



A study of the electronic structures and optical properties of CuXTe_2 ($\text{X} = \text{Al}, \text{Ga}, \text{In}$) ternary semiconductors

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ABSTRACT

Using first-principle calculations, the electronic and optical properties of chalcopyrite crystals CuXTe_2 ($\text{X} = \text{Al}, \text{Ga}, \text{In}$) are analyzed in detail. It is found that the prominent peaks of the dielectric functions and the refractive index significantly shift towards lower energies in the order of $\text{Al} \rightarrow \text{Ga} \rightarrow \text{In}$, and these compounds are characterized by similar optical spectra with some anisotropy effects. Present findings predict that CuGaTe_2 and CuInTe_2 are promising for photovoltaics due to their high absorption of solar radiations and photoconductivity in the visible range.

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1. Introduction

The ternary semiconductors with the general formula I–III–VI₂ have attracted considerable attention due to their widely applications in nonlinear optical (NLO) devices, detectors, solar cells etc. [1–4]. I–III–VI₂ chalcogenides are the most extensively used materials for photovoltaic applications, and there are numerous studies on these semiconductors, especially CuInSe_2 , due to its high conversion efficiency and low cost used for high-efficient solar cells. Jaffe and Zunger [5] have studied the chemical trends on a series of Cu-based ternary chalcopyrite semiconductors using the potential variation mixed basic (PVMB) approach. Laksari et al. [6] calculated the structural, electronic and optical properties of CuGaS_2 and AgGaS_2 using the full potential linear augmented plane wave (FP-LAPW) method and Reshak and Auluck [7] reported related electronic properties of the chalcopyrite compounds CuAlX_2 ($\text{X} = \text{S}, \text{Se}, \text{Te}$) using the same method. The optical properties of CuGaS_2 have been studied by Ahuja et al. [8] using the full-potential linear muffin-tin orbital (FPLMTO) method. Recently, Brik [9] and Zhou et al. [10] discussed the electronic and optical properties for series of CuXS_2 ($\text{X} = \text{Al}, \text{Ga}, \text{In}$) and CuAlX_2 ($\text{X} = \text{S}, \text{Se}, \text{Te}$) semiconductors.

Despite considerable efforts made by the previous studies, there are still no theoretical studies focusing on the I–III–VI₂ ternary semiconductors of CuXTe_2 ($\text{X} = \text{Al}, \text{Ga}, \text{In}$) up to now. In this work, we first analyze the electronic properties of the I–III–VI₂ ternary semiconductors of CuXTe_2 ($\text{X} = \text{Al}, \text{Ga}, \text{In}$) compounds, based on

first-principle calculations. Then, the optical properties of these compounds are discussed in more details and we found that they are characterized by similar optical spectra with some anisotropy effects.

2. Details of calculation

All calculations are performed using the ultrasoft pseudopotential method based on the density functional theory (DFT) [11] packaged in Cambridge Serial Total Energy Package (CASTEP) [12]. The exchange–correlation effects are treated by the Perdew–Berke–Ernzerhof (PBE) [13] functional within the generalized gradient approximation (GGA) [14]. The Brillouin-zone is sampled with the Monkhorst–Pack scheme ($8 \times 8 \times 4$ k-mesh) [15]. The optimization of the lattice constants is made by minimization of the total energy. We choose a cutoff energy of 800 eV which gives energy convergence less than 5.0×10^{-7} eV/atom. The maximum force tolerance, the maximum stress, and the maximum displacement tolerance are selected as 0.03 eV/nm, 0.05 GPa, 10^{-4} nm, respectively. The electronic configurations are taken as $\text{Al}: 3s^2 3p^1$, $\text{Ga}: 3d^{10} 4s^2 4p^1$, $\text{In}: 4d^{10} 5s^2 5p^1$, $\text{Cu}: 3d^{10} 4s^1$, $\text{Te}: 5s^2 5p^4$ in the calculations.

3. Results and discussion

3.1. Structural and electronic properties

Fig. 1 presents considered semiconductors of CuXTe_2 ($\text{X} = \text{Al}, \text{Ga}, \text{In}$) crystallized in the chalcopyrite structure, space group I-42d

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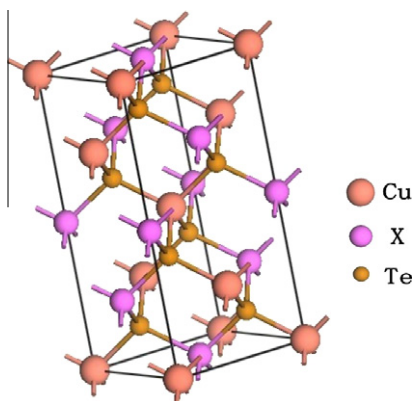


Fig. 1. A unit cell of CuXTe_2 ($X = \text{Al, Ga, In}$) crystals with the chalcopyrite structure.

Table 1

Crystal lattice constants a and c , as well as the volume (V) of CuXTe_2 ($X = \text{Al, Ga, In}$).

| | CuAlTe_2 | | | CuGaTe_2 | | CuInTe_2 | |
|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|
| | This work | Exp. ^a | Cal. ^b | This work | Exp. ^a | This work | Exp. ^a |
| $a(\text{\AA})$ | 6.10 | 5.964 | 6.022 | 6.109 | 5.994 | 6.315 | 6.167 |
| $c(\text{\AA})$ | 12.14 | 11.78 | 12.309 | 12.18 | 11.91 | 12.616 | 12.34 |
| $V(\text{\AA}^3)$ | 452 | 419 | 446 | 454 | 428 | 503 | 469 |

^a Ref. [18].

^b Ref. [17].

(No. 122), with four formula units in a unit cell. Each atom in this structure is fourfold coordinated, like in the zinc-blend or diamond crystal structures [16]. Calculated lattice parameters at ambient conditions are summarized in Table 1, and present results agree well with the available experimental data and other theoretical calculations listed in Table 1, considering the overestimation of GGA for the lattice constants. From Table 1, one can see that calculated lattice constants increase with increasing atomic number of the trivalent metal ion $\text{Al} \rightarrow \text{Ga} \rightarrow \text{In}$ for the CuXTe_2 series.

Based on optimized structures, the band structures of CuXTe_2 ($X = \text{Al, Ga, In}$) and the density of states (DOSs) for CuAlTe_2 are presented in Figs. 2 and 3, respectively. The band structures in Fig. 2a show that the highest energy of valence bands and the lowest energy of conduction bands are both located at gamma point, giving a direct band gap of 0.94 eV for CuAlTe_2 . This is in accordance with

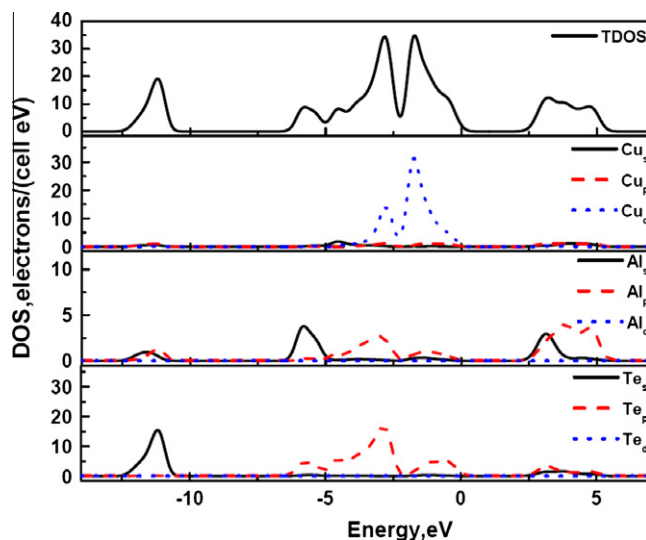


Fig. 3. Total and atom-projected density of states (DOSs) for CuAlTe_2 .

previous theoretical values of 1.066 eV [10] and 1.032 eV [17] (using the virtual-crystal approximation) and is underestimated in comparison with the experimental data of 2.06 eV [18]. To overcome the discrepancy between experimental and theoretical results, the so called scissor operator corrections [19] was applied for the compounds considered in the calculations. The experimental value of the band gap for CuGaTe_2 is about 1.23 eV, whereas our present calculations gave a value as about 0.088 eV, smaller than the value of 0.43 eV given in Ref. [20] (using the LDA), such a result may be attributed to different calculation methods. So a scissor operator of 1.142 eV was applied. At last, the experimental value of the band gap for CuInTe_2 is 1.06 eV while our calculations performed gave practically a zero value, which is smaller than 0.18 eV [20], but is similar to CuInS_2 (the theoretical value of the band gap is also zero) [9], so the scissor operator was used with a value of 1.06 eV to overcome such a discrepancy. The band gaps discussed above are summarized in Table 2, including the experimental, theoretical band gaps and the scissor operator values. It is obvious the band gaps of CuXTe_2 decrease in the $\text{Al} \rightarrow \text{Ga} \rightarrow \text{In}$ series.

Fig. 3 plots the total and atom-projected density of states (DOSs), and it shows that the conduction band between about 2.0 and 6.0 eV are mainly composed of s and p states of Al atoms,

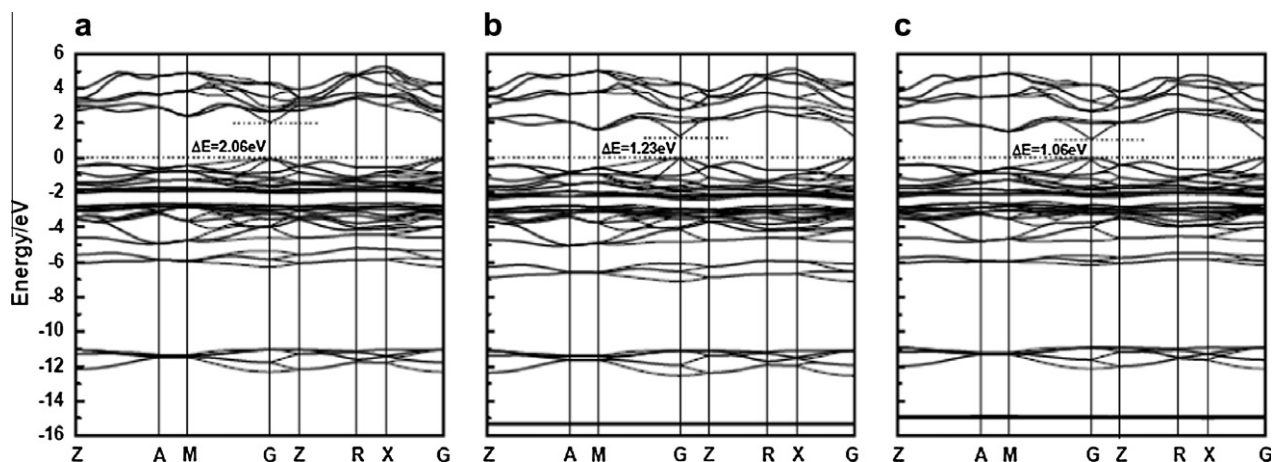


Fig. 2. Band structures of CuAlTe_2 (a), CuGaTe_2 (b) and CuInTe_2 (c) along the high symmetry directions.

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