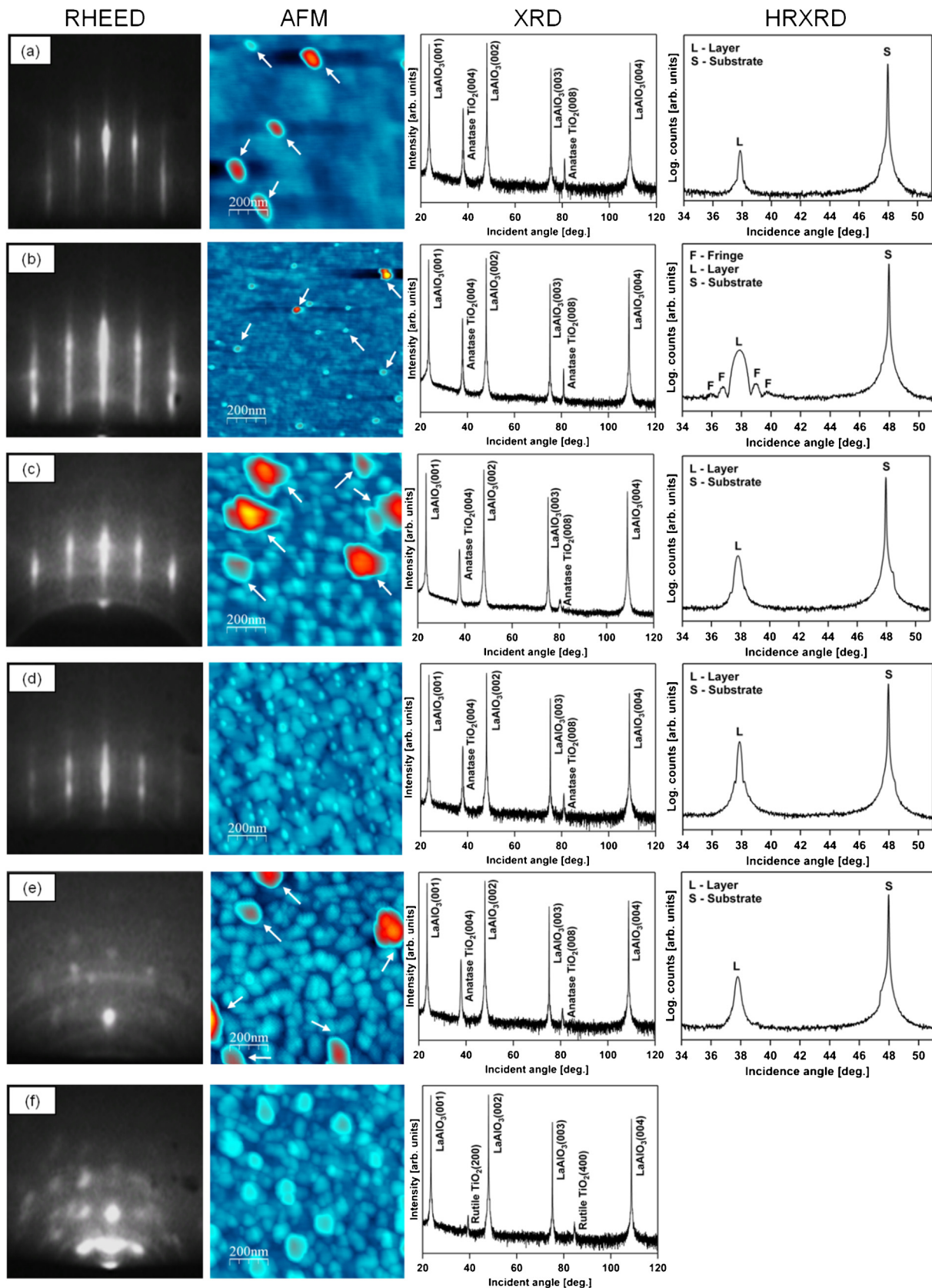


Fig. 1. RHEED, AFM, XRD and HRXRD results of TiO₂ on LaAlO₃(0 0 1)—Samples A–F (described in Table 1). The quality of the TiO₂ thin films depends strongly on the growth conditions. (a) Sample A—different laser energy; (b) sample B—different sample–target distance; (c) sample C—oxygen pressure changed during sample annealing; (d) sample D—sample temperature during growth T = 670 °C; (e) sample E—sample temperature during growth T = 690 °C; (f) sample F—different laser pulse frequency.

Download English Version:

<https://daneshyari.com/en/article/5353426>

Download Persian Version:

<https://daneshyari.com/article/5353426>

[Daneshyari.com](https://daneshyari.com)