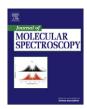


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First high-resolution analysis of the v_{15} , v_{12} , v_{5} , v_{10} and v_{2} bands of oxirane

J.-M. Flaud ^a, W.J. Lafferty ^{b,*}, F. Kwabia Tchana ^a, A. Perrin ^a, X. Landsheere ^a

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ABSTRACT

Fourier transform spectra of oxirane (ethylene oxide, $c-c_2H_4O$) have been recorded in the 730–1560 cm⁻¹ (6.4–13.7 µm) spectral region using a Bruker IFS125HR spectrometer at a resolution of 0.0019 cm⁻¹. A total of six vibration bands, v_{15} , v_{12} , v_5 , v_3 , v_{10} and v_2 , have been observed and analyzed. The corresponding upper state ro-vibrational levels were fit using Hamiltonian matrices accounting for various interactions. Satisfactory fits were obtained using the following polyads $\{15^1,12^1,5^1\}$ and $\{10^1,2^1\}$ of interacting states. As a result, an accurate and extended set of Hamiltonian constants were obtained. The following band centers were derived: v_0 (v_{15}) = 808.13518(60) cm⁻¹, v_0 (v_{12}) = 822.27955(37) cm⁻¹, v_0 (v_5) = 876.72592(15), v_0 (v_3) = 1270.37032(10) cm⁻¹, v_0 (v_{10}) = 1471.35580(50) cm⁻¹ and v_0 (v_2) = 1497.83309(15) cm⁻¹ where the uncertainties are one standard deviation.

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1. Introduction

Oxirane(ethylene oxide, $c-C_2H_4O$) has been the subject of a number of both low- and medium-resolution infrared studies [1–3] but, to the best of our knowledge, only one band, the v_3 band located at 1270.4 cm⁻¹, has been analyzed at high-resolution [4]. However, there have been quite a few microwave studies on both the ground state and excited states of this molecule [5–9]. Because of this a very accurate set of rotational and centrifugal distortion constants have been derived for the ground state leading to the possibility of using combination differences to check the assignments of the infrared bands, a majority of which are perturbed. Despite the number of infrared studies there is still some uncertainty of the correct vibrational assignment of a few of the IR bands.

In this paper we present an extensive high-resolution analysis of the three interacting bands v_{15} , v_{12} and v_5 which fall in the 750 cm⁻¹ to 950 cm⁻¹ range, of the v_3 band between 1200 cm⁻¹ and 1350 cm⁻¹ and of the interacting v_2 and v_{10} bands which absorb between 1420 cm⁻¹ and 1560 cm⁻¹.

The upper state rotational levels were fit using Hamiltonian (written in the I^r (x = b, z = a, y = c) representation) models taking into account the numerous interactions affecting the various states. In this way, very satisfactory fits were obtained which lead to accurate Hamiltonian constants. The oxirane molecule possesses C_{2v} symmetry. In order to simplify the following discussions, its

structure as well as its symmetry properties and nuclear spin statistical weights are illustrated in Fig. 1.

2. Experimental details

The high-resolution absorption spectra of oxirane were recorded with the Bruker IFS125 SHR Fourier transform spectrometer¹ located at the LISA facility in Creteil. A KBr beam splitter, Globar source (silicon carbide, SiC) and a HgCdTe (MCT) photovoltaic liquid nitrogen cooled detector were used for these recordings. The spectrometer was evacuated to roughly 6.7 Pa (0.05 Torr) in order to minimize residual absorptions by H₂O and CO₂. An entrance aperture diameter of 1.3 mm and a maximum optical path difference (d_{MOPD}) of 473.68 cm was used. According to the Bruker definition, this corresponds to a resolution of 0.0019 cm $^{-1}$ (Resolution = 0.9/d_{MOPD}). The White-type multipass absorption cell, constructed of Pyrex glass and equipped with CsBr windows, was connected to the FTS with a dedicated optical interface (six mirrors) inside the sample chamber of the instrument. Its base length is 0.80 m and, for the experiment described here, an optical path of 32.049 m was used. The oxirane sample of stated purity 99.8% was purchased from Sigma-Aldrich and used without further purification.

Six spectra were recorded at a stabilized room temperature of 295.2 K and with pressures of 8.399(13) Pa (0.0630(1) Torr),

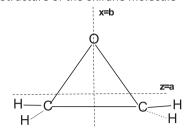
^a Laboratoire Interuniversitaire des Systèmes Atmosphériques (LISA), UMR CNRS 7583, Universités Paris Est Créteil et Paris Diderot, Institut Pierre Simon Laplace, 61 Avenue du Général de Gaulle, 94010 Créteil Cedex, France

^b Optical Technology Division, National Institute of Standards and Technology, Gaithersburg, MD 20899-8440, USA

^{*} Corresponding author. Fax: +1 301 975 2950. E-mail address: walter.lafferty@nist.gov (W.J. Lafferty).

¹ Certain commercial equipment, instruments or materials are identified in this paper to adequately specify the experimental procedure. Such identification does not imply recommendation or endorsement by the National Institute of Standards and Technology nor does it imply that the materials or equipment identified are necessarily the best available for the purpose.

Structure of the oxirane molecule



C_{2v} group character table

		-	C _{2x}	σ_{xz}	σ_{xy}	Polar	Axial	Vibrations
						vector	vector	
	A_1	1	1	1	1	T _x		v_1, v_2, v_3, v_4, v_5
	A ₂	1	1	-1	-1		R_x	v_6, v_7, v_8
	B ₁	1	-1	1	-1	T _z	R _v	$v_9, v_{10}, v_{11}, v_{12}$
	B ₂	1	-1	-1	1	T _v	R,	v_{13}, v_{14}, v_{15}

Nuclear spin statistical weights for the ground state rotational levels

K_a	K _c	Weight		
Even	Even	10		
Even	Odd	6		
Odd	Even	6		
Odd	Odd	10		

Fig. 1. Structure and symmetry properties of the oxirane (ethylene oxide) molecule.

 $33.86(40)\ Pa\ (0.2540(3)\ Torr),\ 16.40(13)\ Pa\ (0.1230(1)\ Torr)\ ,\ 8.626(13)\ Pa\ (0.0647(1)\ Torr),\ 20.465(25)\ (0.1535(2)\ Torr),\ and\ 62.128(80)\ (0.4660(6)\ Torr)\ in\ order\ to\ get\ the\ best\ line\ positions\ for\ each\ band.\ Sample\ pressures\ in\ the\ multipass\ cell\ were\ measured\ using\ a\ high-accuracy\ capacitance\ manometer\ (266.65\ Pa\ (2\ Torr)\ full\ scale,\ MKS\ instruments\ type\ 627D\ Baratron)\ which\ has\ a\ stated\ uncertainty\ of\ 0.12\%\ of\ full\ scale\ according\ to\ the\ manufacturer.\ Spectra\ were\ ratioed\ against\ an\ empty\ cell\ single-channel\ background\ spectrum\ which\ was\ taken\ at\ a\ resolution\ of\ 0.4864\ cm^{-1},\ in\ order\ to\ ensure\ the\ best\ possible\ signal-to-noise\ in\ the\ ratioed\ spectra.\ The\ spectra\ were\ the\ result\ of\ the\ co-addition\ of\ 300\ inter-$

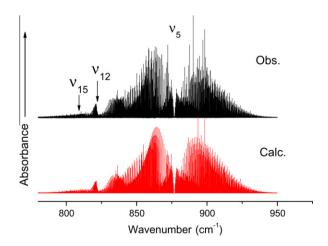


Fig. 2. Overview of the oxirane spectrum between 1.05 μ m and 1.33 μ m. The C-type ν_{15} , B-type ν_5 and A-type ν_{12} band centers are indicated. The upper trace is the observed spectrum recorded with a resolution of 0.0019 cm⁻¹. The lower trace is a synthetic spectrum calculated using the line list generated in this work. The observed and synthetic spectra are plotted in absorbance and are shifted vertically for clarity.

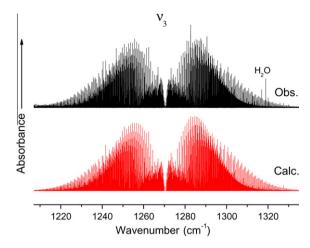


Fig. 3. Overview of the oxirane spectrum between 8.2 and 7.5 μ m. The band center of the B-type ν_3 band is shown.

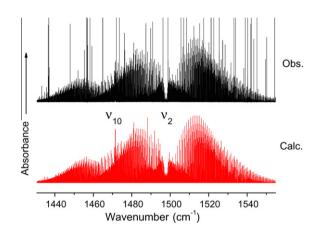


Fig. 4. Overview of the oxirane spectrum between 6.61 μ m and 7.04 μ m. The band centers of the A-type ν_{10} and B-type ν_{2} bands are given. Note the presence of strong residual water lines in the observed spectrum. Most of them have been used for calibration.

Table 1Range of quantum numbers observed for experimental energy levels and a statistical analysis of the results of the energy level calculation for the $\{15^1, 12^1, 5^1\}$, 3^1 and $\{10^1, 2^1\}$ ro-vibrational levels of oxirane.

Vibrational state	{15 ¹	12 ¹	5 ¹ }	3 ¹	$\{10^{1}$	2 ¹ }
Number of levels	278	863	1199	1219	832	1087
J_{Max}	33	49	53	46	49	54
K_{Max}	29	35	40	28	31	37
$0.000 \leqslant \delta < 0.0006$ (%)	64.4	84.5	79.7	92.4	91.6	94.9
$0.0006 \leqslant \delta < 0.0012$ (%)	25.5	11.0	15.0	5.5	6.7	4.3
$0.0012 \leqslant \delta < 0.0032$ (%)	10.1	4.5	5.3	2.1	1.7	0.8
Std. deviation $(10^{-3} \text{ cm}^{-1})$		0.75		0.59		0.33

 $\delta = |E_{\text{Obs.}} - E_{\text{Calc}}| \text{in cm}^{-1}.$

ferograms. For the Fourier transform a Mertz-phase correction, 1 cm $^{-1}$ phase resolution, zero-filling factor of 2 and boxcar apodization function were applied to the averaged interferograms. All the spectra were calibrated with residual CO₂ and H₂O lines observed in the spectra with their wavenumbers taken from HITRAN89 [10]. The resulting precision is $\pm 0.00012~\rm cm^{-1}$ (RMS) for well isolated lines. The absolute accuracy can be estimated as the RMS agreement of the CO₂ and H₂O lines, i.e. $0.0007~\rm cm^{-1}$. Figs. 2–4 give overviews of the 750 cm $^{-1}$ –950 cm $^{-1}$, 1200 cm $^{-1}$ –1350 cm $^{-1}$ and 1420 cm $^{-1}$ –1560 cm $^{-1}$ spectral domains showing the high signal-to-noise ratios which were obtained. The band centers of the various bands are

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