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## Absolute line intensities in methyl bromide: The 7-µm region

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#### Abstract

This work deals, for the first time, with the modeling of absolute line intensities in the fundamental  $v_2$  and  $v_5$  bands of CH<sub>3</sub><sup>81</sup>Br and CH<sub>3</sub><sup>81</sup>Br at 7 µm. For that, four unapodized absorption spectra of CH<sub>3</sub>Br (natural abundance, 99% purity,  $P \times L = 0.082 - 0.165$  atm × cm, room temperature) were measured in the range 1260–1560 cm<sup>-1</sup>, at a resolution of 0.002 cm<sup>-1</sup> using a Fourier transform spectrometer Bruker IFS 120 HR. For both isotopomers, 313 line intensities were analyzed within the dyad system required to account properly for the strong Coriolis coupling between  $v_2$  and  $v_5$ . The intensity fit of experimental data led to the determination of the dipole moment derivatives  $d_2 = \partial \mu/\partial q_2$  and  $d_5 = \partial \mu/\partial q_5$  relative to the  $v_2$  and  $v_5$  bands, as well as the first-order Herman–Wallis correction in K to  $d_5$ . The observed line intensities are fitted to 3.0% (3.3%) for  $v_2$  at 1309.9 cm<sup>-1</sup> and 2.6% (3.0%) for  $v_5$  at 1442.9 cm<sup>-1</sup>, respectively for CH<sub>3</sub><sup>79</sup>Br and CH<sub>3</sub><sup>81</sup>Br. The values derived for the vibrational band strengths of  $v_2$  and  $v_5$  are 55.7(0.6) and 39.2(0.3) cm<sup>-2</sup> atm<sup>-1</sup> at 296 K, respectively. The corresponding assignments and line positions of the dyad from previous work [F. Kwabia Tchana, I. Kleiner, J. Orphal, N. Lacome, O. Bouba, J. Mol. Spectrosc. 228 (2004) 441] are combined with the present intensity study to provide an improved CH<sub>3</sub>Br database for atmospheric applications.

Keywords: Methyl bromide; Infrared spectra; Dyad; Line intensities; Dipole moment

#### 1. Introduction

Methyl bromide (CH<sub>3</sub>Br) is an atmospheric trace gas of interest because of its contribution to stratospheric ozone depletion. Methyl bromide has both natural and anthropogenic origins. Its known sources include natural production from oceans [1] and biomass burning [2]. Methyl bromide is also industrially produced for use as an agricultural fumigant. With a tropospheric mixing ratio of 9–11 pptv in the Northern Hemisphere (with an increase of about 0.15 pptv per year) and about 8 pptv in the Southern Hemi-

sphere, it is believed to be the single largest contributor of stratospheric bromide [3]. However, until present, no attempts have been made to determine atmospheric concentrations of CH<sub>3</sub>Br using infrared spectroscopy. For this, accurate modeling of the infrared spectrum of CH<sub>3</sub>Br, including line intensities, is indispensable.

There have been various investigations in the past on the infrared and microwave spectra of methyl bromide. An extensive review of the spectroscopy of this molecule was given by Graner [4]. The most recent infrared work on this molecule including line intensities was published in 2002 by Brunetaud et al. [5]. In that work, high-resolution spectra of the  $v_6$  band of CH<sub>3</sub>Br between 820 and 1120 cm<sup>-1</sup> were recorded and line positions and intensities were predicted for atmospheric remote-sensing applications.

Although the line positions in the 7- $\mu$ m region (containing the two interacting fundamentals  $v_2$  and  $v_5$ ) have been

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Table 1 Experimental conditions used to record the FTIR spectra of CH<sub>3</sub>Br<sup>a</sup>

Spectrum No.	Path (cm)	Pressure <sup>b</sup> (mbar)	Temperature (K)	Resolution (cm <sup>-1</sup> )
Bruker IFS 120 HR a	t LADIR, Paris, Bandpass	: 1150–1550 cm <sup>-1</sup>		
1.	415 (1)	0.1991 (8)	297 (1)	0.002
2.	415 (1)	0.2778 (11)	296 (1)	0.002
3.	415 (1)	0.3415 (14)	298 (1)	0.002
4.	415 (1)	0.4028 (16)	296 (1)	0.002
5.°	27.0 (0.1)	4.693 (5)	296 (1)	0.004

Note. The resolution is equal to 0.9/(maximum optical path difference) and 1 atm = 1013 mbar.

- <sup>a</sup> The numbers in parentheses represent the absolute uncertainty in the units of the last digit quoted.
- <sup>b</sup> The absolute uncertainty is estimated to be 0.4% of the measured pressure.
- <sup>c</sup> Recorded using the Bruker IFS 120 HR located at LPPM, Orsay (see [6] for more details).

reinvestigated recently at high spectral resolution [6], little is known about the line intensities in this region. The integrated band intensities of these two overlapping bands were measured in the past by different groups [7–10], using band separation techniques applied to low-resolution spectra. However, no line by line study of absolute intensities at 7 μm of CH<sub>3</sub>Br is presently available. At the present time, no spectroscopic information (including line positions, intensities, and linewidths) on CH<sub>3</sub>Br is available from either the HITRAN [11] or the GEISA [12] databases.

The purpose of the present study was to measure the absolute infrared intensities for the two fundamentals  $v_2$  and  $v_5$  of CH<sub>3</sub><sup>79</sup>Br and CH<sub>3</sub><sup>81</sup>Br, and to provide a CH<sub>3</sub>Br compilation at 7 µm for databases. The intensity parameters were derived by analyzing a set of individual line intensities accurately measured using a Fourier transform spectrometer Bruker IFS 120 HR. Modeling of the intensities was achieved within a two-interacting-band system, i.e.,  $v_2$  and  $v_5$ , such a model being required to account properly for the strong Coriolis coupling between  $v_2$  and  $v_5$ . The formulation of the model was developed in [13] for a vibrational system including up to five interacting

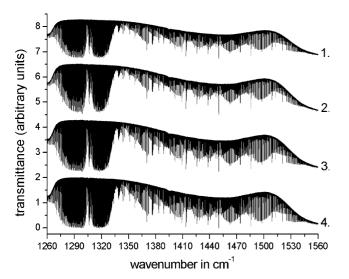


Fig. 1. Unapodized absorption spectra of CH<sub>3</sub>Br between 1260 and 1560 cm<sup>-1</sup>, recorded at a resolution of 0.002 cm<sup>-1</sup> using a Bruker IFS 120 HR located at LADIR (Paris, France), with the experimental conditions reported in Table 1.

bands. The intensity fit of experimental data led to the determination of the dipole moment derivatives  $d_2 = \partial \mu / \partial q_2$  and  $d_5 = \partial \mu / \partial q_5$  relative to the  $v_2$  and  $v_5$  bands, as well as the first-order Herman-Wallis correction in K to  $d_5$ . A new evaluation of the individual band strengths  $S_2$  and  $S_5$  is made and compared with previous determinations [7–9].

The following sections present, respectively, the experimental details, the theoretical treatment leading to line intensity data, the procedure used for extracting the individual band strengths  $S_2$  and  $S_5$ , and comparisons with previous works. From the resulting values of the dipole moment derivatives, a global line-by-line prediction is now available for atmospheric applications. For applications of our database to atmospheric remote sensing and retrieval of methyl bromide concentrations, we plan also to investigate line broadening in this region. That work is in progress.

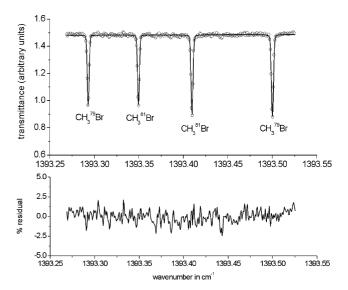


Fig. 2. Line-by-line retrieval of positions and intensities using non-linear least squares fitting technique at 1393 cm<sup>-1</sup>. The differences are minimized by adjusting the assumed positions and intensities of lines in the synthetic spectrum. The methyl bromide gas pressure is  $0.4028 \pm 0.0016$  mbar at  $296 \pm 1$  K, the optical path  $415 \pm 1$  cm, and the resolution is 0.002 cm<sup>-1</sup>. Upper panel: observed (open circles) and synthetic (solid line) spectra overlaid. Lower panel: differences between the two spectra in percent.

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