

On the interfacial reactions between VO₂ and thin metal films



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H I G H L I G H T S

- Competitive synthesis of VO₂(M1) phase that undergo MIT transition.
- Thin-film metal-VO₂ couples were annealed for 1 h up to 900 °C.
- Solid-state interfacial reactions between VO₂ and metals Co, Hf, Ni, Pd, Au and Pt.

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We have investigated solid-state interfacial reactions between VO₂ and the metals Co, Hf, Ni, Pd, Pt and Au to establish possible criteria for whether a metal will react with VO₂ or not. Thin-film metal-VO₂ couples were annealed for periods ranging from 45 to 60 min at temperatures in the range 400–900 °C, and characterized using Rutherford backscattering spectrometry and X-ray diffraction. No interfacial reactions were detected between VO₂ and Co, Ni, Pd, Au and Pt, but Hf did react. Metals with a positive heat of reaction ΔH and electronegativity ϕ^* , as defined by Miedema, greater than 4.9 V did not react with VO₂ whereas Hf with $\phi^* < 4.9$ V and a negative ΔH did. This correlation between the heats of reaction and the Miedema electronegativity of metals offers an empirical criterion for predicting whether a metal will react with VO₂ or not.

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1. Introduction

Some vanadium oxides, such as V₂O₃, V₂O₅, V₆O₁₃, undergo a first-order phase transition at specific temperatures. This transition is accompanied by changes in their electrical, magnetic and optical properties [1–3]. VO₂ is the most-studied of these phases because its transition temperature (68 °C) is close to room temperature [1,4,5]. VO₂ behaves like a semi-metal above its transition temperature, reflecting a wide range of solar wavelengths, whereas below 68 °C it tends to reflect significantly less in the near infrared [5]. As a result, it has been suggested that this particular oxide can be used as a smart window coating [6]. The transition temperature of VO₂ can be increased or reduced by doping with metals [7,8]. Doping with W, Nb and Mo lowers the transition temperature, whereas doping with Ti and Sn raises it [8,9]. Zhou et al. [10] used

Au (200 nm) and very thin Ti (20 nm) layer to fabricate and study electrically triggered ultrafast phase transition in VO₂ device switches with metal/VO₂/metal structure making this work valuable.

Some metals may react with VO₂ while others do not. Although work has been done on testing and predicting interfacial reactions between SiO₂ and various metals [11,12], we found no literature on the interfacial reactions between VO₂ and metals. In contrast, the Miedema electronegativity parameter ϕ^* indicates a correlation between a reactivity of SiO₂ and some metals and their calculated heats of reaction [13]. We therefore investigated the ϕ^* value of a metal and the calculated heat of reaction (ΔH_R), that results when a thin film metal is placed in contact with VO₂, in order to formulate a simple rule to predict whether the value of ϕ^* could be used to predict interfacial reactions between VO₂ and the metal.

2. Experimental procedure

Glass, chemically cleaned in an ultrasonic bath and then rinsed in de-ionized water, was used as the substrate for the prepared

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samples. The thin films of VO₂ were then deposited on the glass substrates by means of an rf-inverted cylindrical magnetron sputtering system whose details are given elsewhere [2]. VO₂ was obtained by sputtering vanadium onto glass in atmosphere (10⁻² mbar) containing oxygen. The deposited VO₂ film was made sufficiently thick that should the metal and VO₂ react, there would be enough VO₂ left on the substrate. The deposition time of the VO₂ (M1) phase films was 2–3 h and the substrate temperature was kept at 450 °C through-out the deposition. X-ray diffraction (XRD) was carried out on the samples to ensure that VO₂ had indeed been deposited on the glass substrate. The structure and morphology of the deposited films were further analyzed by grazing X-ray diffraction (GXR) and Atomic Force Microscope (AFM), respectively, using a Bruker D8 Advance X-ray diffractometer using a Cu K_α (λ = 0.154 nm) radiation source. Data were recorded 2θ = 10–70° range. An optical transmission spectrophotometry conducted on a Cecil 2000 unit in the spectral range 200–1100 nm was used to extract the hysteresis of the VO₂ (M1). The films were heated from 25 °C to 95 °C in the steps of 5 °C per measurement and subsequently cooled back.

The metals Co, Hf, Ni, Pd and Pt were then deposited individually on top of the VO₂ film using an electron beam evaporator in vacuum of better than 4 × 10⁻⁸ kPa, forming a composite sandwich structure glass/VO₂/metal. Annealing was conducted in vacuum of better than 2 × 10⁻⁸ kPa for times ranging from 45 to 60 min. The interfaces were analyzed using Rutherford backscattering spectrometry (RBS) and XRD.

3. Results and discussion

To find out whether we had managed to synthesize VO₂ (and not one of the many other oxides) we used XRD. The existence of VO₂ film was determined by using XRD. The XRD diffraction pattern of the deposited VO₂ films can be indexed according to the monoclinic phase, which is compatible with the standard value *a* = 5.7430 Å, *b* = 4.5170 Å, *c* = 5.3750 Å (JCPDS No. 72-514). Fig. 1 shows a pattern of the film showing a polycrystalline phases of VO₂ deposited on glass substrate. The first strong peak appearing at 2θ = 27 °C corresponds to an oxide of vanadium phase, VO₂ (M1). Other peaks falling between 27 °C and 65 °C show that VO₂ is polycrystalline.

The transmittance of the film is measured as function of temperature using UV–Vis spectrometer technique to further confirm the presence of the right VO₂ phase, the M1 phase. Fig. 2 shows an atomic microscopy image of VO₂ film showing the growth of the

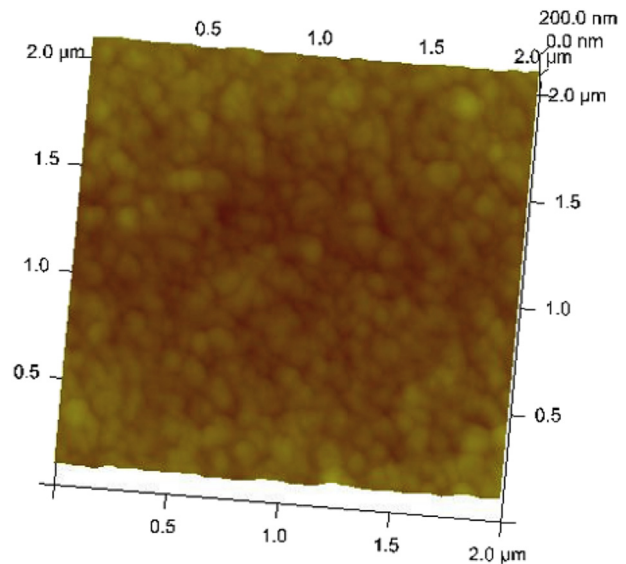


Fig. 1. X-Ray diffraction pattern of VO₂ deposited on a corning glass substrate.

Fig. 2. X-Ray diffraction pattern of VO₂ deposited on a corning glass substrate.

film making small compact particles. Using Nanoscope software, the root mean square roughness (*R_q*) was calculated and found to be 14.3 nm.

Fig. 3 shows the experimental transmittance curves of VO₂ film at different temperatures, when the film temperature is raised from 25 °C (RT) to 95 °C and subsequently reduced back to 25 °C (RT) in small steps of 5 °C. From Fig. 3, we observe a change in the transmittance as function of wavelength. The spectrum transmittance of the low-temperature semiconducting phase and high-temperature metallic phase are sharply contrasting in the infrared region, furthermore this VO₂ film exhibited thermo-chromism. The transmittance of the VO₂ film on glass was observed as function of temperature as shown in Fig. 4.

The heating was done at the steps of 5 °C, and its transmittance at various temperatures was measured at a wavelength of 1100 nm. The transition temperature of the film was in a good range reported [13,14], it should be between 60 °C to 70 °C depending on number of thing like thickness nano-structuring etc. The semiconductor-to-metal transition of the deposited film was investigated using transmittance measurements as a function of temperature. The transmittance of the film shows a decrease with an increase in

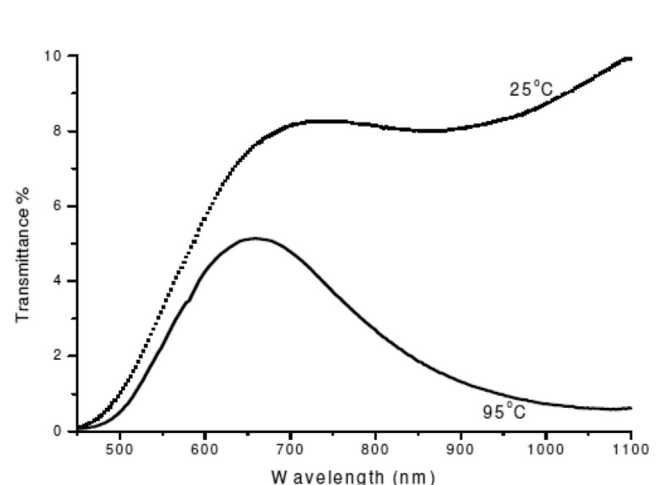


Fig. 3. Transmittance (%) vs. temperature (°C) for VO₂ film by reactive ion sputtering.

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