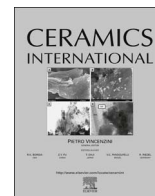




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# Improved thermoelectric performances in textured $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y/\text{Ag}$ composites

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## ABSTRACT

$\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y$  thermoelectric ceramics with small Ag additions (0, 1, 3, and 5 wt%) have been textured using the laser floating zone method. Microstructure has shown a slight decrease on the secondary phases content and a better grain alignment in Ag added samples. These microstructural features are reflected in the thermoelectric properties, which have shown a significant decrease of electrical resistivity, when the Ag content is raised. In spite of a corresponding decrease of Seebeck coefficient, all the Ag-containing samples possess higher Power Factor values than the  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y$  ones. Moreover, the maximum Power Factor values (about  $0.36 \text{ mW/K}^2\cdot\text{m}$  at  $650^\circ\text{C}$ ) have been measured in  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y + 3 \text{ wt\% Ag}$  samples, which are the best results reported in this family of materials.

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## 1. Introduction

Due to the present and future energetic challenges, thermoelectric (TE) energy conversion can be exploited for harvesting wasted heat in the classical energy transformation systems, such as automobile, thermal power plants, etc. The use of TE systems can raise the efficiency of these systems, decreasing the fossil fuels consumption and, consequently, the release of greenhouse gases. On the other hand, for these applications TE materials with high energy conversion efficiency, quantified through the dimensionless Figure-of-Merit,  $ZT (= TS^2/\rho\kappa)$ , where  $S$ ,  $\rho$ ,  $\kappa$ , and  $T$ , are Seebeck coefficient, electrical resistivity, thermal conductivity, and absolute temperature [1], are necessary. Nowadays, different metallic alloys possess high performances ( $ZT > 1$ ) [2] which can make them attractive for these practical applications. However, the most common, as  $\text{PbTe}$  [2] or  $\text{Bi}_2\text{Te}_3$  [3] are based on heavy and/or toxic elements which can be oxidized or released when working at high temperatures.

In 1997, a broad research field has been opened by the discovery of relatively high thermoelectric properties in  $\text{Na}_x\text{CoO}_2$  ceramic material ( $ZT \sim 0.26$  at  $300 \text{ K}$ ) [4]. Since then, many works have been performed on CoO-based materials with attractive thermoelectric performances [5–9]. Structural studies of these compounds have shown that their crystalline structure can be

described as two different alternately stacked layers: a common conductive  $\text{CdI}_2$ -type  $\text{CoO}_2$  layer with a two-dimensional triangular lattice, and a block layer, composed of insulating rock-salt-type (RS) layers. Both sublattices (RS block and  $\text{CdI}_2$ -type  $\text{CoO}_2$  layer) possess common  $a$ - and  $c$ -axis lattice parameters and  $\beta$  angles but different  $b$ -axis length, causing a misfit along the  $b$ -direction [10,11].

The coexistence of these two sublattices produces a strong crystallographical anisotropy and anisotropic properties. As a consequence, the grains alignment is necessary when trying to achieve bulk properties close to the determined on single crystals. Among the successful techniques previously used in oxide ceramics, it is worth to mention the template grain growth (TTG) [11], sinter-forging [12], spark plasma [13], laser floating zone (LFZ) [14], or the electrically assisted LFZ [15]. On the other hand, cationic substitution has been found to be beneficial for the thermoelectric performances in previous works [16,17], leading to improved electrical behaviour. Moreover, metallic Ag additions have also shown to enhance the thermoelectric properties of these CoO-based compounds without any evident reaction with the thermoelectric phases [18].

Taking into account previous results on textured [19], Pb doped [20], and Ag added [18]  $\text{Bi}_2\text{Ba}_2\text{Co}_2\text{O}_x$  thermoelectric materials, the aim of this work is producing high performance TE materials by the simultaneous Ag addition to the optimally Pb doped  $\text{Bi}_2\text{Ba}_2\text{Co}_2\text{O}_x$  compound, followed by an adequate texturing process through the well established LFZ technique.

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## 2. Experimental

The initial  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y + x \text{ wt\% Ag}$  (with  $x=0, 1, 3,$  and  $5$ ) polycrystalline ceramics were prepared from commercial  $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$  ( $\geq 98\%$ , Aldrich),  $\text{PbO}$  (Aldrich, 99%),  $\text{BaCO}_3$  (98.5%, Panreac),  $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$  (98%, Panreac), and metallic Ag (99%, Aldrich) powders by a sol-gel via nitrates method. They were weighed in the appropriate proportions and dissolved in a mixture of distilled water and concentrated  $\text{HNO}_3$  (analysis grade, Panreac). Once a clear pink solution has been obtained, citric acid (99.5%, Panreac), and ethylene glycol (99%, Panreac), were added to this solution in the adequate proportions. Evaporation of the solvent was slowly performed to evaporate water and decompose the nitric acid excess, increasing the pH values. The raise of pH is produced until the polymerization reaction between ethylene glycol and citric acid is started, leading to a solid pink gel [21,22] where all the cations are coordinated. Further heating at 350–400 °C produced the organic material decomposition through a slow self combustion. The remaining powder was mechanically ground and calcined at 700 and 750 °C for 12 h, with an intermediate manual milling. This step is very important for samples textured by LFZ technique as, if barium carbonates are not totally decomposed prior texturing, they would decompose in the molten zone, leading to the solidification front destabilization. Afterwards, the resulting powders were then cold isostatically pressed in form of cylinders ( $\sim 120$  mm long and 2–3 mm diameter), at 200 MPa for 1 min, which were used as feed in a LFZ system successfully applied in previous works [23]. The growth rate (30 mm/h) has been chosen to be attractive for practical applications on the basis of previous results in this system [19,24]. After the texturing process, long (more than 150 mm) cylindrical and geometrically homogeneous samples have been obtained. The textured materials were then cut into shorter pieces with the adequate sizes for their electrical characterization ( $\sim 15$  mm long).

The structural identification of all the samples was performed by powder XRD utilizing a Rigaku D/max-B X-ray powder diffractometer (Cu  $K\alpha$  radiation) with  $2\theta$  ranging between 10 and 70 degrees. Microstructural observations were performed on polished samples using a Field Emission Scanning Electron Microscope (FESEM, Carl Zeiss Merlin) fitted with energy dispersive spectrometry (EDS) analysis.

Steady-state simultaneous measurements of resistivity and Seebeck coefficient were determined by the standard dc four-probe technique in a LSR-3 apparatus (Linseis GmbH) between 50 and 650 °C. From these data, PF values, as a function of temperature, were calculated in order to evaluate the thermoelectric samples performances.

## 3. Results and discussion

Powder XRD patterns for all  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y$  samples with different amounts of Ag are plotted (from 10 to 40 degrees for clarity) in Fig. 1. They show very similar patterns where the most intense peaks correspond to the misfit cobaltite  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y$  phase, in agreement with previous reported data [25]. From the graph, it is clear that the cobaltite phase appears as the major one, independently of Ag content. Moreover, the peaks appearing at around 20.5 and 28.5 degrees, marked by a \*, indicate the formation of  $\text{Bi}(\text{Pb})\text{BaO}_3$  secondary phase with  $Fm\bar{3}m$  space group (#225) [26]. Furthermore, in Ag added samples a new peak appears at around 38 degrees, associated to metallic Ag (111) plane (indicated by #) [27]. This peak indicates that at least some of Ag does not react with the thermoelectric ceramic, leading to the formation of a ceramic-Ag composite with metallic particles distributed inside the matrix, as observed in similar ceramic

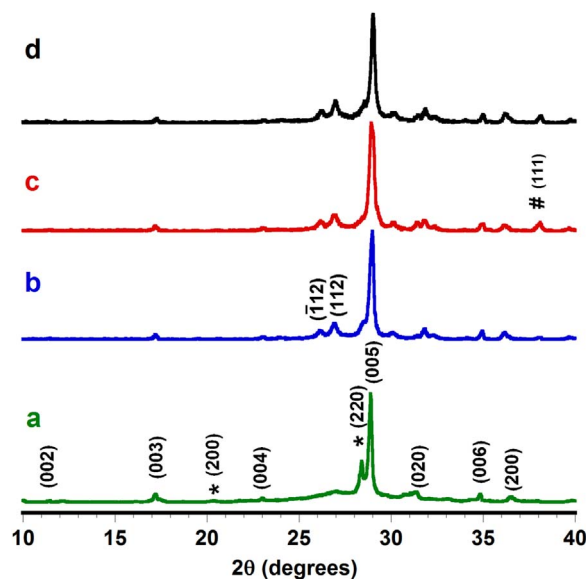


Fig. 1. Powder XRD diagrams for the  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y$  textured samples with a) 0; b) 1; c) 3; and d) 5 wt% Ag. Crystallographic planes have been indicated on the peaks corresponding to the  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y$  phase. Other phases are indicated by symbols: \* Co-free  $\text{BiBaO}_3$  secondary phase; and # Ag.

systems [28]. Moreover, from these data it is clear that the amount of secondary phases decrease with Ag addition, illustrated by the drastic decrease of the (220) peak of  $(\text{Bi}, \text{Pb})\text{BaO}_3$  phase.

SEM micrographs performed on polished longitudinal sections of all samples are shown in Fig. 2. In this figure, it can be clearly seen that all samples are composed by three main contrasts, associated through EDS to different phases. Grey contrast (#1) corresponds to the  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y$  phase, as the major one, in agreement with the XRD data. Dark and light grey ones (#2, and #3, respectively), have been associated to  $\text{Bi}_5\text{Co}_5\text{O}_{14}$ , and  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{BaO}_3$  secondary phases, respectively. Moreover, in the Ag added samples, a new contrast has been identified, very similar to the thermoelectric phase, as a grey one (#4 in Fig. 2d), corresponding to metallic Ag. When considering the microstructural modifications produced by Ag addition it can be observed that secondary phases content is decreased, in agreement with the XRD observations. Furthermore, the grains orientation seems to be improved with Ag addition until 3 wt%, slightly decreasing for higher Ag proportion.

The temperature dependence of the electrical resistivity as a function of the Ag content is presented in Fig. 3. As it can be easily seen, the  $\rho(T)$  curves are very similar for  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y$  samples with 0 and 1 wt% Ag, with semiconducting-like behaviour ( $d\rho/dT < 0$ ) in the whole measured temperature range, while for higher Ag addition, a metallic-like one ( $d\rho/dT > 0$ ) is found between room temperature and  $\sim 200$  °C, changing to a slight semiconducting one at higher temperatures. Moreover, the electrical resistivity values are clearly decreased when the Ag content is increased, in the whole measured temperature range, reflecting the raise in the grain alignment and the decrease of the secondary phases content observed in the SEM micrographs. Furthermore, the lower electrical resistivity can be also reflecting an improved electrical grains connectivity promoted by Ag addition, in agreement with the microstructural observations and previously reported data [23,29], which demonstrated the formation of a ceramic-Ag composite. The minimum resistivity value at room temperature,  $\sim 6.5 \text{ m}\Omega \text{ cm}$ , is obtained for the  $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Ba}_2\text{Co}_2\text{O}_y + 5 \text{ wt\% Ag}$  samples, which is much lower than the best reported for sintered  $\text{Bi}_2\text{Ba}_2\text{Co}_2\text{O}_8$  materials (pure and Ag-doped), typically between 550 and 21  $\text{m}\Omega \text{ cm}$  [25,30–35].

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