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Complete conformational analyses of perfluoro-*n*-pentane, perfluoro-*n*-hexane, and perfluoro-*n*-heptane

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ABSTRACT

Complete conformational analyses of perfluoro-n-pentane, perfluoro-n-hexane, and perfluoro-n-heptane were carried out at both the semiempirical PM3 and the ab initio Hartree–Fock RHF/3-21G^(*) levels of theory, and confirmed at the RHF/6-31G* level of theory, using the PC Spartan Pro computational package. These analyses indicated that the lowest-energy conformation of each of the perfluoroalkanes is helical. The conformational analyses also demonstrated that while helical or partially helical conformations dominate the lower-energy portions of the total conformational manifolds of these molecules, the fully staggered, zig-zag conformer analogous to the corresponding hydrocarbon molecules was also present in the lower-energy portions of the conformational manifolds of these molecules. The dihedral angle (\angle F-C-C-F) leading to helical structures in these perfluorocarbon molecules ranges from approximately 46° at the ends of the molecules to approximately 49° near the centers of the molecular chains, compared to the 60° dihedral angle (\angle H-C-C-H) expected for the fully staggered, zig-zag structure observed for hydrocarbons and also observed as higher-energy conformations in the perfluorocarbons.

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1. Introduction

Poly(tetrafluoroethylene), commonly known as Teflon® or PTFE, has been known since 1954 to form helical chain conformations [1]. In fact, PTFE exhibits three phases, all with helical structures, at atmospheric pressure near room temperature [2]. Below 19 °C, the polymer chains exhibit a 54/25 ratio of CF₂ groups to helical turns [3]; this structure is referred to as phase II. Between 19 °C and 30 °C, the polymer chains adopt phase IV, a very slightly expanded 15/7 helix [4]. Such helical structures do not exist for poly(ethylene), the hydrocarbon analog of PTFE. However, at temperatures above 30 °C, PTFE does slowly convert to phase I in which a 2/1 planar zig-zag structure, similar to that of poly(ethylene), appears [5]. Similar spiral conformations with 15/7 helices have been observed in perfluoro-n-eicosane ($C_{20}F_{42}$) and perfluoro-n-tetracosane ($C_{24}F_{50}$) [6].

Perfluoroalkyl substituent groups have also been shown experimentally to exhibit structural helicity. For example, the perfluorohexyl group substituents on phosphines and phosphine ligands bound to transition metals [7,8] and on binaphthyl [9] exhibit helical structures. In some cases, however, the structures also develop "kinks" – described as *pseudo*-boat conformations – that interrupt the helicity of the perfluorocarbon substituent.

There are a small number of reports of the investigation of the structures of perfluorocarbon molecules or PTFE from a computational standpoint. Perfluoro-n-butane was studied using the 6-31G* basis set [10] and with the DZ+D_c basis set [11]; the calculated geometries did not differ significantly when the results from the two basis sets were compared. The lowest-energy conformations of perfluoro-n-butane, -n-pentane, and -n-hexane were examined using both the 4-31G and D95+* basis sets in order to develop an understanding of the conformational structure of PTFE [12]; again, the calculated geometries did not differ significantly when the results from the two basis sets were compared. A density functional theory method, B3LYP/6-31G*, has been used to investigate the lowest-energy structures and properties of perfluoromethane, -ethane, -propane, -n-butane, and -n-hexane in order to develop uncorrelated force field parameters that were then applied to longer perfluorocarbons: perfluoro-n-decane and -n-eicosane [13]. Finally, the local density functional AIMPRO code [14] was used to investigate the structures and properties of perfluoromethane, -ethane, and -propane [15]. While the C_{2v} -symmetry ground state of perfluoropropane was found to be fully staggered, a C_2 symmetry conformation of perfluoropropane was also identified [15] that was 0.525 kcal/mol (2.20 kJ/mol) higher in energy than the ground state. The torsion angles in this higher-energy conformer suggested the potential for helicity even in this short, three-carbon molecule.

Although there have been some partial conformational studies of perfluorocarbon molecules published, none have reported more

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Table 1The RHF/3-21G^(*) and RHF/6-31G* absolute and relative energies (in the RHF/6-31G* order) of the conformational isomers of perfluoro-*n*-pentane, perfluoro-*n*-hexane, and perfluoro-*n*-heptane and the RHF/6-31G* torsion angles that define the conformers

perfluoro-n-heptane and the RHF/6-31G* torsion angles that define the conformers											
Conformer	Electronic energy (a.u.)			Relative energy (kJ/mol)			Torsion angles (RHF/6-31G*)				
	RHF/3-21G(*)	RHF/6-31	IG* 1	RHF/3-21G ^(*)	RHF/	6-31G*	∠F–C¹–C	c^2 - C^3 $\angle C$	$C^1 - C^2 - C^3 - C^4$	$\angle C^2$ - C^3 - C^4 - C^5	$\angle C^3 - C^4 - C^5 - F$
Perfluoro-n-	-pentane (1)										
1a	-1375.12734	80 –1382.60	067048	0.000	0.00	00 5	50.82°	1	164.47°	164.46°	50.95°
b	-1375.12112			16.346	2.17		50.70°		179.99°	179.99°	60.71°
c	-1375.12383			9.218	4.13		51.27°	1	165.66°	55.50°	51.40°
d	-1375.11801			24.496	10.90		51.68°		58.74°	58.78°	51.72°
e	-1375.11935	49 –1382.59	990037 .	20.986	18.48	51 5	51.65°	-	101.66°	62.65°	54.80°
Conformer	Electronic energ	y (a.u.)	Relative	Relative energy (kJ/mol)) Torsion angles		s (RHF/6-31G*)			
	RHF/3-21G ^(*) RHF/6-31G [*]		RHF/3-21G ^(*) RHF/0		7/6-31G*	$\angle F-C^1-C^2-C$	- , , ,		$\angle C^2 - C^3 - C^4 - C^5$	$C^2 - C^3 - C^4 - C^5$ $\angle C^3 - C^4 - C^5 - C^6$	
D (1		KIII/0-31G	KIII/3-	ZIG KIII	70-31G	Zr-C -C -C			ZC -C -C -C	ZC -C -C -C	$\angle C^4 - C^5 - C^6 - F$
Perfluoro-n-	-hexane (2)										
2a	-1610.5876630				000	50.59°		64.18°	162.22°	164.22°	50.70°
b	-1610.5841755				322	51.96°		66.56°	55.36°	166.54°	51.89°
c d	-1610.5843208 -1610.5783057				360 515	51.86° 60.74°		55.94° 79.96°	163.25° 179.94°	163.90° 179.95°	50.74° 60.73°
e	-1610.5859846				599	51.15°		66.90°	−96.10°	166.90°	51.26°
f	-1610.5857047				533	50.60°		64.21°	165.24°	−95.67°	51.91°
g	-1610.5801658	-1619.34400			747	51.14°		55.12°	164.34°	55.20°	51.32°
h	-1610.5797124			10.8		53.08°		61.42°	57.12°	164.32°	50.61°
i	-1610.5789787			12.6		45.67°		57.22°	179.77°	-57.50°	-45.42°
J k	-1610.5824151 -1610.5845116			12.8 16.2		51.04° 50.12°		55.38° 64.00°	166.33° -89.65°	−96.76° −93.73°	52.09° 50.86°
k l	-1610.5739083			16.5		50.12°		58.33°	-89.65° 59.46°	-93.73° 58.38°	50.86°
m	-1610.5839536			17.3		51.69°		96.16°	167.90°	−96.20°	51.81°
n	-1610.5802280	-1619.34054	62 19.521	17.8	318	53.65°	1	00.12°	-178.14°	58.35°	−43.07°
0	-1610.5797040			18.2		55.96°		62.48°	−101.00°	167.28°	51.40°
p	-1610.5794660			19.3		55.15°		71.43°	63.03°	-101.10°	52.04°
q	-1610.5818031			23.0		53.92°		01.18°	179.93°	101.74°	-54.17°
r s	-1610.5785484 -1610.5719452			25.9 41.1		57.91° 54.32°		61.56° 96.61°	−95.99° −69.62°	−90.94° 97.14°	49.60° -53.75°
									00.02	07111	55.75
Conformer	Electronic energy	(a.u.)	Relative ener	gy (kJ/mol)	_	angles (RHF/					
	RHF/3-21G ^(*)	RHF/6-31G*	RHF/3-21G ^(*)	RHF/6-31G	* ∠F–C¹–	C^2 – C^3 $\angle C^1$ – C	C ² -C ³ -C ⁴	4 $\angle C^{2}$ – C^{3} – C^{4}	1 -C ⁵ \angle C ³ -C ⁴ -C ⁵	$-C^6 \angle C^4 - C^5 - C^6 - C^7$	$\angle C^5 - C^6 - C^7 - F$
Perfluoro-n-	-heptane (3)										
3a	-1846.0479058	-1856.0879489	0.000	0.000	50.78°	164.	22°	162.40°	162.40°	164.22°	50.74°
b		-1856.0863567	8.462	4.180	50.92°	164.		163.99°	55.65°	166.75°	51.90°
c	-1846.0446390	-1856.0861916	8.577	4.614	50.72°	164.	.09°	161.84°	163.15°	56.07°	52.02°
d		-1856.0852957		6.966	60.75°	179.		179.98°	179.98°	179.99°	–60.74°
e			18.580	8.463	50.73°		90°	164.75°	55.02°	166.39°	51.53°
f g		-1856.0846478 -1856.0846471	4.950 4.067	8.667 8.669	51.03° 50.96°	164. 164.		161.99° 164.68°	165.17° -95.42°	−95.53° 166.50°	51.94° 50.87°
g h		-1856.0846124		8.760	52.07°		95°	162.86°	163.00°	55.81°	51.76°
i	-1846.0401265			11.077	50.66°	163.		162.07°	57.23°	61.60°	52.83°
j	-1846.0414484	-1856.0836157	16.954	11.377	50.58°	164.	46°	59.74°	59.64°	164.36°	50.50°
k		-1876.0834699		11.760	46.76°		38°	178.10°	−58.41°	160.26°	49.49°
l	-1846.0431087			12.596	51.84°	166.		55.44°	166.66°	-96.94	-52.41°
m n	-1846.0429274 -1846.0426102			12.950 13.063	51.39° 51.72°		54° 80°	165.87° 163.18°	-96.64° 164.97°	166.88° -95.77°	50.93° 51.70°
n o	-1846.0368325			13.395	47.87°		97°	174.11°	-173.03°	-55.23°	-48.56°
р	-1846.0367231			15.091	51.11°	166.		174.54°	−63.77°	−51.88°	−50.50°
q	-1846.0357204			15.815	53.24°		51°	56.71°	163.06°	55.51°	51.09°
r	-1846.0451903		7.130	16.073	50.78°	165.		−91.71°	−91.95°	165.30°	50.32°
S	-1846.0447792		8.209	16.128	51.01°	164.		162.37°	-89.10°	-94.18°	51.23°
t		-1856.0816087 -1856.0815785		16.646 16.725	48.84° 49.37°		45°	61.10° 59.30°	57.81° –177.14°	163.61° 99.42°	50.49° -53.23°
u v	-1846.0425389 -1846.0439984			16.725	49.37° 51.90°	158. –95.		59.30° 164.98°	-177.14° 165.11°	99.42° -95.62°	−53.23° 52.05°
w	-1846.0445738		8.748	17.309	50.86°	166.		−95.72°	167.24°	-95.38°	51.59°
X	-1846.0408945	-1856.0809667	18.408	18.332	51.79°	169.		−99.27°	177.56°	-58.71°	−42.20°
у		-1856.0809556		18.361	50.75°	164.		164.92°	-100.16°	62.27°	56.35°
Z	-1846.0338140			18.747	48.31°		57°	174.59°	-64.01°	-52.11°	-50.44°
aa	-1846.0396082			19.216	55.94°	173.		63.32°	-101.40°	167.84°	51.96°
ab	-1846.0380074 -1846.0399311			19.816 20.367	52.36° 50.54°	–96. 164.		165.30° 167.28°	57.33° 63.65°	61.09° -101.11°	52.81° 53.45°
ac ad	-1846.0399311 -1846.0414728			20.367	50.54° 52.68°		50°	167.28° 163.86°	-90.45°	-101.11° -93.52°	53.45° 50.92°
ae	-1846.0369751			22.479	51.34°		74°	166.07°	-101.28°	62.57°	56.00°
af	-1846.0293379			22.626	50.20°		79°	59.36°	59.48°	57.71°	49.96°
ag	-1846.0435193			22.806	53.37°	171.		-102.82°	−77.03°	-103.72°	54.89°
ah	-1846.0355232			23.308	50.35°		72°	64.54°	-174.42°	98.58°	−52.60°
ai	-1846.0360813			24.478	49.07°		66°	167.55°	63.04°	-100.80°	53.09°
aj	-1846.0429726	-1850.0784743	12.952	24.876	50.74°	−93 .	30°	164.64°	-89.09°	-94.78°	51.29° d on next page)
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