



Structural and piezo-ferroelectric properties of $\text{K}_{0.5}\text{Na}_{0.5}\text{NbO}_3$ thin films grown by pulsed laser deposition and tested as sensors

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ABSTRACT

We have grown $\text{K}_{0.5}\text{Na}_{0.5}\text{NbO}_3$ (KNN) films on $\text{Pt}/\text{TiO}_2/\text{SiO}_2/\text{Si}$ substrates by using pulsed laser deposition. Thin films with pure KNN crystal phase and homogenous morphology were successfully prepared by optimizing the deposition conditions, mainly the oxygen partial pressure and deposition temperature. For oxygen pressure as low as 8 Pa a tungsten-bronze phase is formed, while for higher values i.e. 16 and 32 Pa, the desired KNN perovskite structure is obtained. To prepare pure KNN thin films, a deposition temperature as low as 675 °C is required, however, increasing the temperature to 725 °C promotes better homogeneity and densification of the films. The piezoelectric and ferroelectric behavior, confirmed by piezoresponse force microscopy, revealed a local piezoelectric coefficient of $d_{33} = 37 \text{ pm/V}$ and a coercive voltage of 3.5 V. Finally, we measured the frequency response of this material prepared as a sensor, using the photoacoustic method. This measurement shows that the resonant frequency of a KNN thin film of 230 nm thickness is centered at 3 MHz.

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1. Introduction

Ferroelectric materials are widely used for many applications, such as actuators, sensors and buzzers [1]. Until now, the most used compound for these applications is $\text{Pb}(\text{Ti,Zr})\text{O}_3$ (PZT), which contains around 60 Wt% of lead, which is a toxic element. Currently, many workers are developing the lead-free ferroelectrics needed to replace PZT. One of the most promising lead-free ceramics is $\text{K}_{0.5}\text{Na}_{0.5}\text{NbO}_3$ (KNN) [2–4], because of its high Curie temperature and remnant polarization [4–5]. KNN films have been prepared by different methods, including sol-gel, pulsed laser deposition (PLD) and sputtering [6–14]. By the Sol-gel method, high purity KNN films can be obtained, but it requires expensive and unstable reagents (niobium alkoxide) [6–7]. In addition, an excess of K^+ and Na^+ is needed to reduce departure from stoichiometry [8–9]. Thin films prepared by this technique show piezoelectric constants (d_{33}) ranging from 17 to 74 pm/V, depending on the excess of alkaline elements added [6–7]. On the other hand, sputtering and PLD are easier and demand inexpensive reagents [10–11,13]. Using sputtering, and different substrates, local d_{33} coefficients of 26 to 58 pm/V were measured [12] while, for PLD, values of 49 pm/V for $\text{Li}^{+1}\text{-Ta}^{+5}$ co-doped KNN have been reported [10]. However, in all of these cases, ferroelectric hysteresis loops of films with thicknesses

under 250 nm have not been found, perhaps due to the large number of defects, mostly caused by the K^+ and Na^+ losses. In this work, it is shown that by optimizing the deposition parameters in the PLD process, pure KNN films with less than 200 nm thicknesses can be prepared, with well-defined piezo-ferroelectric response determined by piezoresponse force microscopy. We show that these films can be used as sensors.

2. Experimental

The ceramic target used for film deposition was obtained by the solid state reaction of stoichiometric amounts of K_2CO_3 , Na_2CO_3 and Nb_2O_5 . The reagents, mixed using an agate mortar and acetone, were calcined at 850 °C for 1 h. Then, the powders were compacted into a pellet under 300 MPa using a uniaxial press. Once the pellet was sintered at 1060 °C for 2 h, it was used to grow thin films on $\text{Pt}/\text{TiO}_2/\text{SiO}_2/\text{Si}$ substrates. During film deposition, the fluence of the laser pulse and target to substrate distance were kept constant at 3 mJ/cm² and 5 cm, respectively. The substrate temperature was varied between 650 °C and 800 °C. The oxygen partial pressure was changed at 8, 16 and 32 Pa. The repetition rate was set at 5 or 10 Hz. Table 1 shows the growth conditions for the most representative samples. In order to confirm the target and film crystal structure, X-ray diffraction (XRD) patterns were collected in a Bruker D8 diffractometer, using $\text{Cu-K}\alpha_1$ (1.5406 Å) radiation.

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Table 1
Growth conditions for KNN films.

Sample	Oxygen pressure (Pa)	Temp. (°C)
KNN1	8	700
KNN2	16	700
KNN3	16	675
KNN4	16	750
KNN5	16	800
KNN6	32	700
KNN7	32	725

For crystal size and morphology observations, a JEOL 7600F field emission scanning electron microscope (SEM) was employed. Piezoelectric properties, ferroelectric domains and local phase ferroelectric hysteresis loops were determined by piezoresponse force microscopy (PFM) and contact resonance (CR)-PFM, in a Dimension 3100 Nanoscope IV scanning probe microscope (SPM) System (Veeco). Platinum tips, 25Pt300B (Rocky Mountain Nanotechnology) and an *ac* bias of 10 Vpp were used.

3. Results and discussion

The synthesis of KNN ceramics has been demonstrated to be difficult due to potassium and sodium volatility at high temperature [4,15]. As a result, the presence of secondary phases, which strongly diminish the ferroelectric and piezoelectric properties, is common. Our ceramic target, however, does not show any secondary phases and only a pure perovskite KNN is obtained (Fig. 1).

The analysis of the films obtained is described below, starting with one representative sample i.e. the one grown at 16 Pa oxygen pressure, 800 °C and 10 Hz (marked as KNN5 in Table 1), followed by a description of the effects of changes in the deposition parameters. The KNN sample shows a pure orthorhombic KNN crystal phase, as can be seen in the top XRD pattern of Fig. 1, which agree with the data reported in literature [8,16–17]. The lattice parameters extracted from the XRD for the ceramic target are $a = 4.0192$, $b = 3.9628$ and $c = 4.0179$ Å, respectively. While for the KNN5 film are $a = 3.9767$ and $b = 3.9755$ Å, respectively. Furthermore, (100) preferential orientation is observed, this is due to the influence of the substrate, mainly the influence of Pt that also induces a slightly change in a lattice parameter, as reported before [17–18] and clearly evidenced in the values mentioned before. The morphology is composed of a loosely packed quasi-columnar growth over a denser and homogenous layer as can be appreciated in Fig. 2.

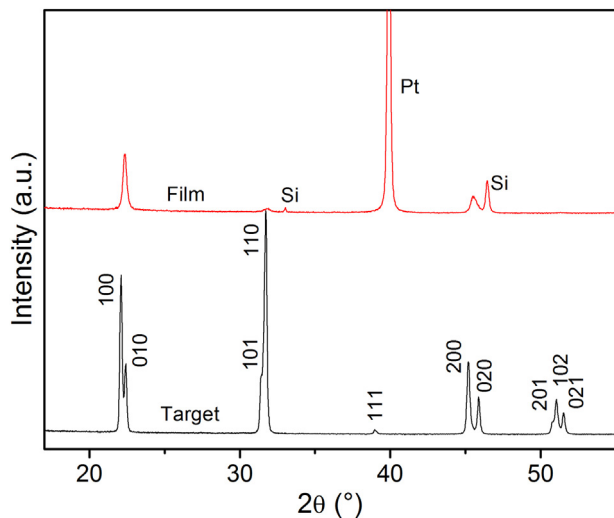


Fig. 1. XRD patterns of the ceramic target sintered at 1060 °C, and KNN5 film prepared at 800 °C, 10 Hz and 16 Pa of oxygen pressure.

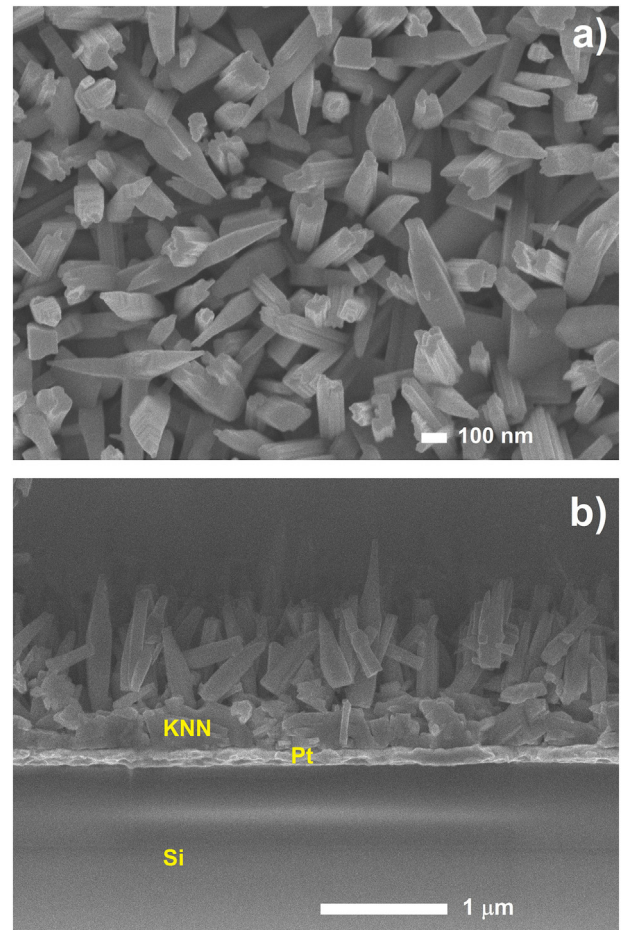


Fig. 2. (a) Top and (b) transversal SEM images of KNN5 thin films grown at 800 °C, 10 Hz and 16 Pa of oxygen pressure.

In order to promote densification and homogeneity in the films, the deposition parameters were changed. Reducing the deposition temperature, while maintaining all the other parameters constant, the thin films gradually transforms the columnar growth observed at 800 °C, into to a completely cubic-like morphology at 675 °C, with a wide grain size distribution characteristic of these compounds, as can be seen in Fig. 3. Moreover, the crystal average size for KNN2 and KNN3 (Fig. 3c and d) are 132 and 86 nm, respectively. This indicates that reducing the deposition temperature it was obtained smaller crystal and size distribution. Fig. 4 shows how the oxygen partial pressure affects the crystal phase of the films. For the lower pressure used (8 Pa KNN1 in Table 1), tungsten-bronze peaks of $K_6Nb_{10.8}O_{30}$ are observed, perhaps because the available oxygen is not enough to produce the perovskite structure. Increasing the oxygen pressure to 16 and 32 Pa generates the pure orthorhombic KNN. Other important SEM images of films prepared at different oxygen partial pressures are shown in Fig. 5. As can be seen, there is no a significant variation in crystal size and morphology, even for the sample with a prevalent tungsten-bronze phase, the one prepared at 8 Pa.

In agreement with the above analysis, although a pure KNN perovskite crystal structure is obtained using 16 and 32 Pa, these higher oxygen pressures may have a more important effect: avoiding oxygen vacancy formation and consequently, reducing dielectric losses. Furthermore, to prevent the columnar, loosely-packed morphology observed in films grown at high temperatures, i.e. at 800 °C, but at the same time ensuring pure KNN crystal phase formation, 725 °C was chosen as the optimal temperature and the laser frequency was lowered from 10 Hz to 5 Hz.

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