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# Synthesis and Structure of Transparent Zinc-Niobate-Tellurite Glasses with Low Hydroxyl Content

Masato Shimoda<sup>1</sup>, Masayuki Uchida<sup>1</sup>, Tomokatsu Hayakawa<sup>1\*</sup>, Philippe Thomas<sup>2</sup>

<sup>1</sup>Field of Advanced Energy Conversion, Department of Frontier Materials, Nagoya Institute of Technology, Gokiso, Showa, Nagoya 466-8555, Japan

<sup>2</sup>Science des Procédés Céramiques et de Traitements de Surface (SPCTS), UMR CNRS 7315, Faculté des Sciences, Université de Limoges, Centre Européen de la Céramique, 12 rue Atlantis 87068 Limoges Cedex, France

\*Corresponding author: E-mail address: hayatomo@nitech.ac.jp

## ABSTRACT

Low-hydroxyl (OH)-content, transparent zinc-niobate-tellurite glasses ( $\text{TeO}_2\text{-Nb}_2\text{O}_5\text{-ZnO}$  and  $\text{TeO}_2\text{-Nb}_2\text{O}_5\text{-ZnCl}_2$ ) were synthesized by using dry-air atmospheres, stirring of the melt, and preheat treatment before melt preparation. A low OH level ( $79.8 \pm 6.3$  ppm) was achieved by performing a preheat treatment of mixed raw chemicals with zinc chloride and then melting in both open- and dry-air atmospheres with several counts of intermittent stirring. However, this sequence of processes resulted mainly in non-crystallized, opaque products. The basis of this opacity was investigated by performing Raman spectroscopy, X-ray diffraction (XRD), and X-ray fluorescence (XRF) measurements on both the opaque and transparent glasses. Raman spectroscopy revealed that the proportion of Te-O-Te bonds, relative to tellurite unit structures ( $\text{TeO}_4$ ,  $\text{TeO}_{3+1}$ , and  $\text{TeO}_3$ ), in the opaque glasses decreased with increasing isolated structures of  $\text{TeO}_3^{2-}$ . This decrease resulted possibly from atomic-scaled inhomogeneity and immiscible domain formation in the melts and corresponding glasses, which have less amounts of residual OH

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