



Effect of annealing conditions on formation of SrRuO₃ films by interfacial reaction of SrO/RuO₂ bi-layer films



Ji-Hoon Ahn^{a,*}, Ja-Yong Kim^{b,*}

^a Department of Electronic Material Engineering, Korea Maritime and Ocean University, 727 Taejong-ro, Yeongdo-gu, Busan 49112, Republic of Korea

^b Research & Development Division, SK Hynix Semiconductor Inc., San 136-1, Ami-ri, Bubal-eub, Icheon-si, Kyongki-do 467-701, Republic of Korea

ARTICLE INFO

Article history:

Received 27 May 2015

Received in revised form 1 August 2015

Accepted 22 September 2015

Available online 25 September 2015

Keywords:

Capacitor electrode
Strontium ruthenate
Rapid thermal annealing
Atomic layer deposition

ABSTRACT

In this study, we investigated the effect of annealing conditions on the formation of SrRuO₃ films by the interfacial reaction of SrO/RuO₂ bi-layer films. We found that the annealing temperature and thickness of the SrO layer along with the annealing atmosphere were critical variables in the formation of the conformal SrRuO₃ film. By annealing SrO(20 nm)/RuO₂ bi-layer film at 700 °C in O₂ atmosphere at 1 Torr, the conformal SrRuO₃ film was formed. Finally, we evaluated the potential applicability of the SrRuO₃ film as a functional electrode for perovskite-structured dielectrics, and the dielectric constant of SrTiO₃ film deposited on the SrRuO₃ electrode increased by almost 2.5 times in comparison with that of a film on a Ru electrode.

© 2015 Elsevier B.V. All rights reserved.

1. Introduction

As the size of dynamic random access memory (DRAM) has been scaled down, perovskite-structured dielectric materials, such as SrTiO₃, have received considerable attention as DRAM capacitor dielectrics due to their high dielectric constant [1–4]. However, it is known that the dielectric constant of a SrTiO₃ film decreases abruptly due to inadequate crystallinity when the thickness decreases to a range applicable for a DRAM capacitor [5–6]. SrRuO₃ is a conductive oxide, which has a pseudo-cubic perovskite structure with a lattice constant of 3.93 Å, and the lattice mismatch between SrTiO₃ and SrRuO₃ is roughly 0.5% (whereas the lattice constant of perovskite-SrTiO₃ is 3.91 Å) [7]. Therefore, the SrRuO₃ film can be used as a functional bottom electrode to enhance the crystallinity and dielectric properties of perovskite-structured dielectrics in metal-insulator-metal (MIM) capacitors [8–10]. Furthermore, according to the International Technology Roadmap for Semiconductors, the combination of the perovskite-structured dielectric film and the SrRuO₃ thin film as a bottom electrode is the only potential solution for sub-15-nm-feature DRAM. Therefore, research on the deposition of conformal SrRuO₃ thin film is as important as developing new high-k materials with regards to continuous scaling of DRAM technology.

In this study, we investigated the effect of annealing conditions on the formation of SrRuO₃ films by the interfacial reaction of SrO/RuO₂ bi-layer films, because it is known that Ru or RuO₂, which is a

component of SrRuO₃, is a redox active material and thermally unstable [11–12]. We also evaluated the potential applicability of the SrRuO₃ film as a functional electrode for perovskite-structured dielectrics.

2. Experimental details

SrRuO₃ thin films were prepared by the deposition of a SrO layer on a 30-nm-thick layer of RuO₂ (SrO/RuO₂ bi-layer stack) and post-annealing. RuO₂ film acted as the first layer and was formed by atomic layer deposition (ALD) on a SiO₂ substrate using Ru(EtCp)₂ [Ru(C₂H₅C₂H₄)₂] and oxygen at 230 °C. Ru(EtCp)₂ vapor was carried in argon gas through a bubbler at 50 °C. After RuO₂ deposition, a SrO layer was deposited by plasma-enhanced atomic layer deposition (PEALD) using Sr(DPM)₂ [Sr(C₁₁H₁₉O₂)₂] and oxygen plasma at 225 °C. Sr(DPM)₂ was dissolved in butyl acetate (0.2 M) and supplied to the reaction chamber by a liquid delivery system. After deposition of the SrO/RuO₂ bi-layer stacks, deposited samples were annealed to investigate the effect of the annealing conditions on SrRuO₃ formation.

X-ray diffraction (XRD, Rigaku) analysis using Cu Kα radiation (λ = 1.5405 Å) was used to determine the crystal structures of the annealed SrO/RuO₂ bi-layer samples. The morphology and surface roughness were observed using scanning electron microscopy (SEM, Hitachi S-4800) and atomic force microscopy (AFM, SPA400), respectively. To examine the dielectric properties of SrTiO₃ thin films (detailed deposition conditions were reported in a previous paper [3]), sputtered Pt dots were used as a top electrode, and the dielectric constant was measured using a C–V analyzer (Keithley 590) at a frequency of 1 MHz.

* Corresponding authors.

E-mail addresses: ajh1820@kmou.ac.kr (J.-H. Ahn), jayong.kim@sk.com (J.-Y. Kim).

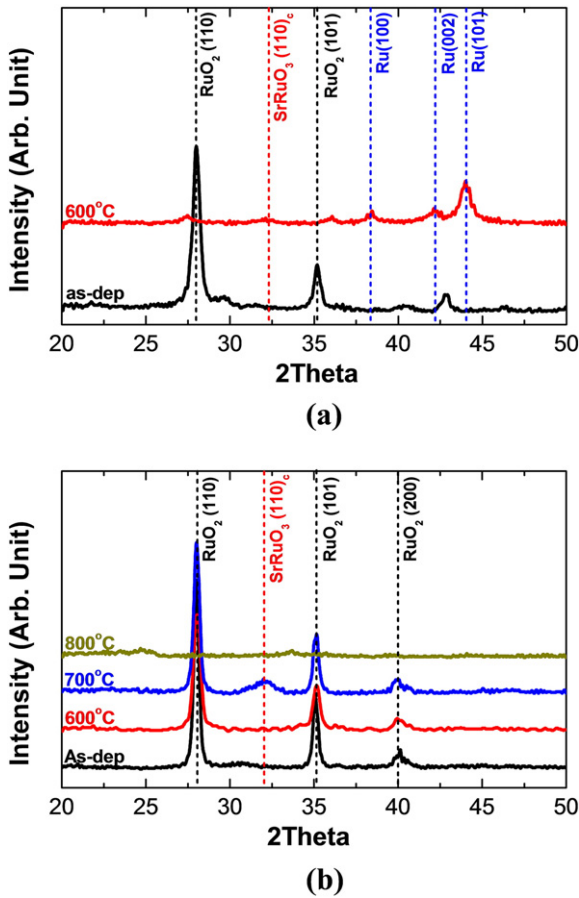


Fig. 1. XRD patterns of SrO(20 nm)/RuO₂(30 nm) bi-layer stack and samples annealed in a temperature range from 600 to 800 °C in (a) 1 Torr N₂ atmosphere and (b) 1 Torr O₂ atmosphere for 10 min.

3. Results and discussion

First, we investigated the effect of annealing atmosphere and temperature on the formation of the SrRuO₃ film from 20-nm-thick SrO and 30-nm-thick RuO₂ bi-layer stacks. In a N₂ atmosphere, as shown in Fig. 1a, it was observed that the RuO₂ layer at the bottom of the bi-layer stack was reduced to metallic Ru. Therefore, it was impossible to obtain the SrRuO₃ film through annealing in the N₂ atmosphere. However, in a O₂ atmosphere, the SrRuO₃(100) peak was observed at an annealing temperature of 700 °C, and there was no serious degradation of the morphology due to SrRuO₃ formation after annealing, as shown in Fig. 2. When the annealing temperature increased to 800 °C, all the peaks of RuO₂ and SrRuO₃ disappeared, and it was confirmed that the deposited SrO/RuO₂ bi-layer was removed, as shown in

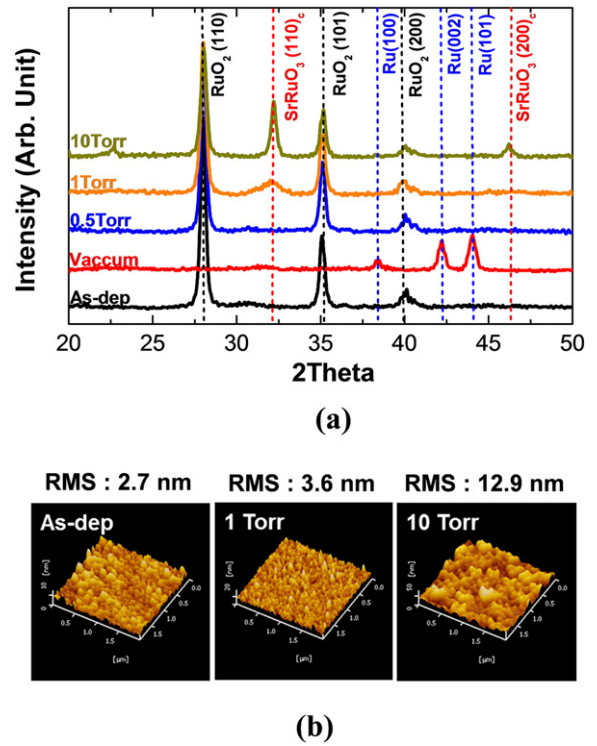


Fig. 3. (a) Annealing pressure dependence of XRD patterns and (b) the RMS roughness change of SrO(20 nm)/RuO₂(30 nm) bi-layer stack. All annealing processes were conducted in an O₂ atmosphere at 700 °C for 10 min.

Fig. 2c. These results might be due to complete etching of the RuO₂ bottom layer after annealing under those conditions, because RuO₂ films can decompose to RuO₄, a volatile oxide, at such a high temperature [12].

Fig. 3a shows the annealing pressure dependence of the XRD patterns when SrO(20 nm)/RuO₂(30 nm) bi-layer stacks were annealed in the O₂ atmosphere at 700 °C for 10 min. In vacuum annealing, the RuO₂ bottom layer changed to Ru, similar to the case of annealing in the N₂ atmosphere, and the intensity of the SrRuO₃ peak increased with annealing pressure. However, as shown in Fig. 3b, the root-mean-square (RMS) roughness measured by AFM also increased with annealing pressure. In other words, although the crystallinity of the SrRuO₃ film improved by increasing the annealing temperature to 10 Torr, the surface morphology was roughened, which can degrade the electrical properties of dielectric materials. Therefore, it is thought that the optimum annealing condition for the formation of the SrRuO₃ film is 700 °C in a 1 Torr O₂ atmosphere.

Next, we investigated the effect of the thickness of the SrO upper layer on the formation of SrRuO₃ film. Fig. 4a shows the XRD patterns of annealed SrO/RuO₂(30 nm) bi-layer stack films with variation of

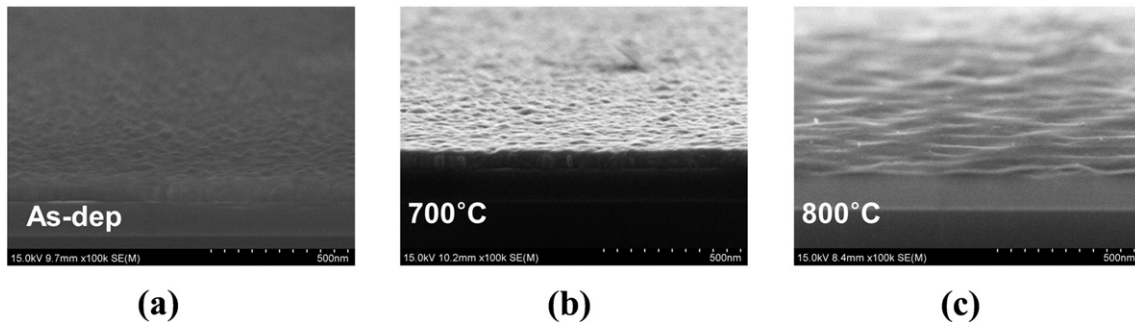


Fig. 2. Tilted SEM images of SrO(20 nm)/RuO₂(30 nm) at various annealing temperatures in a O₂ atmosphere.

Download English Version:

<https://daneshyari.com/en/article/541189>

Download Persian Version:

<https://daneshyari.com/article/541189>

[Daneshyari.com](https://daneshyari.com)