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Thermodynamic, transport and optical properties of formamide+1,2-ethanediol, 1,3-propanediol and poly (ethylene glycol) 200 binary liquid mixtures

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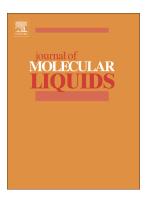
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ACCEPTED MANUSCRIPT

Thermodynamic, transport and optical properties of Formamide + 1,2-Ethanediol, 1,3-

Propanediol and Poly (ethylene glycol) 200 binary liquid mixtures

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Abstract

Densities and viscosities of formamide (FA) + 1,2-ethanediol (1,2-ED), 1,3-propanediol (1,3-

PD) and poly (ethylene glycol) 200 (PEG 200) binary liquid mixtures at temperatures of 293.15,

298.15 and 303.15 K and refractive indices at T = 298.15 K were measured over the entire range

of composition and atmospheric pressure (0.1 MPa). From the experimental measurements,

several thermophysical properties including the excess molar volumes (V_m^E) , coefficient of

thermal expansion α_P , excess coefficient of thermal expansion α_P^E , partial molar volume \bar{V}_{mi} ,

excess refractive index n^E and excess viscosity η^E were calculated. The obtained excess

parameters were used to discuss the inter-intra molecular interactions in the liquid mixtures.

Moreover, the excess properties were correlated with Redlich-Kister polynomial equation and

the viscosities were correlated with Eyring-Margules, McAllister and Andrade models. Finally,

Soave-Redlich-Kwong (SRK) and Peng-Robinson (PR) equation of state were applied to the

densities of studied binary mixtures to examine the ability of these models to predict

thermodynamic properties.

Keywords: Formamide, Diols, Excess properties, Prediction equations

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