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Fabrication of highly efficient $TiO_2/Ag/TiO_2$ multilayer transparent conducting electrode with N ion implantation for optoelectronic applications

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ABSTRACT

Fabrication of highly conductive and transparent TiO₂/Ag/TiO₂ (referred hereafter as TAT) multilayer films with nitrogen implantation is reported. In the present work, TAT films were fabricated with a total thickness of 100 nm by sputtering on glass substrates at room temperature. The as-deposited films were implanted with 40 keV N ions for different fluences $(1 \times 10^{14}, 5 \times 10^{14}, 1 \times 10^{15}, 5 \times 10^{15} \text{ and } 1 \times 10^{16} \text{ ions/cm}^2)$. The objective of this study was to investigate the effect of N⁺ implantation on the optical and electrical properties of TAT multilayer films. X-ray diffraction of TAT films shows an amorphous TiO₂ film with a crystalline peak assigned to Ag (111) diffraction plane. The surface morphology studied by atomic force microscopy (AFM) and field emission scanning electron microscope (FESEM) revealed smooth and uniform top layer of the sandwich structure. The surface roughness of pristine film was 1.7 nm which increases to 2.34 nm on implantation for $1 \times 10^{14} \, \text{ions/cm}^2$ fluence. Beyond this fluence, the roughness decreases. The oxide/metal/oxide structure exhibits an average transmittance ~80% for pristine and ~70% for the implanted film at fluence of 1×10^{16} ions/ $\rm cm^2$ in the visible region. The electrical resistivity of the pristine sample was obtained as $2.04 \times 10^{-4} \Omega$ cm which is minimized to $9.62 \times 10^{-5} \Omega$ cm at highest fluence. Sheet resistance of TAT films decreased from 20.4 to 9.62 Ω/\Box with an increase in fluence. Electrical and optical parameters such as carrier concentration, carrier mobility, absorption coefficient, band gap, refractive index and extinction coefficient have been calculated for the pristine and implanted films to assess the performance of films. The TAT multilaver film with fluence of 1×10^{16} ions/cm² showed maximum Haacke figure of merit (FOM) of $5.7 \times 10^{-3} \Omega^{-1}$. X-ray photoelectron spectroscopy (XPS) analysis of N 1s and Ti 2p spectra revealed that substitutional implantation of nitrogen into the TiO₂ lattice added new electronic states just above the valence band which is responsible for the narrowing of band gap resulting in the enhancement in electrical conductivity. This study reports that fabrication of multilayer transparent conducting electrode with nitrogen implantation that exhibits superior electrical and optical properties and hence can be an alternative to indium tin oxide (ITO) for futuristic TCE applications in optoelectronic devices.

1. Introduction

Transparent conducting electrode (TCE) is a key component in any optoelectronic device including solar cells, flat panel displays, light emitting diodes (LED) [1,2] etc. A good TCE should exhibit low resistivity and high transparency in the visible region of the electromagnetic spectrum [3,4]. Indium tin oxide has been widely used as TCE due to its high conductivity, high transparency and ease of deposition. However, In is high-cost material due to its increasing demand and natural scarcity. Doped single layers and multilayers of

TiO₂ with metal embedded between the two oxide layers have emerged as TCE with comparable optical and electrical properties for flat panel displays, LEDs and solar cells [5–7]. In addition, TiO₂ is a potential semiconductor material with technological importance because of high refractive index (2.7), wide band gap (~3.2 eV), high transmittance (~ 95%) in the visible region and high mechanical and chemical stability [8–10]. Doped metal oxides like Nb-doped TiO₂ (NTO) [11], Al-doped ZnO (AZO) [12], and F doped-SnO₂ (FTO) [13] have been commonly used as TCE in optoelectronic devices. In addition, electrical conductivity has been further increased by sandwiching a thin metal (M) film

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between these two dielectric (D) layers to form a DMD multilayer. Among metals, Ag is a good candidate because it exhibits low resistance and high transmittance (as a very thin layer) in the visible region, so Ag (<20 nm thick) [14] is commonly used as the interlayer in DMD multilayers. Many advantages of the metal embedded multilayer structure have been reported over the single layer TCO [15,16]. Researchers have been investigating various multilayers employing the Ag layer as a middle layer with different oxides such as SnO₂/Ag/ SnO₂ [17,18], SnO_x/Au/SnO_x [19], ZnO/Ag/ZnO [20], AZO/Ag/AZO [21], ITO/Ag/ITO [22] and TiO₂/Ag/TiO₂ [23] exhibiting low resistance and high transmittance in the visible region. Kim et al. [24] investigated the effect of Ag layer thickness on electrical and optical properties of TiO₂/Ag/TiO₂. Authors found highest transmittance of 86.3-97% at 591 nm for different Ag thicknesses, and sheet resistance slightly decreasing from 6.17 to 2.27 Ω /sq with increase in the Ag layer thickness. Dhar et al. [25] deposited TiO₂/Ag/TiO₂ by sputtering on the flexible substrate; Ag thickness was varied between 5 and 13 nm. It was reported that Ag layer became contiguous as it approaches the critical thickness of 9.5 nm and multilayer with Ag thickness 9.5 nm had the best figure of merit, $6.14 \times 10^{-4} \Omega^{-1}$. Many other researchers [26,27] have also investigated the properties of the TiO₂/Ag/TiO₂ multilayer thin films. Tin dioxide based multilayer SnO₂/Ag/SnO₂ films studied by Yu et al. [28] as a function of Ag thickness varied between 0 and 9 nm showed an average transmittance of >83%, sheet resistance of 9.61 Ω / sq and highest figure of merit $(1.6 \times 10^{-2} \Omega^{-1})$ for the SnO₂ (25 nm)/Ag (5 nm)/SnO₂ (25 nm) tri-layer film.

Ion doping is a unique method to restore the vacant oxygen sites in a TiO₂ lattice, and N-ion was taken as a particular choice [29]. It was reported that ion-doped TiO₂ causes band bending due to an alteration in electron concentration via charge transfer between a dopant and conduction band or valence band [30]. In N-doped TiO₂, new energy states are introduced above the valence band of the TiO₂, by increasing the electron density [31]. According to Asahi et al. [29] doping of N into TiO₂ contributes to band gap narrowing to provide visible-light response. Many research groups [29,32,33] have analyzed the N 1s XPS spectra and found that the peak at 396 was assigned to atomic β -N and substitutional N doping. Peaks at 400 and 402 eV are attributed to molecularly chemisorbed (γ -N₂) and organic impurities [34]. Jangheum et al. [35] reported smoothing of Cu film on N doping as indicated by surface morphological study using AFM & SEM.

In this work, we have fabricated $TiO_2/Ag/TiO_2$ multilayer films by sputtering and the pristine films were implanted by the N⁺ 40 keV ions of various fluence. The thickness of upper and lower TiO_2 layer was kept constant at ~45 nm, and that of embedded Ag layer was taken ~9 nm to form a continuous layer, necessary for good electrical conduction.

2. Experimental details

2.1. Sample preparation

TiO₂ powder (Alfa-Aesar, 99.99% purity) was used for target preparation. This powder was pressed into 2-in. pellet and then sintered at 1100 °C for 10 h to get better structural stability for film deposition. Commercial Ag target (99.99% purity) was used for Ag layer deposition. Glass substrates of $1 \times 1 \text{ cm}^2$ were cleaned using standard cleaning processes. Before deposition, TiO₂ and Ag targets were pre-sputtered for 15 min to remove contaminants. RF source for oxide layers and DC source for metal layer deposition were used in TAT structure preparation using sputtering method at 100 W and 16 W with 15 sccm Ar flow, respectively. The TiO₂/Ag/TiO₂ (45 ± 5 nm/9 ± 2 nm/45 ± 5 nm) structure was sequentially prepared at room temperature on glass substrates. The base pressure of the chamber was ~5×10⁻⁶ mbar and working pressure during TiO₂ deposition was $\sim 6 \times 10^{-3}$ mbar and $\sim 1.6 \times 10^{-2}$ mbar for Ag layer. The as-deposited TAT films were implanted with 40 keV N⁺ ions with fluences of 1×10¹⁴,

 5×10^{14} , 1×10^{15} , 5×10^{15} and 1×10^{16} ions/cm² at room temperature using the low energy ion beam facility at the Inter-University Accelerator Centre (IUAC), New Delhi, India. The nuclear energy loss and electronic energy loss values for 40 keV N⁺ ions were obtained as 20.13 and 13.3 eV/Å respectively; as calculated by the SRIM-2010 software. These values show that entire passage of these ions in the film is dominated by nuclear energy loss. The stopping range of N ions was 60 nm, which is less than the film thickness (100 nm). RBS was performed using 1.7 MV tandem accelerator facility with 2 MeV He⁺ ions at scattering angle of 165° at IUAC, New Delhi for depth profiling and compositional studies.

2.2. Characterization

The crystal structure of the TAT multilayer was investigated by using an X-Ray Diffractometer (Panalytical X-Pert Pro) with CuKa radiation. The morphology and topography of the films were imaged from Nova Nano FE-SEM 450 (FEI) and atomic force microscopy done by Multimode Scanning Probe Microscope (Bruker). Optical transmittance and absorbance were measured using LAMBDA 750 (Perkin Elmer) UV-vis-NIR Spectrophotometer. Carrier concentration, carrier mobility and resistivity of TAT multilayer films were measured by HMS3000, Ecopia using a magnetic field of 0.57 T. Sheet resistance was evaluated using the four-point-probe technique. Composition, valence and chemical states of TAT films were analyzed by the X-ray photoelectron spectroscopy (XPS, Omicron ESCA). For XPS measurement monochromatic Al K_{α} (1486.7 eV) source with a mean radius of 124 mm was used and monochromatic X-ray resolution as confirmed by FWHM was 0.6 eV. The vacuum pressure of the chamber was $\sim 4 \times 10^{-10}$ mbar.

3. Results and discussion

3.1. Structural properties

The X-ray diffraction (XRD) pattern of the pristine and implanted $TiO_2/Ag/TiO_2$ multilayer films for different fluence is shown in Fig. 2. The absence of any peak for TiO_2 shows its amorphous behavior which is expected because the deposition was done at room temperature. Peaks at 20=38.15°, 44.3°, 64.6° and 77.5°, are assigned to the (111), (200), (220) and (311) planes of Ag (JCPDS No. 04-0783) respectively. After implantation, the intensity of the Ag (111) peak slightly increased with the ion fluences and the full width at half maximum FWHM (β) changed from 0.6605 to 0.5188 and crystallite size increased from 13.3 nm to 16.8 nm as the fluence changed from 0 to 1×10^{16} ions/cm². Residual stress and lattice constant calculated for pristine and N ion implanted films are shown in Table 1.



Fig. 1. Schematic showing the fabrication of nitrogen ion implanted TAT multilayer film.

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