

# Embedded structure of silicon monoxide in SiO<sub>2</sub> films

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Received 2 April 2007; accepted 27 September 2007

Available online 2 October 2007

## Abstract

The structure of SiO<sub>x</sub> ( $x = 1.94$ ) films has been investigated using both X-ray photoelectron spectroscopy (XPS) and time-of-flight secondary ion mass spectrometry (TOF-SIMS). The SiO<sub>x</sub> films were deposited by vacuum evaporation. XPS spectra show that SiO<sub>1.94</sub> films are composed of silicon suboxides and the SiO<sub>2</sub> matrix. Silicon clusters appeared only negligibly in the films in the XPS spectra. Si<sub>3</sub>O<sup>+</sup> ion species were found in the TOF-SIMS spectra with strong intensity. These results reveal the structure of the films to be silicon monoxide embedded in SiO<sub>2</sub>, and this structure most likely exists as a predominant form of Si<sub>3</sub>O<sub>4</sub>. The existence of Si–Si structures in the SiO<sub>2</sub> matrix will give rise to dense parts in loose glass networks.

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PACS : 61.43.Fs; 68.49.Sf; 68.49.Uv; 68.55.Jk

Keywords: SiO<sub>x</sub> films; Silicon suboxides; X-ray photoelectron spectroscopy; Secondary ion species; Time-of-flight secondary ion mass spectrometry

## 1. Introduction

SiO<sub>x</sub> films are well known and extensively studied materials [1–6] that are very attractive for applications as dielectrics [7,8] and transparent gas barrier films [9,10]. In particular, additional interest in these materials has arisen out of curiosity in their structure in the nanoscale due to the non-stoichiometric silicon oxides [11–14].

Silica glass is thought to be comprised of a randomly networked-structure of Si–O bonds with four-coordination number of silicon atoms surrounded by oxygen atoms and linked to other units through Si–O–Si bridges [1,6]. Although the ratio of silicon to oxygen atoms is 2.0 for the most-energetically stable, that is, fully-oxidized and stoichiometric silicon oxide, thin film compositions with ratios less than 2.0 are frequently used for applications in electronics [7] and optics [8]. These have the composition SiO<sub>x</sub> [1–6], wherein  $x$  is in the range  $0 < x < 2$ . Due to excess silicon atoms producing optical absorption in the visible range, compositions with  $x$  close to 2.0 are preferable in order to attain transparency. For instance, thin films with  $x$  between 1.9 and 2.0 have been used as transparent

gas barrier films [9,10]. In the Si–SiO<sub>2</sub> phase diagram, the intermediate oxidation states of silicon cannot be identified; it is not clear that silicon monoxide exists in the phase diagram [1,15,16]. There is also significant curiosity and interest in the structure of SiO<sub>x</sub> films, in particular, transparent SiO<sub>x</sub> films in the nanoscale.

Secondary ion species in time-of-flight secondary ion mass spectrometry (TOF-SIMS) spectra could play a powerful role in providing information on solid structures in the nanoscale [17,18]. SiO<sub>1.94</sub> films were investigated in this study using TOF-SIMS spectra in association with peak analysis in X-ray photoelectron spectroscopy (XPS) spectra [17,19].

## 2. Experimental

Two types of samples were used in this study: 25 nm-thick SiO<sub>2</sub> films and 94 nm-thick SiO<sub>x</sub> films. 25 nm-thick thermally-oxidized as well as plasma-enhanced CVD-prepared SiO<sub>2</sub> films on single-crystal n-type (1 0 0) silicon wafers (resistivity of 4–6 Ω cm) were received from Tohoku Semiconductor Co. SiO<sub>x</sub> films with 94-nm thickness were deposited onto poly-(ethylene terephthalate) (PET) films at room temperature using electron beam evaporation of a mixture of silicon and silicon dioxide. The vacuum pressure was  $5.0 \times 10^{-3}$  Pa. The atomic

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composition  $x$  of the films was determined by XPS and was shown to be  $1.94 \pm 0.03$ .

XPS measurements were performed using an ULVAC-PHI Quantera SXM spectrometer with an Al  $K\alpha$  X-ray source.

Sputter etching was accomplished using either  $\text{Ar}^+$  ions at 500 V or  $\text{C}_{60}^+$  ions at 10 kV rastered over a  $2 \text{ mm} \times 2 \text{ mm}$  area. This corresponds to an etching rate of about 0.02 nm/s for the  $\text{SiO}_2$  films. For the  $\text{C}_{60}^+$  ion source, a differentially pumped  $\text{C}_{60}$

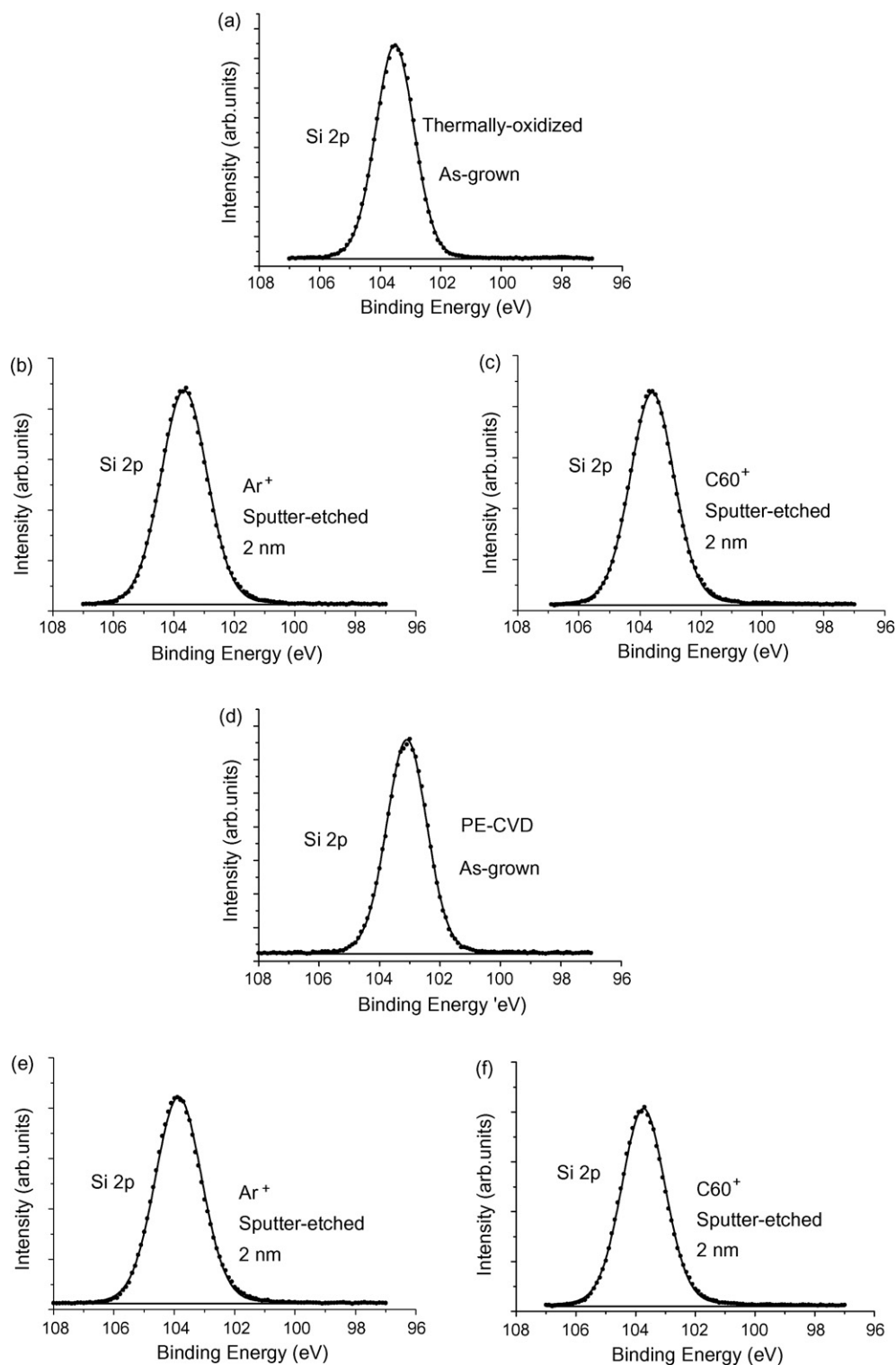


Fig. 1. XPS spectra from  $\text{SiO}_2$  films on Si: (a) as-grown surface of a thermally-oxidized film; (b) 2-nm depth from the surface for the thermally-oxidized film sputter-etched using 500 V  $\text{Ar}^+$  ions; (c) 2-nm depth from the surface for the thermally-oxidized film sputter-etched using 10 kV  $\text{C}_{60}^+$  ions; (d) as-grown surface of a CVD oxide film; (e) 2-nm depth from the surface for the CVD oxide film sputter-etched using 500 V  $\text{Ar}^+$  ions; and (f) 2-nm depth from the surface for the CVD oxide film sputter-etched using 10 kV  $\text{C}_{60}^+$  ions.

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