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Theoretical study of the Wittig, aza-Wittig and arsa-Wittig reactions of Me₃P=XH ylide (X= CH, N and As) with cyclic ketones in the gas phase and solution phase

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Abstract: A computational study of the comparative reactivity of the Wittig, aza-Wittig and arsa-Wittig reactions of cyclopentanone and cyclobutanone with Me₃P=XH ylide (X=CH, N and As) in the gas phase has been performed at B3LYP/6-31G** level of theory. Calculations have also been performed using integral equation formalism polarizable continuum model (IEFPCM), with THF solvent at B3LYP/6-31G** level of theory. In the gas phase as well as in solvent, these reactions were predicted to be two step processes with two complexes R and P (reactant complex and product complex), two transition states, one for addition (TS1) and the other for elimination (TS2) and two four-membered cyclic intermediates (INT1 and INT2), except that INT2 was not found in the aza-Wittig reaction. In our calculations, the Wittig reaction is found to be more exothermic and to involve a lower activation barrier to formation of INT1 than the aza and arsa-Wittig reactions. The activation barriers in the reactions of cyclopentanone are higher as compared to the reaction are overestimated in the solvent model as compared to gas phase reactions.

Keywords: Wittig reaction – aza-Wittig reaction – arsa-Wittig reaction – cyclic ketone – cyclopentanone – cyclobutanone

1. Introduction

In 1919, iminophosphoranes isoelectronic with methylenephosphoranes (Wittig reagent), were first synthesized by Staudinger and Meyer [1]. These have drawn much attention both experimentally and theoretically because of their wide application in organic synthesis. The aza-Wittig reaction [2-3] in which iminophosphoranes can react with carbonyls, is one of the best methods to form a -C=N- double bond (imine group). Imines ($R_2C=NR'$), one of the products from the aza-Wittig reaction are important intermediate compounds in organic synthesis due to their versatile chemical and physical properties and are also used in coordination chemistry [4-5].

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