## **Accepted Manuscript**

How does aluminium foaming agent impact the geopolymer formation mechanism?

Ailar Hajimohammadi, Tuan Ngo, Priyan Mendis

PII: S0958-9465(16)30370-5

DOI: 10.1016/j.cemconcomp.2017.03.022

Reference: CECO 2810

To appear in: Cement and Concrete Composites

Received Date: 12 July 2016

Revised Date: 27 January 2017

Accepted Date: 27 March 2017

Please cite this article as: A. Hajimohammadi, T. Ngo, P. Mendis, How does aluminium foaming agent impact the geopolymer formation mechanism?, *Cement and Concrete Composites* (2017), doi: 10.1016/j.cemconcomp.2017.03.022.

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



#### ACCEPTED MANUSCRIPT

## Original manuscript before revision:

# How does aluminium foaming agent impact the geopolymer formation mechanism?

Ailar Hajimohammadi a\*, Tuan Ngo and Priyan Mendisa

<sup>a</sup>Department of Infrastructure Engineering, University of Melbourne, Victoria 3010, Australia

\*Corresponding author

### **Abstract**

Aluminium is widely used as a foaming agent in lightweight geopolymer concretes. The impact of aluminium reaction on the phase evolution of fly-ash based geopolymers is investigated. In-situ Fourier transform infrared spectroscopy (FTIR) is used to track the changes in the nanostructure of geopolymers. By combining the FTIR functional group analysis with Atomic-force microscopy (AFM) results, it is possible to evaluate the phase development during the early hours of reaction. At the initial stages of geopolymerization, aluminium reaction induces the formation of aluminium hydroxide gel which precipitates on fly ash and conceals the reactive surface of ash particles. Therefore, the dissolution rate of fly ash declines and the strength development is delayed. The high release rate of alumina into the solution, as a result of the aluminium metal reaction, leads to a better connectivity of unreacted particles and better microstructural development as shown in SEM images. Also, by consuming part of the sodium hydroxide catalyst at the beginning of the reaction, aluminium powder protects some of the alkali content of geopolymer matrix from carbonation.

Keywords: Geopolymer foam, Aluminium powder, mechanism, In-situ FTIR, kinetics

### Download English Version:

# https://daneshyari.com/en/article/5436877

Download Persian Version:

https://daneshyari.com/article/5436877

<u>Daneshyari.com</u>