

Angular distributions of desorbing N₂ in thermal N₂O decomposition on Rh(100)

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Abstract

The angular distribution of desorbing N₂ was studied in the decomposition of N₂O(a) on Rh(100) at 60–140 K by means of angle-resolved temperature-programmed desorption. N₂ desorption shows two peaks at around 80 K and 110 K. At low N₂O coverage, the former collimates far from the surface normal toward the [001] direction, whereas at high coverage, the desorption sharply collimates along the surface normal. The adsorption form of N₂O and its dissociation were also examined by DFT-GGA calculations. Dissociating N₂O is proposed to be lying along the [001] direction at low coverage and to change to an upright form bonding through the terminal oxygen at high coverage.

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1. Introduction

The reduction of nitrous oxide on rhodium surfaces has received much attention because of its importance in controlling the catalytic deNO_x process. N₂O is not only an undesirable byproduct in the NO reduction on this best catalyst but also the key intermediate in controlling the selectivity toward N₂ [1]. Knowledge of this intermediate is still limited because of the presence of several surface-nitrogen removal pathways in the deNO_x process. This paper is the first to deliver the angular distribution of desorbing N₂ in the thermal decomposition of adsorbed N₂O on Rh(100). N₂ desorption shows two peaks at around 80 K (β₂-N₂) and 110 K (β₁-N₂). The former sharply collimates at 66° off normal in the plane along the [001] direction at low N₂O coverage, indicating four-directional desorption, whereas, at high coverage, its desorption shifts along the surface normal. The adsorption form of N₂O and

its dissociation were also examined by density-functional theory calculations (DFT) with the generalized gradient approximation (GGA).

The characteristic spatial distribution of desorbing N₂ in the deNO_x process is useful to identify the intermediates emitting products. The desorption of N₂ with hyper-thermal energy in the N₂O decomposition is sharply collimated in an inclined way in the plane along the N–N–O bond on Pd(110) and Rh(110), i.e., the parent molecule orientation is preserved in the distribution. On Rh(100), similar inclined N₂ desorption along the [001] direction is observed only at limited N₂O coverage. At high coverage, another N₂O adsorption form is suggested.

2. Technical details

2.1. Experimental

Two UHV apparatuses were used. One has low-energy electron diffraction (LEED) and X-ray photon-electron spectroscopy (XPS) facilities for the survey of

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surface-cleaning procedures, and the other has three chambers for angle-resolved temperature-programmed desorption (AR-TPD) [2]. The reaction chamber has a mass spectrometer for angle-integrated (AI) desorption analysis as well as a cryo-plate cooled to about 25 K yielding a pumping rate of about $9 \text{ m}^3/\text{s}$, which is large enough to prevent the N_2 scattered in the reaction chamber from penetrating the analyzer [3].

A rhodium crystal with (100) planes (1 mm thick with a 9 mm diameter) was rotated to change the desorption angle (polar angle; θ) in the normally directed plane at the crystal azimuth along either the [001] or [011] direction. The LEED pattern showed a sharp (1×1) form after the surface was cleaned by Ar^+ ion bombardments, heating in 5×10^{-8} Torr oxygen at a surface temperature (T_s) of 850 K and then in 6×10^{-8} Torr hydrogen. The crystal was heated in H_2 before every N_2O exposure to remove the surface oxygen [4]. The $^{15}\text{N}_2\text{O}$ coverage, $\Theta_{\text{N}_2\text{O}}$, was determined from the $^{15}\text{N}_2\text{O}$ exposure relative to the value to a monolayer. The completion of a monolayer was defined by the appearance of a sharp $^{15}\text{N}_2\text{O}$ desorption peak from multilayers at 84 K. The density of the monolayer was estimated to be one fourth of the surface rhodium atom density from a comparison with the $^{13}\text{C}^{18}\text{O}$ desorption [5]. Hereafter, the isotope ^{15}N is described as N in the text. No super-structures were found in LEED observations after N_2O adsorption at 78 K.

2.2. Computational

The calculations were performed in the framework of DFT-GGA [6] using the plane-wave basis set and ultrasoft pseudopotentials [7,8] as implemented in the PWscf code contained in the Quantum ESPRESSO package [9], while molecular graphics were generated using the XCRYSDEN program [10]. The perfect Rh(100) surface was modeled by periodically repeated slabs consisting of five (100) layers and N_2O adsorption was modeled at $1/4$ and $1/9$ ML coverages using (2×2) and (3×3) supercells, respectively. Further computational details are described elsewhere [11]. Transition states and activation energies for N_2O dissociation have been calculated by climbing-image nudged-elastic-band method [12] using the (2×2) supercells.

3. Results

3.1. AR-TPD results

N_2O was introduced at $T_s = 55\text{--}60$ K. $\text{N}_2\text{O}(\text{a})$ is either desorbed in the subsequent heating or decomposed, emitting N_2 at 60–130 K (Fig. 1a). The N_2O desorption is noticeable above $\Theta_{\text{N}_2\text{O}} = 0.5$ in the range of 100–130 K. The N_2 desorption peaks at 100–120 K ($\beta_1\text{-N}_2$) and 70–95 K ($\beta_2\text{-N}_2$). The $\beta_1\text{-N}_2$ signal above $\Theta_{\text{N}_2\text{O}} = 0.7$ is largely corrected by the N_2O fragmentation in the analyzer. The $\beta_2\text{-N}_2$ signal shows remarkable enhancement in the surface normal direction. It increases rapidly above $\Theta_{\text{N}_2\text{O}} = 0.3$,

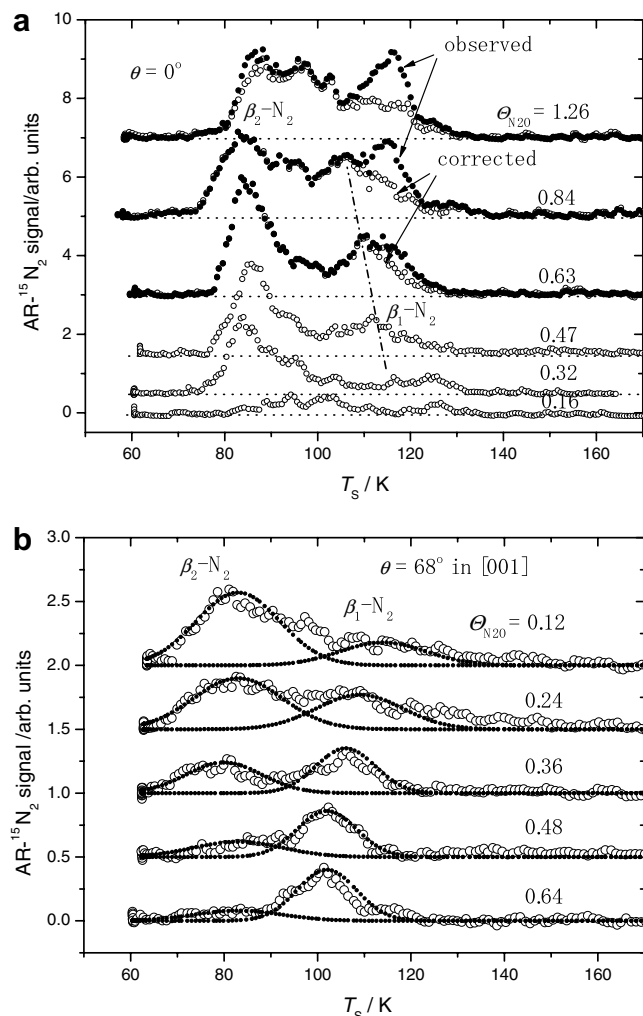


Fig. 1. AR-TPD spectra of desorbing $^{15}\text{N}_2$ from Rh(100) exposed to various amounts of $^{15}\text{N}_2\text{O}$ at 60 K. The desorption angles θ are (a) 0° and (b) 68° toward the [001] direction. Closed symbols; observed signal of mass/e = 30. Open symbols; signals after fragment correction. The heating rate was 1.5 K/s.

whereas the $\beta_1\text{-N}_2$ signal increases slowly. Below $\Theta_{\text{N}_2\text{O}} = 0.20$, the $\beta_2\text{-N}_2$ signal is intensified at around 66° off normal toward the [001] direction. The $\beta_2\text{-N}_2$ signal at this position decreases with increasing N_2O coverage (Fig. 1b). On the other hand, the $\beta_1\text{-N}_2$ signal increases and its peak shifts to lower temperatures. These comparisons indicate that the angular distribution of $\beta_2\text{-N}_2$ changes from an inclined form to a normally directed way with increasing N_2O coverage.

The AR-TPD spectra of $\beta_2\text{-N}_2$ below $\Theta_{\text{N}_2\text{O}} = 0.15$ are intensified at around $\theta = 68^\circ$ in the plane along the [001] direction (Fig. 2a). In the analysis of the noisy AR signals, the average level was estimated by curve fitting, in which a Gaussian form was assumed [13]. The resultant angular distribution along the [001] direction is sensitive to the coverage (Fig. 2b). The distribution changes from a three-directional form to the normally directed one with increasing N_2O coverage. The inclined desorption component is sharp as approximated in a $\cos^{25}(\theta \pm 66)$ form at

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