

Note

The [1 + 1] two-photon dissociation spectra of CO_2^+ ($X^2\Pi_g(\Omega = 3/2)$) via $A^2\Pi_{u,3/2}$ (v_1v_20) $\leftarrow X^2\Pi_{g,3/2}(000)$ transitions

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ABSTRACT

The mass-resolved [1 + 1] two-photon dissociation spectra of CO_2^+ ($X^2\Pi_g(\Omega = 3/2)$) via $A^2\Pi_{u,3/2}$ (v_1v_20) $\leftarrow X^2\Pi_{g,3/2}(000)$ transitions were studied by introducing a dissociation laser with a wavelength of 283–353 nm. CO_2^+ ($X^2\Pi_{g,3/2}(000)$) was prepared by the [3 + 1] multiphoton ionization of CO_2 at 333.69 nm. The vibronic bands of (v_120 ; $v_1 = 0-5$) $\mu^2\Pi_{3/2}$ and (v_120 ; $v_1 = 0-5$) $\kappa^2\Pi_{3/2}$ involving the bending mode of CO_2^+ ($A^2\Pi_{u,3/2}$) were assigned. Based on the assignments, the spectral constants of $T_e = 27969.3 \pm 1.2 \text{ cm}^{-1}$ [above CO_2^+ ($X^2\Pi_{g,3/2}$)], $\nu_1 = 1125.89 \pm 0.53 \text{ cm}^{-1}$, $\chi_{11} = -0.659 \pm 0.010 \text{ cm}^{-1}$, $\nu_2(\mu^2\Pi_{3/2}) = 429.5 \pm 9.7 \text{ cm}^{-1}$, and $\nu_2(\kappa^2\Pi_{3/2}) = 528.7 \pm 8.0 \text{ cm}^{-1}$ for CO_2^+ ($A^2\Pi_{u,3/2}$) were deduced. The photodissociation dynamics of CO_2^+ via $A^2\Pi_{u,3/2}(v_1v_20) \leftarrow X^2\Pi_{g,3/2}(000)$ transitions are discussed.

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The linear triatomic CO_2^+ ion is an important component in planetary atmospheres [1], and its spectroscopy and dynamic behavior have attracted the attention of experimenters and theoreticians [2]. The vibronic structures of CO_2^+ ($X^2\Pi_g, A^2\Pi_u, B^2\Sigma_u^+, C^2\Sigma_g^+$) have been investigated previously using various experimental techniques, including emission [3–5], laser absorption [6,7], HeI photoelectron [8–15], threshold photoelectron (TPE) [16,17], photoionization efficiency [18–20], photoelectron–photon coincidence [21], photoion–photon coincidence [22], photoelectron–photoion coincidence [23], pulsed field ionization (PFI)–photoelectron (PFI-PE) [24–28], PFI-PE-photoion (PFI-PE-PI) coincidence [29] measurements, and resonance enhanced multiphoton ionization [30,31]. However, there are few reports on the vibronic level-selected excitation and dissociation spectroscopy for CO_2^+ , which could provide more detailed information for the photochemistry of CO_2^+ in a direct way compared to the photoelectron spectra of neutral molecule. In our previous study, we obtained the [1 + 1] two-photon dissociation spectra of CO_2^+ ($X^2\Pi_g(\Omega = 1/2)$) via $A^2\Pi_{u,1/2}(v_1v_20) \leftarrow X^2\Pi_{g,1/2}(000)$ transitions [32]. Because the (000) band intensity of CO_2^+ ($X^2\Pi_g(\Omega = 3/2)$) is much weaker than that of CO_2^+ ($X^2\Pi_g(\Omega = 1/2)$), a higher signal to noise ratio is needed to study CO_2^+ ($X^2\Pi_g(\Omega = 3/2)$) ions. In this study, we present a similar study on the [1 + 1] two-photon dissociation spectra of CO_2^+ ($X^2\Pi_g(\Omega = 3/2)$) with a pulsed tunable laser with a wavelength range of 283–353 nm. By combining data from this study with the data from our previous study, more integrated data on the $A^2\Pi_u$ state of CO_2^+ was obtained.

The experimental setup has been previously described [32–34]. Jet-cooled CO_2 molecules were produced by supersonic expansion

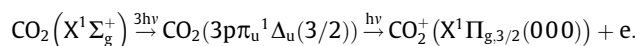
of pure CO_2 gas at about 3 atm through a pulsed nozzle (General Valve) with a nozzle orifice diameter of 0.5 mm and were introduced into a photoionization chamber. The ionization laser at 333.69 nm was perpendicularly focused on the molecular beam of CO_2 by a quartz lens with $f = 135 \text{ mm}$ to produce a CO_2^+ ion via [3 + 1] resonance-enhanced multiphoton ionization (REMPI) of CO_2 molecules [30,31]. The dissociation laser ($\sim 0.2 \text{ mJ/pulse}$) at 283–353 nm counterpropagating with the ionization laser was focused by a quartz lens with $f = 320 \text{ mm}$ to excite CO_2^+ ($X^2\Pi_{g,3/2}$). Both lasers were temporally and spatially matched with each other at the laser-molecular interaction point. The wavelength of the lasers was calibrated using a wavemeter.

The ions were extracted and accelerated into a TOF (time of flight) mass spectrometer and drifted along a 70 cm long TOF tube. The ions were detected by a microchannel plate (MCP) detector, and the signals from the MCP output were amplified by a preamplifier (Stanford model SR240A). The mass-resolved data were collected by averaging the amplified signals for the selected mass species with a transient recorder and stored in a personal computer. The intensities of the ionization laser and the dissociation lasers were monitored simultaneously during the experiment.

As revealed by Wu and co-workers [35], in the resonance-enhanced multiphoton ionization photoelectron spectrum (REMPI-PES) of CO_2 via $3p\pi_u^1\Delta_u(\Omega = 3/2)$ spin-orbit state, the (000) level of CO_2^+ ($X^2\Pi_{g,3/2}$) is much stronger than those related to the higher vibrational levels of CO_2^+ ($X^2\Pi_{g,3/2}$) (see Fig. 1b in Ref. [35]). In this study, the CO_2^+ ions were prepared in $X^2\Pi_{g,3/2}(000)$ states with minimum amounts of CO^+ , O^+ , and C^+ ions, with a lens with $f = 135 \text{ mm}$ to focus the ionization laser ($\sim 5 \text{ mJ/pulse}$) at 333.69 nm. The involved [3 + 1] REMPI process of CO_2 to prepare CO_2^+ ($X^2\Pi_{g,3/2}(000)$) ions has been previously described [30,31].

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The mass-resolved photodissociation spectra (the depletion spectrum of parent ion CO_2^+ and the enhanced spectrum of fragment ions CO^+ , O^+ , and C^+) in Fig. 1 were obtained by scanning the dissociation laser in the range of 283–353 nm. Based on the spectroscopy of CO_2^+ previously reported [3,4,15,27,28,36], the photodissociation spectra could be completely assigned as the electronic transition of $\text{CO}_2^+(\text{A}^2\Pi_{u,3/2}(v_1v_20)) \leftarrow \text{CO}_2^+(\text{X}^2\Pi_{g,3/2}(000))$, where v_1 and v_2 represent vibrational quantum numbers for the v_1 (symmetric stretching) and v_2 (bending) modes, respectively. The possible assignments of the photodissociation spectra shown in Fig. 1 are summarized in Table 1. Note that the photodissociation spectra can be complicated by vibration–electronic interaction (Renner–Teller effect) and Fermi resonance interaction related to $\text{CO}_2^+(\text{A}^2\Pi_{u,3/2}(v_1v_20))$. The vibration–electronic interaction induces the Renner–Teller splitting of a $^2\Pi$ state (lower μ and upper κ components of $\text{A}^2\Pi_{u,3/2}$ in Table 1) whenever the bending vibration v_2 is excited [37]. Because the vibrational frequencies of $\text{CO}_2^+(\text{A}^2\Pi_{u,3/2})$ have the approximate relationship $v_1 \sim 2v_2$, a group of vibrational levels involving the symmetric stretch v_1 and overtones of the bending vibration v_2 , such as the $(v_1,0,0)$ and $(v_1-1,2,0)$ levels of $\text{CO}_2^+(\text{A}^2\Pi_{u,3/2})$ can be coupled through strong Fermi resonances. Because of the strong Fermi resonance interaction, the unfavorable Franck–Condon transitions $\text{A}^2\Pi_{u,3/2}(v_1-1,2,0) \leftarrow \text{X}^2\Pi_{g,3/2}(000)$ can occur with comparable intensities due to strongly allowed transitions $\text{A}^2\Pi_{u,3/2}(v_100) \leftarrow \text{X}^2\Pi_{g,3/2}(000)$, which is indicated in Fig. 1.

As shown in Table 1, the v_{expt} values of $\text{A}(v_100; v_1 = 0 - 6)^2\Pi_{3/2}$, $\text{A}(020; 520) \mu^2\Pi_{3/2}$, and $\text{A}(020; 320; 420; 520) \kappa^2\Pi_{3/2}$ are in good agreement with the known v_{PES} values [28]. Based on the data for the $\text{A}^2\Pi_{u,1/2} \leftarrow \text{X}^2\Pi_{g,1/2}$ transition in our previous study and the $\text{A}^2\Pi_{u,3/2} \leftarrow \text{X}^2\Pi_{g,3/2}$ transition presented in this work, it is now possible to identify several unresolved $\text{A}(v_1v_20) \mu^2\Pi_{3/2,1/2}$ and $\text{A}(v_1v_20) \kappa^2\Pi_{3/2,1/2}$ bands [28]. For example, we could certainly give the resolved v_{expt} values of $\text{A}(120, 220, 320, 420) \mu^2\Pi_{3/2}$ and $\text{A}(120, 220, 320, 420) \mu^2\Pi_{3/2,1/2}$ and $\text{A}(120, 220) \kappa^2\Pi_{3/2,1/2}$ bands in Ref. [28].

The spectral constants of $T_e = 27969.3 \pm 1.2 \text{ cm}^{-1}$ [above $\text{CO}_2^+(\text{X}^2\Pi_{g,3/2})$], $v_1 = 1125.89 \pm 0.53 \text{ cm}^{-1}$, $\chi_{11} = -0.659 \pm 0.010 \text{ cm}^{-1}$, $v_2(\mu^2\Pi_{3/2}) = 429.5 \pm 9.7 \text{ cm}^{-1}$, and $v_2(\kappa^2\Pi_{3/2}) = 528.7 \pm 8.0 \text{ cm}^{-1}$ for $\text{CO}_2^+(\text{A}^2\Pi_{u,3/2})$ were deduced from the $\text{A}^2\Pi_{u,3/2}(v_1v_20) \leftarrow \text{X}^2\Pi_{g,3/2}(000)$ transitions of CO_2^+ . The deduced vibrational frequencies for

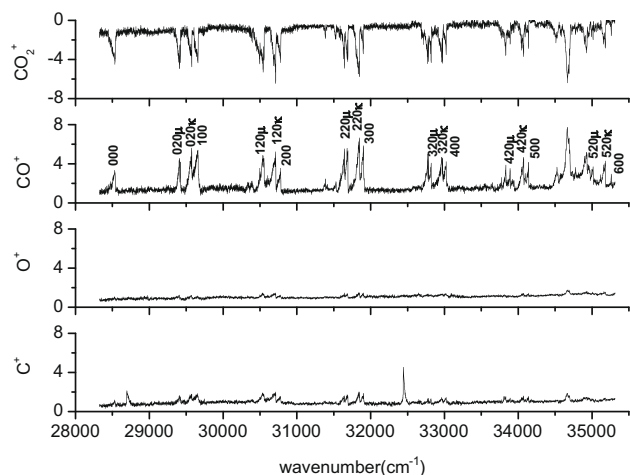


Fig. 1. The mass-resolved [1 + 1] photodissociation spectra of CO_2^+ obtained in the wavelength range of 283–353 nm. The spectra are assigned to the $\text{A}^2\Pi_{u,3/2}(v_1v_20) \leftarrow \text{X}^2\Pi_{g,3/2}(000)$ transitions of CO_2^+ .

$\text{CO}_2^+(\text{A}^2\Pi_{u,3/2})$ are in accordance with those of $v_1 = 1127 \text{ cm}^{-1}$, and $v_2 = 461 \text{ cm}^{-1}$ given by emission spectra [3,4] and PFI-PE [28] spectra.

In addition, the spectral band of $\text{B}^2\Sigma_u^+(000) \leftarrow \text{X}^2\Pi_{g,3/2}(000)$ was also observed due to the overlap between $\text{A}^2\Pi_{u,3/2}$ and $\text{B}^2\Sigma_u^+$.

Considering that the one-photon excitation energy (3.51–4.38 eV) related to the resonance peaks in Fig. 1 cannot access the dissociation limit of CO_2^+ for the formation of O^+ (5.2738 eV), CO^+ (5.6724 eV) [29], and C^+ (6.98 eV) [38] from its electronic ground state, two photon excitation energy (7.02–8.76 eV) is needed to dissociate CO_2^+ . This means that the dissociation process of CO_2^+ via $\text{A}^2\Pi_{u,3/2}(v_1v_20) \leftarrow \text{X}^2\Pi_{g,3/2}(000)$ transitions to produce CO^+ , O^+ , and C^+ is a [1 + 1] two-photon process, that is, the intermediate state in this photodissociation process is the $\text{A}^2\Pi_{u,3/2}$ state of CO_2^+ .

Fig. 1 shows that CO^+ is the main ionic product in the [1 + 1] photodissociation process of CO_2^+ compared to O^+ and C^+ . Moreover, the C^+ yield is higher than the O^+ yield, even though the dissociation limit to form O^+ is the lowest. The percentage branching ratios of $[\text{CO}^+]/([\text{CO}^+] + [\text{O}^+] + [\text{C}^+])$, $[\text{C}^+]/([\text{CO}^+] + [\text{O}^+] + [\text{C}^+])$, and $[\text{O}^+]/([\text{CO}^+] + [\text{O}^+] + [\text{C}^+])$ measured in this work in the two photon energy range of $57000\text{--}70600 \text{ cm}^{-1}$ are shown in Fig. 2. The $[\text{CO}^+]/([\text{CO}^+] + [\text{O}^+] + [\text{C}^+])$ percentage branching ratio of 69–79% is much larger than the percentage branching ratios of $[\text{C}^+]/([\text{CO}^+] + [\text{O}^+] + [\text{C}^+])$ and $[\text{O}^+]/([\text{CO}^+] + [\text{O}^+] + [\text{C}^+])$, which were 14–20% and 6–11%, respectively. Similar photofragment branching ratios were also observed for the $(v_1 + 1, 0, 0)^2\Pi_{3/2}$, $(v_1, 2, 0)\mu^2\Pi_{3/2}$, and $(v_1, 2, 0)\kappa^2\Pi_{3/2}$ bands.

Since CO^+ is the main ionic product compared to O^+ for the ionization energy of CO_2 exceeding 19.5000 eV [5.7235 eV above $\text{CO}_2^+(\text{X}^2\Pi_{g,3/2}(000))$] [29] in the PFI-PEPICO experiment, it is reasonable to assume that the dissociation dynamics for the vibrationally excited levels of $\text{CO}_2^+(\text{C}^2\Sigma_g^+)$ [39,29] are also applicable in the internal energy range of 7.02–8.76 eV above $\text{CO}_2^+(\text{X}^2\Pi_{g,3/2}(000))$. In terms of energy, a two photon energy of 7.02–8.76 eV can reach the high vibrational levels in the $(\text{C}^2\Sigma_g^+)^{\dagger}$ state via the allowed transitions of $(\text{C}^2\Sigma_g^+)^{\dagger} \leftarrow \text{A}^2\Pi_{u,1/2} \leftarrow \text{X}^2\Pi_{g,1/2}$, where the superscript “ \dagger ” denotes high vibrational levels in the corresponding electronic states. In addition, both the repulsive $\text{CO}_2^+(b^4\Pi_u/4B_1)$ state connected with $\text{CO}^+(\text{X}^2\Sigma^+) + \text{O}(\text{P}^3_g)$ and the repulsive $\text{CO}_2^+(a^4\Sigma_g^-)$ state connected with $\text{O}^+(\text{S}^4_u) + \text{CO}(\text{X}^1\Sigma^+)$ cover the energy range of 7.02–8.76 eV, where $b^4\Pi_u/4B_1$ represents either $^4\Pi_u$ for the linear geometry or 4B_1 for the bent geometry (more stable geometry for this state) [39]. Thus, the dissociation should depend on the ways to reach $\text{CO}_2^+(b^4\Pi_u/4B_1)$ or $\text{CO}_2^+(a^4\Sigma_g^-)$. The direct transitions of $b^4\Pi_u/4B_1 \leftarrow \text{A}^2\Pi_u$ and $a^4\Sigma_g^- \leftarrow \text{A}^2\Pi_u$ are forbidden by selection rules of $u \leftarrow u$ or/and $S = 3/2 \leftarrow | \rightarrow S = 1/2$. Additionally, the direct predissociation of $\text{CO}_2^+(\text{C}^2\Sigma_g^+)^{\dagger}$ ions via the $a^4\Sigma_g^-$ state to form $\text{O}^+(\text{S}^4_u) + \text{CO}(\text{X}^1\Sigma^+)$ is not possible because the spin–orbit coupling integral for these states $\langle a^4\Sigma_g^-, m_s = 3/2 | \mathbf{L} \cdot \mathbf{S} | \text{C}^2\Sigma_g^+, m_s = 1/2 \rangle$ gives no contribution for $\Delta m_s = 1$. Moreover, the predissociation of the vibrationally excited $\text{CO}_2^+(\text{C}^2\Sigma_g^+)^{\dagger}$ state at energies above the second dissociation limit $\text{CO}^+(\text{X}^2\Sigma^+) + \text{O}(\text{P}^3_g)$ can take place via the repulsive $b^4\Pi_u/4B_1$ state because the spin–orbit coupling between these two electronic states is highly efficient ($\approx 40 \text{ cm}^{-1}$). Other ways leading to $\text{O}^+(\text{S}^4_u) + \text{CO}(\text{X}^1\Sigma^+)$ by coupling between $\text{CO}_2^+(b^4\Pi_u/4B_1)$ and $\text{CO}_2^+(a^4\Sigma_g^-)$, which leads to $\text{CO}^+(\text{X}^2\Sigma^+) + \text{O}(\text{P}^3_g)$ by the spin–orbit interaction between $\text{CO}_2^+(b^4\Pi_u/4B_1)$ and $\text{CO}_2^+(\text{X}^2\Pi_g)^{\dagger}$ are also possible.

As indicated by the branching ratios of 7.2–11.5 for $[\text{CO}^+]/[\text{O}^+]$, a strong preference in the [1 + 1] photodissociation process to form CO_2^+ is observed for the formation of the $\text{CO}^+(\text{X}^2\Sigma^+) + \text{O}(\text{P}^3_g)$ channel compared to the formation of the lowest product channel $\text{O}^+(\text{S}^4_u) + \text{CO}(\text{X}^1\Sigma^+)$. This fact can be rationalized by the more efficient spin–orbit couplings between this state and the $b^4\Pi_u/4B_1$ state because $\text{CO}_2^+(\text{C}^2\Sigma_g^+)^{\dagger}$ is prepared in states at 7.02–8.76 eV.

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