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## High-resolution infrared spectra of bicyclo[1.1.1]pentane <sup>☆</sup>

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### ABSTRACT

Infrared spectra of bicyclo[1.1.1]pentane ( $C_5H_8$ ) have been recorded at a resolution (0.0015 cm<sup>-1</sup>) sufficient to resolve for the first time individual rovibrational lines. This initial report presents the ground state constants for this molecule determined from the detailed analysis of three of the ten infraredallowed bands,  $v_{14}(e')$  at 540 cm<sup>-1</sup>,  $v_{17}$  ( $a_2''$ ) at 1220 cm<sup>-1</sup>,  $v_{18}(a_2'')$  at 832 cm<sup>-1</sup>, and a partial analysis of the  $v_{11}(e')$  band at 1237 cm<sup>-1</sup>. The upper states of transitions involving the lowest frequency mode,  $v_{14}(e')$ , show no evidence of rovibrational perturbations but those for the  $v_{17}$  and  $v_{18}(a_2'')$  modes give clear indication of Coriolis coupling to nearby e' levels. Accordingly, ground state constants were determined by use of the combination-difference walues, yielding the following constants for the ground state (in units of cm<sup>-1</sup>):  $B_0 = 0.2399412(2)$ ,  $D_J = 6.024(6) \times 10^{-8}$ ,  $D_{JK} = -1.930(21) \times 10^{-8}$ . For the unperturbed  $v_{14}(e')$  fundamental, more than 3500 transitions were analyzed and the band origin was found to be at 540.34225(2) cm<sup>-1</sup>. The numbers in parentheses are the uncertainties (two standard deviations) in the values of the constants. The results are compared with those obtained previously for [1.1.1]propellane and with those computed at the *ab initio* anharmonic level using the B3LYP density functional method with a cc-pVTZ basis set.

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#### 1. Introduction

The properties of molecules with great Bayer ring strain have in recent years been at the focus of much attention of organic and quantum chemists. Among these molecules are the cyclopropanes, cyclobutanes, propellanes and others. All of these molecules contain carbon rings, the smaller of which exhibit the greatest amount of strain [1–6]. Except for the smallest members of these families, most have not been examined by high-resolution spectroscopic methods. Only cyclopropane ( $C_3H_6$ ) has been extensively studied [7], as it is commercially available and is stable in spite of the ring strain caused by the small  $60^{\circ}$  C—C—C angles.

Other interesting small cyclic molecules, such as cubane ( $C_8H_8$ ) [8], [1.1.1]propellane ( $C_5H_6$ ) [9], bicyclo[1.1.1]pentane ( $C_5H_8$ ) [10], and spiropentane are not commercially available and must be synthesized. Such molecules exhibit high symmetry and, in the case of propellane, very unusual chemical bonding. These features make them of special interest for structural investigations by X-ray, neu-

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tron, and electron diffraction methods and, in particular, by highresolution spectroscopic techniques of the free molecules in the gaseous state at low pressures. Comparisons of moments of inertia, interatomic bond lengths, and interbond angles obtained by diffraction, spectroscopic, and theoretical methods have been discussed at length in several publications that point out the great importance of accurate structural and spectroscopic data [11-13]. In addition to structural parameters, high-resolution spectroscopic studies yield many other standard molecular parameters such as the B, D, v,  $\alpha$ ,  $\zeta$ , x, k, ... coefficients for the various vibrational states, as well as less common vibration-rotation interaction terms. Furthermore, band intensities can give information about the electronic environment in which the atomic nuclei execute their vibrational motions. Many of these molecular constants can be calculated by *ab initio* methods, which can be used as a guide for the analysis of vibration-rotation spectra and, conversely, can be improved through comparisons with experimentally determined parameters. In this paper, we present the results of such studies for bicyclo[1.1.1]pentane and offer comparisons with [1.1.1.] propellane. The  $D_{3h}$  structures of these molecules are shown in Fig. 1 and both are appealing for study since they are small enough for high level ab initio calculations and are symmetric top rotators so that detailed spectroscopic analysis is readily feasible.



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Fig. 1. Structural arrangements for [1.1.1]propellane and bicyclo[1.1.1]pentane. Both molecules possess the symmetry of the  $D_{3h}$  point group.

Wiberg et al. [14] have studied the vibrational spectrum of the bicyclo[1.1.1]pentane as well as the intensities of the infrared transitions of this parent  $(-d_0)$  molecule and its mono  $(-d_1)$ and dideutero  $(-d_2)$  isotopomers, with the D-atoms located on the principal  $C_3$  symmetry axis [15]. (For brevity, we henceforth omit the d<sub>0</sub> and [1.1.1] specifiers for bicyclopentane and propellane.) Values for most of the infrared and Raman active vibrational fundamentals of the parent compound and the two isotopomers were reported. Band contours of the infrared bands of the parent compound and its symmetric top d<sub>1</sub> and d<sub>2</sub> isotopomers were recorded at a resolution of 0.25 cm<sup>-1</sup> which aided in making the vibrational assignments. At a higher resolution of 0.06 cm<sup>-1</sup>, infrared spectra of the  $v_{17}$  and  $v_{18}$  rovibrational bands of the parent d<sub>0</sub> and d<sub>2</sub> isotopomers of bicyclopentane were partially resolved and this was sufficient to give reasonable estimates of the  $I_B$  inertial constants. An electron diffraction study by Almenningen et al. [16] provided structural data which were in good accord with the spectroscopic  $I_B$  value for bicyclopentane. A normal coordinate analysis and ab initio calculations at the MP2/6-31G\* level were also performed by Wiberg and coworkers. When scaled, the vibrational frequencies were in fair agreement with the measured values, as were the calculated moments of inertia of the  $d_0$  and  $d_2$  isotopomers with the  $I_B$  values determined from the band analyses. Tables 1 and 2 present the diffraction and spectroscopic results, along with ab initio results we obtained using anharmonic methods and a larger basis set than in previous calculations.

Much higher resolution spectroscopic measurements are currently possible than in the earlier work on bicyclopentane and on propellane. In view of the particular chemical interest of

1.557(2)

1.874(4)

1.107(4)

111.7(18)

1.107(4)

70 38

81.77

0.2395

0.2062

Table 2

Fundamental modes of bicyclopentane (cm<sup>-1</sup>).

Mode		Ab initio result	Ab initio results <sup>a</sup>	
		Harmonic	Anharmonic	
a'1	1	3092	2950	2976
·	2	3042	2906	2887
	3	1554	1500	1509
	4	1119	1096	1107
	5	908	888	898
$a'_2$	6	3095	2952	
-	7	969	954	
e'	8	3100	2956	2976
	9	3037	2888	2887
	10	1504	1492	1456
	11	1259	1226	1232
	12	1117	1093	1098
	13	898	874	886
	14	542	538	540
a''	15	1004	977	
a''_	16	3087	2952	2976
2	17	1252	1221	1220
	18	844	825	832
e″	19	1216	1186	
	20	1144	1113	
	21	1029	1005	1006
	22	774	748	769

<sup>a</sup> This work, B3LYP/cc-pVTZ calculations using Gaussian 03 with Anharm/Vibrot options.

Ref. [15].

the latter, we recently reexamined its spectrum at a resolution of about 0.002 cm<sup>-1</sup> and reported some of our results in two previous publications [17,18]. The detail revealed in that case by the 30-fold increase in resolution over that available for the earlier studies [19] prompted us to also reinvestigate the infrared spectra of bicyclopentane (C<sub>5</sub>H<sub>8</sub>) as well as of monodeuterated bicyclopentane-d<sub>1</sub> (C<sub>5</sub>H<sub>7</sub>D) with C<sub>3v</sub> symmetry. We report here for the C<sub>5</sub>H<sub>8</sub> molecule upper state constants for the unperturbed  $v_{14}(e')$  band, along with ground state constants obtained from the combination-difference analysis of the combined data sets of the  $v_{14}(e')$ ,  $v_{17}(a''_2)$ , and  $v_{18}(a''_2)$  fundamental bands. These parameters are compared to analogous results obtained for propellane and to ab initio values computed at the anharmonic level. For the  $v_5(a)$  1216 cm<sup>-1</sup> band of bicyclopentane-d<sub>1</sub> and the  $v_{17}(a_2'')$  and  $v_{18}(a_2'')$  upper states of bicyclopentane, evidence for Coriolis interactions with nearby e states was seen and the analysis of this effect and the parameters for these states will be given in a subsequent paper.

Ab initio B3LYP/cc-pVTZb

 $(R_0)$ 

1.524

1.577

1.088

115

58 87

87.91

0.2863

0.1918

 $(R_e)$ 

1.515

1 568

1.082

115.1

58 19

86.89

0.2897

0.194

<b>Table 1</b> Experimental and	ab initio structu	aral parameters for bicyclopent	ane and propellar	ne.	
Parameter <sup>a</sup>	Unit	Bicyclopentane			Propellane
		Electron diffraction <sup>a</sup>	Ab initio B3LYP/cc-pVTZb		Electron diffraction <sup>ba</sup>
		$(R_a)$	$(R_e)$	$(R_0)$	$(R_a)$

1.554

1.876

1.09

111.6

1.088

69 87

80.94

0.2413

0.2083

<sup>a</sup> Thermal-average (*R<sub>a</sub>*) parameters from Refs. [16,29]. (*R<sub>e</sub>*) and (*R*<sub>0</sub>) refer to parameters for the equilibrium and ground state structures. The uncertainties in the last digits are given in parentheses.

1.562

1 886

1.095

111.4

1.097

70 67

81.94

0.2387

0.2058

1.522(2)

1.592(5)

1.100(5)

58 82

87.22

0.2866

0.1933

116.0(19)

This work, using Gaussian 03 with Anharm/Vibrot options.

Å

Å

Å

Å

amu-Å<sup>2</sup>

amu-Å<sup>2</sup>

 $cm^{-1}$ 

 $\mathrm{cm}^{-1}$ 

Cax-Ceq

C<sub>eq</sub>-H

Н-С-Н

Cax-H

 $I_B$ 

 $I_C$ В

С

Cax

-C<sub>ax</sub>

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