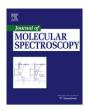
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# Photodissociation of ND<sub>3</sub> and ND<sub>2</sub>H at 193.3 nm: Symmetry dependence of the rotational distributions and vibrational excitation of the ND<sub>2</sub> ( $\widetilde{A}^2A_1$ ) fragment

Geoffrey Duxbury a,\*, Jonathan P. Reid b

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#### ABSTRACT

A rotational and vibrational analysis has been made of the ND<sub>2</sub>  $\widetilde{A}^2A_1 - \widetilde{X}^2B_1$  emission spectrum produced from the ultraviolet laser induced dissociation of both jet cooled and room temperature deuterated ammonia, ND<sub>3</sub>, and di-deuterated ammonia, ND<sub>2</sub>H. The pattern of the strong features in the emission spectra is very different in the fragmentation of ND<sub>3</sub> and ND<sub>2</sub>H, with a much wider range of angular momentum states being observed from the photolysis of the predissociative  $(\widetilde{A})$  state of unsymmetrical parent ND<sub>2</sub>H  $(\widetilde{A})$ . The analysis is based upon the earlier studies of the electronic spectrum of ND<sub>2</sub>, and model calculation based upon the stretch-bender Renner–Teller Hamiltonian. The spectra consist of two types, transitions from a narrow distribution of high angular momentum states in the photolysis of ND<sub>3</sub> and ND<sub>2</sub>H, and in the photolysis of ND<sub>2</sub>H strong emission from threshold states in three high energy regions. The threshold states are in the third bending level,  $v_2' = 3$  of the ND<sub>2</sub>  $(\widetilde{A})$  state, and have no angular momentum about the axis of least moment of inertia,  $K_a' = 0$ .

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#### 1. Introduction

The electronic spectrum of the  $\mathrm{NH}_2$  free radical was one of the first examples of the use of high resolution spectroscopy to study a transient polyatomic molecule. This absorption spectrum was recorded, following the flash photolysis of ammonia,  $\mathrm{NH}_3$ , by Ramsay [1]. Following a further note by Dressler and Ramsay [2] on the role that  $\mathrm{NH}_2$  played as an example of the Renner effect [3], they published a detailed analysis of the electronic absorption spectra of the  $\widetilde{A}^2A_1 - \widetilde{X}^2B_1$  transition in  $\mathrm{NH}_2$  and  $\mathrm{ND}_2$  [4]. In this paper they made a comparison of the experimental energy level pattern with that calculated by Pople and Longuet-Higgins [5], which showed clearly the role played by Renner, or now more commonly Renner–Teller, coupling.

Since this seminal paper [4] in 1959, with the exception of one paper by Muenchausen et al. in 1985 [6] on electronic and vibrational spectra of ND<sub>2</sub>, and a detailed study by Morino and Kawaguchi [7] of the high resolution far infrared spectra of NH<sub>2</sub>, NHD and ND<sub>2</sub>, little work was done on the spectroscopy of ND<sub>2</sub>, particularly electronic spectroscopy, although the electronic spectrum of NH<sub>2</sub> has been studied extensively [8–12]. The Experiments carried out on ND<sub>2</sub> in the period since 1988 have been primarily concerned with studying the photodissociation dynamics of ammonia and its isotopologues. The first measurements [13,14] were carried

out by Mordaunt et al. by H(D) Rydberg atom photofragment translational spectroscopy. A detailed theoretical model of the photodissociation dynamics was given by Dixon [15]. These studies were followed by a series of papers [16–20] on the vibrational dynamics of the photofragmentation of NH<sub>3</sub>, NH<sub>2</sub>D, ND<sub>2</sub>H and ND<sub>3</sub>, by Leone and his colleagues, who studied the photofragmentation in detail using Fourier transform spectroscopy of the emission from the triatomic fragment molecules.

More recently a detailed analysis has been made of the fluorescence spectra of ND2 obtained by the photolysis of ND3 [21] using an excimer laser operating at 193.3 nm. In the present paper this analysis has been extended to the emission spectra of ND2 obtained by the photolysis of ND<sub>2</sub>H. In the comparison of the emission spectra of ND<sub>2</sub> produced by the photolysis of ND<sub>3</sub> and ND<sub>2</sub>H Mordaunt et al. [13,14] and Reid et al. [17,18,20] have shown that the differing fragmentation energies of the NH bond in ND<sub>2</sub>H and the ND bond of ND<sub>3</sub> need to be taken into account. The available energy from the excimer laser corresponds to ca. 51 733 cm<sup>-1</sup>. The bond dissociation energy of the NH bond of ND<sub>2</sub>H given by Mordaunt et al. [12,13] corresponds to 37 300 cm<sup>-1</sup>, whereas that of the ND bond in ND<sub>3</sub> is equivalent to 38 010 cm<sup>-1</sup>. Since the lowest ro-vibronic level in the A state of ND<sub>2</sub> lies at 11 120 cm<sup>-1</sup> the excess energy for populating excited rovibronic levels of the A state of ND<sub>2</sub> is 3313 cm<sup>-1</sup> from the fragmentation of ND<sub>2</sub>H, but only 2603 cm<sup>-1</sup> from ND<sub>3</sub> fragmentation. As a result the highest wavenumber spin doublet observed from the emission of ND<sub>2</sub> from the jet cooled ND<sub>3</sub> parent is a very weak doublet at 13 391 cm<sup>-1</sup> due to  $2_0^{1} P_{7,7}$ .

<sup>&</sup>lt;sup>a</sup> Department of Physics, University of Strathclyde, Glasgow G4 ONG, Scotland, United Kingdom

<sup>&</sup>lt;sup>b</sup> School of Chemistry, University of Bristol, Bristol BS8 1TS, United Kingdom

<sup>\*</sup> Corresponding author. Fax: +44 141 552 2891. E-mail address: g.Duxbury@strath.ac.uk (G. Duxbury).

One of the complications of the analysis of the emission spectrum of  $ND_2$  is that none of the energy levels in the  $\widetilde{A}^2A_1$  state of  $ND_2$ , from which emission has been recorded, correspond to those observed by Dressler and Ramsay [4] in their absorption spectra. As a result of this no direct comparisons of the absorption and emission spectra may be made. Furthermore, many of the observed fluorescence spectra arise from states with high angular momentum about the molecular a-rotational axis, whereas the spectra recorded by Dressler and Ramsay [4] accessed only vibronic levels up to K' = 3.

It is the purpose of the present paper to explore the effects that the photolysis of the unsymmetrical ND<sub>2</sub>H exerts on the form of the observed emission spectrum, in particular to the reduction of the regularity of the pattern of the emission spectra of ND<sub>2</sub> from that observed from the photolysis of the symmetrical isotopologue ND<sub>3</sub>.

The emission spectra recorded following the photolysis of  $ND_3$  showed considerable repeating regularity in their structure [21], whereas those recorded following the photolysis of the unsymmetrical isotopologue,  $ND_2H$ , exhibit far less regularity in their repeti-

tive structure. One possible hypothesis is that as the trigonal symmetry of the parent molecule has been broken, efficient transfer of energy to motion about the b/c rotational axes of the ND<sub>2</sub> is much more likely, and may account for some of the complexity of the emission spectrum.

#### 2. Experimental

A detailed description of the experimental apparatus has been given in previous publications on the dissociation dynamics of ammonia and deuterated ammonias [16–20], so only a very brief summary will be provided here. The studies were performed both at room temperature, with a continuous flow of ND<sub>3</sub>, or a mixture containing NH<sub>3</sub>, NH<sub>2</sub>D, ND<sub>2</sub>H and ND<sub>3</sub>, and jet cooled, with a piezoelectric pulsed valve. The triggering of the ArF eximer laser, which initiates the photodissociation, and also the opening of the pulsed jet for the jet cooled studies, were both synchronised to the position of a moving mirror in a commercial continuous scan

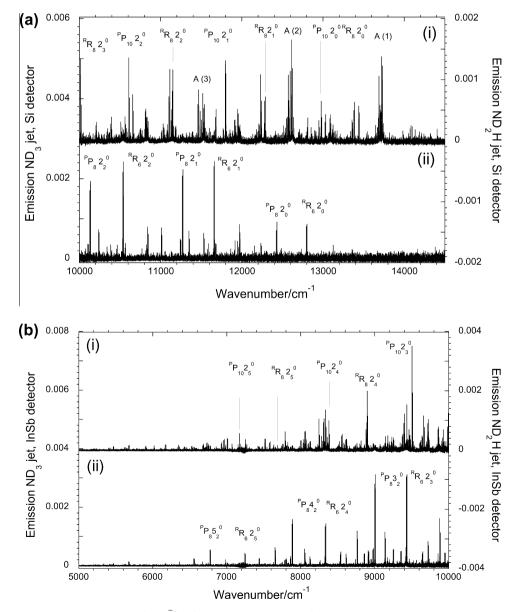


Fig. 1. Survey of the entire product emission spectrum of  $ND_2(\overline{A}^2A_1)$  from the photodissociation of: (i) jet cooled  $ND_2H$  and (ii)  $ND_3$ . (a) 10 000 to 14 450 cm<sup>-1</sup>, Si avalanche detector; (b) 5000–10 000 cm<sup>-1</sup>, InSb detector. Note the much greater density of the spectral lines in the spectrum of jet cooled  $ND_2H$ , and their high intensity at the onset of fluorescence. Features A(1)–A(3) comprise several transitions, including those from the  $K'_a \equiv K' = 0$  levels of  $\nu'_2 = 3$  in the  $ND_2(\overline{A}^2A_1)$  state.

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