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The pH-dependent corrosion behavior of ternary oxide semiconductors and common metals and its application for solution-processed oxide thin film transistors circuit integration

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Abstract. Individual oxide semiconductors and metals experience unique pH-dependent phase-transition into natively stable phases (metal, metal ions, or oxide phases) in various acidic and basic solutions. Thus, the corrosion behavior of oxide semiconductors and metals can be engineered by controlling pH values. In particular, the specific pH value induced interesting corrosion behaviour that oxide semiconductor becomes chemically-stable and metal solely experiences active ionization. First, the pH-dependent corrosion behavior of ternary oxide semiconductors [ZnSnO (ZTO) and InZnO (IZO)] and common metals (Mo and Mo/Cu) was explored based on theoretical Pourbaix diagram and experimental corrosion data. Next, the pH-dependent corrosion behavior based back-channel wet-etch (BCWE) process using pH-controlled wet etchants was designed and applied for chemical damage-,

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