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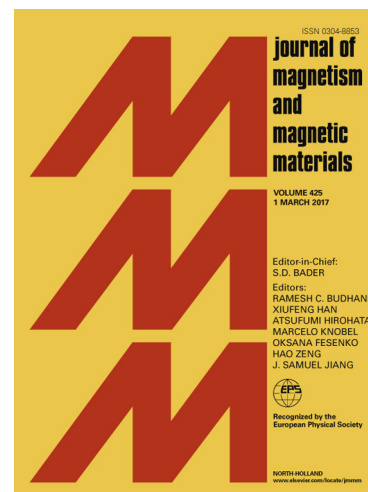
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## Effect of Y-La-Co substitution on microstructure and magnetic properties of M-type strontium hexagonal ferrites prepared by ceramic method

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### Abstract

M-type permanent magnetic ferrites  $\text{Sr}_{0.95-x}\text{Y}_{0.05}\text{La}_x\text{Fe}_{11.75}\text{Co}_{0.25}\text{O}_{19}$  ( $x=0.00, 0.10, 0.20, 0.30, 0.40$ ) have been prepared by ceramic process and investigated systematically. The phase compositions were examined by X-ray diffraction, the results showed that the single phase was obtained when  $x \leq 0.30$ . The morphology of the magnets was investigated by scanning electron microscopy (SEM), the micro-morphology of the particles exhibited the uniform plane hexagonal structure with different lanthanum content. Magnetic properties of the sample were measured by a physical property measurement system-vibrating sample magnetometer (PPMS-VSM). The saturation magnetization ( $M_s$ ) decreases constantly with the increasing of La content ( $x$ ). However, the coercivity ( $H_c$ ) increases as the increasing of doping content.

**Keywords:** Hexagonal ferrites; Micro-morphology; Magnetic properties

### 1. Introduction

Hexagonal M-type  $\text{SrFe}_{12}\text{O}_{19}$  (SrM) ferrites plays an important role in hard magnetic materials especially for its large magnetic anisotropy, high performance-to-cost ratio and good chemical stability [1-5]. In addition, they also display considerable properties in microwave devices, magneto-optics and magnetic recording media [6, 7]. Therefore, for the purpose of achieving more extensive application fields, improving the magnetic properties of M-type hexaferrites is of great significance in magnetic field. The crystal structure of SrM comprises alternative stacks of spinel (S,  $\text{Fe}_6\text{O}^{2+}_8$ ) and hexagonal (R,  $\text{MFe}_6\text{O}^{2-}_{11}$ ) blocks in the form  $\text{RSR}^*\text{S}^*$ , where \* denotes  $180^\circ$  rotation around the hexagonal c-axis [8]. The  $\text{O}^{2-}$  ions exist as close-packed blocks, with the  $\text{M}^{2+}$  ion substituting for the  $\text{O}^{2-}$  ion in R blocks. The  $\text{Fe}^{3+}$  ions

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