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#### Ab-initio study of the structural, electronic and optical properties of BSb 1 (110) and (100) surfaces 2

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### ABSTRACT

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were investigated. The calculations were performed by using Full-Potential Linear Augmented Plane Wave 20 (FP-LAPW) in Density Functional Theory (DFT) framework with various exchange-correlation functionals. 21 Some surface properties such as surface energies, the work functions, surface relaxation, band structures and op- 22 tical properties of cubic BSb (110) and (100) nano-slabs were studied. The (100) surface of cubic BSb has B- and 23 Sb-terminations while the (110) surface has only one termination. The surface properties of all the terminations 24 were presented and discussed. Moreover, some bulk properties such as structural, electronic, band structure and 25 optical properties of BSb compound were investigated for comparison. The obtained band gaps for the BSb (110) 26 nano-slab were 0.38 eV while the BSb (100) nano-slab showed a metallic behavior. Moreover, the surface states 27 of the slabs were identified. The real and imaginary parts of the dielectric function of the BSb (110) nano-slab 28 were also calculated and compared with bulk results.

In the present study, the structural, electronic and optical properties of BSb bulk, BSb (110) and (100) surfaces

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#### 1. Introduction 35

The group III-V semiconductors are widely used for electronic and 36 optoelectronic tools. [1]. Among these compounds, the narrow band 37 gap antimonide based compound semiconductors (ABCS) are widely 38 considered as the first candidate materials for production of the third 39 generation infrared photon detectors and ICs with ultra-high speed 40 and ultra-low power consumption. Due to their unique band gap struc-41 42 ture and physical properties, it makes a wide space to develop new varied devices [2]. However, III-Sb compounds have attracted more 43attention due to their potential usage in rechargeable lithium batteries. 44 45Among III-Sb compounds, boron antimonide exhibits a strong covalent 46 character and shows an uncommon behavior because of small core and lack of "p" electrons in boron atom compared to other group III-V 47 compounds. It makes this material a potential material for high temper-48 49 ature electronic and optical application [3]. The surfaces of group III-V semiconductors have a fundamental importance because their clean-50ness, geometry and photoelectric properties deeply affect the function 5152of modern electronic devices [4]. The (110) nonpolar surface of III-V 53compounds has been broadly investigated and more concentration 54has been on their relaxed geometry, phonon vibrations and band 55structure [5–9]. Surface energy and work function are two important

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http://dx.doi.org/10.1016/j.susc.2014.05.002 0039-6028/© 2014 Elsevier B.V. All rights reserved. parameters for fully recognizing the surfaces [9,10]. Nowadays, various 56 surfaces of group III-V compounds are developing owing to their appli-57 cations in thin films. Regarding the fact that thin film process technolo- 58 gy is firmly linked to the basic investigations on the surfaces and 59 interfaces, their fundamental parameters including surface energy and 60 work function must be specified first. Although they have wide applica- 61 tions, due to their difficulties in measurements few investigations are 62 available [11-13]. Surface energy is defined as the difference of free en- 63 ergy between a surface atom and that of an inner one for a solid which is 64 one of the most fundamental properties for expressing surfaces stabili- 65 ties [14]. In the most experimental works, surface energy is achieved by 66 extrapolating the value of liquid surface energies at high temperatures 67 which give rise to less valid experimental data [15]. Another important 68 property of surfaces is work function which was first introduced by 69 Einstein when he was working on the photoelectric effect and is 70 expressed as minimum required work to separate free electron from 71 the inner of a solid away from the surface into the vacuum region 72 [16]. Kelvin probe is proper device to calculate work function values 73 from contact potential difference between the surface under study 74 and a standardized surface [16,17]. Although this process is simple 75 and nondestructive the work function value is easily affected by impu-76 rities, roughness and misorientation on the surfaces [18]. Some other is-77 sues such as polycrystallinity of the sample and absence of required 78 ideal vacuum also lower the accuracy in experiments [12]. Regarding 79 the abovementioned complications that occur in calculating the surface 80 energy and work function in experimental studies, we believe that 81

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**Fig. 1.** (a) Bulk unit cell of BSb, (b) the top view of the BSb (110) slab, (c) symmetric 17 layer BSb (100) slab in the *z* direction (Sb-termination), (d) symmetric 17 layer BSb (100) slab in the *z* direction (B-termination), (e) symmetric 15 layer BSb (110) slab and (f) fully relaxed symmetric 15 layer BSb (110) slab.

theoretical studies can overcome these deficiencies. Over the last few
decades, surface energy and work function of some pure elements and
many other compounds have been calculated using DFT method. Most

of the investigations on the BSb compound have been theoretical. 85 Bagci and et al. reported pseudopotential study of structural, electronic 86 and vibrational properties of the BSb (110) surface in 2008 [19]. To the 87

#### t1.1 Table 1

t1.2 Calculated lattice constant, a<sub>0</sub> (Å), bulk modulus, B<sub>0</sub> (GPa) and its pressure derivative B' of the BSb compound in the ZB phase using different functionals.

t1.3	Approximations	a(Å)	B(GPa)	B'(GPa)
t1.4	GGA-PBE	5.2809	98.9264	4.7660
t1.5	LDA	5.1954	115.0498	5.0570
t1.6	PBEsol	5.2316	108.2133	4.9070
t1.7	WU-Chen	5.2354	107.7334	4.8818
	Other results	5.278 <sup>a</sup> , 5.279 <sup>b</sup> , 5.278 <sup>c</sup> , 5.191 <sup>d</sup> ,	100 <sup>a</sup> , 96 <sup>b</sup> , 100 <sup>c</sup> , 111 <sup>d</sup> , 116 <sup>e</sup> ,	4.40 <sup>a</sup> , 4.55 <sup>b</sup> , 4.40 <sup>c</sup> , 4.36 <sup>d</sup> , 4.16 <sup>e</sup> ,
t1.8		5.201 <sup>e</sup> , 5.145 <sup>f</sup> , 5.21 <sup>g</sup> , 5.252 <sup>h</sup> , 5.177 <sup>i</sup>	118 <sup>f</sup> , 110 <sup>g</sup> , 103 <sup>h</sup> , 110 <sup>i</sup>	4.31 <sup>f</sup> , 4.07 <sup>g</sup> , 3.62 <sup>h</sup> , 4.237 <sup>i</sup>
t1.9	<sup>a</sup> FP-GGA Ref. [28].			
t1.10	<sup>b</sup> FP-GGA Ref. [3].			
t1.11	<sup>c</sup> FP-GGA Ref. [29].			
t1.12	d FP-LDA Ref. [3].			
t1.13	<sup>e</sup> FP-LDA Ref. [29].			
t1.14	<sup>f</sup> PP-GGA Ref. [30].			
t1.15	<sup>g</sup> PP-GGA Ref. [31].			
t1.16	h FP-GGA Ref. [32].			
t1.17	<sup>i</sup> FP-GGA Ref. [33].			

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