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Thermo-chemo-mechanical effective properties for homogeneous and heterogeneous *n*-phase mixtures with application to curing

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Abstract

Our work presents an extension of a composite sphere model to temperature-dependent elastic effects accompanied by curing. Homogenization of a representative unit cell (micro-RVE) on the heterogeneous microscale which accounts for thermo-chemo-mechanical coupling with linear elasticity yields volumetric effective properties. Two conceptions are considered: Firstly, a homogeneous mixture with n-phases is assumed. Secondly, a geometric arrangement on the microscale is represented by the n-layered composite sphere model. In a numerical study for a 3-phase matrix it is demonstrated that the effective properties lie within certain bounds.

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1. Introduction

Polymeric materials are broadly applied in carbon- and glass fibre-reinforced composites (FRP), epoxy laminates and particle-reinforced polymer structures, cf. [1], [2]. Two important production processes of FRPs are resin transfer molding (RTM) and reaction injection molding (RIM). In both processes a fibre preform or dry fibre reinforcement is packed into a mold cavity. After closing, a resin or resin system as mixture of reactants, e.g. resin and curing agent, is pumped into the mold under pressure until the mold is filled. Subsequently the curing cycle starts.

A reliable and predictive simulation of the production process requires the thermo-chemomechanical effective material properties depending on curing. For this purpose homogeneous and heterogeneous conceptions for the matrix can be distinguished as follows:

• Homogeneous mixture: An equally distributed

mixture is assumed for all constituents, resin, curing agent and solidified material. Concerning the cure dependence of effective properties, several ad hoc assumptions are made in the literature. At least two approaches can be distinguished for the effective properties, e.g. for the compression modulus: According to [3] a linear relationship is assumed for the compression moduli of the monomer (or uncured resin and curing agent) and the solid which essentially represents an upper Voigt bound. Contrary, in [4] a linear relationship is derived for the effective compressibility which essentially represents a lower Reuss bound for the effective compression modulus. It is well known, that the effective properties obtained for a homogeneous mixture are only dependent on phase fractions. These are bounds for more advanced approaches, where a geometrical arrangement of the heterogeneous microscale is taken into account.

• Heterogeneous mixture: A geometrical arrangement of a heterogeneous microstructure is introduced in [6] in a so called *composite sphere*

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model. To solve for effective properties as exact, analytical solutions for a 2-layered composite sphere model, a 2-layered inclusion is compared to an equivalent homogeneous sphere with identical boundary conditions, see [7]. Concerning the cure dependence of the effective properties, in [8] the 2-layered composite sphere model is extended to account for thermo-chemomechanical coupling. In [9], a mesoscopic model for temperature-dependent visco-elastic effects accompanied by curing of FRP is investigated. In comparison to Voigt and Reuss bounds, the effective properties from [8] for a 2-phase mixture are used. The authors conclude, that the eigenstrain state for a fully cured FRP is strongly dependent on the choice of effective properties. Based on the works [6], [7], micromechanical modeling for the effective elastic moduli for an *n*-layered spherical inclusion is introduced in a so called (n + 1)-phase model in [10]. Here, an infinite medium is constituted of an *n*-layered inclusion, embedded in a matrix which is denoted by phase n + 1. An extension of an *n*-layered composite sphere model to pure heat-dilatation is proposed by [11]. Furthermore, for the (n + 1)-phase model a thermo-chemo-mechanical coupling is proposed in [12], where it is also shown that the *n*-layered composite sphere model and the (n+1)phase model yield identical results.

To the authors knowledge, all three constituents, resin, curing agent and solid, occurring in the curing process (of a polymeric matrix) have not been considered so far by a 3-layered composite sphere model to derive volumetric effective properties while accounting for thermo-chemomechanical coupling. This contribution aims to close this gap. In order to become more general, the case with n spherical constituents is taken into account. To this end, two different conceptions are investigated: Firstly, an equally distributed nphase homogeneous mixture is assumed on the microscale. Secondly, an extension of the 2-layered composite sphere model in [8] an *n*-layered composite sphere is considered on the microscale as a heterogeneous mixture.

2. Composite sphere models: Overview

Figure 1 introduces two different idealizations which are used for the derivation of effective properties in the following sections: a) a homogeneous mixture and b) an *n*-layered composite sphere model as extension of [7]. Both idealizations have in common that a *spherical inclusion* is embedded in an infinite homogeneous medium which following [10] is denoted as *matrix*. Both are subjected to



Fig. 1: Two idealizations of the inclusion: a) homogeneous sphere hom, b) heterogeneous *n*-layered composite sphere het[n].

a uniform thermal loading in terms of a prescribed temperature θ and a chemical loading in terms of a degree of cure z (of a thermosetting polymer as in [8]). In addition, the inclusion is subjected to mechanical loading in terms of a prescribed pressure p. The distinguishing features of both conceptions in Figure 1 are as follows:

- a) Homogeneous mixture: Any spherical inclusion with total volume v and radius R_n is homogeneous and represents the effective behavior.
- b) *n*-layered composite sphere model: The constituents $i \in [1, n]$ with partial volumes $v^{(i)}$ (that is for n = 3 from inside to outside: solid (*sol*), curing agent (*ca*), resin (*r*)) and corresponding radii R_i assemble to a total volume *v*.

3. Homogeneous mixture: *n*-phase homogeneous matrix model

In this section, we summarize the results in [4] for weighted effective properties of a 3-phase homogeneous mixture and apply it to the matrix shown Figure 1.a which consists of phases $i \in [1, n]$. At any material point *P* within the inclusion, $dm^{(i)}[t]$ defines the time dependent mass of a constituent $i \in [1, n]$ and dm_0 is the total mass of the mixture which is conserved during the curing reaction, cf. [13]. With this quantities we define the mass fraction of each constituent $\zeta^{(i)}[t] = \frac{dm^{(i)}[t]}{dm_0} \ge 0$ which is used to formulate the mixture rule for the inverse of the bulk density ρ

1.
$$\frac{1}{\rho} = \sum_{i=1}^{n} \frac{\zeta^{(i)}}{\rho^{(i)}}$$
, where 2. $\rho^{(i)} = \rho^{(i)}[p, \theta]$. (1)

Using the assumption that all phases are equally distributed, the number of variables can be reduced by taking into account the stoichiometry of the mixture, as explained in [13]. To this end, the *degree of cure* $0 \le z[t] \le 1$ is introduced which represents the chemical loading in Figure 1.a resulting in the relation $\zeta^{(i)} = \zeta^{(i)}[z]$ for the mass fractions in Eq. (1.1). Combining this with Eq. (1.2) implies that

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