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- The detoxification responses, damage effects and bioaccumulation in the
 scallop *Chlamys farreri* exposed to single and mixtures of benzo[*a*]pyrene
- ³ and chrysene

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ABSTRACT

This study aimed to investigate the detoxification responses, damage effects and biotransformation in scallop 18 Chlamys farreri exposed to benzo[a]pyrene (BaP) (0.1, 1 µg/L), chrysene (CHR) (0.1, 1 µg/L) and BaP + CHR 19 $(0.1 + 0.1, 1 + 1 \mu g/L)$ for 15 days. Results demonstrated that BaP and CHR concentration (BaP < CHR) in tissues 20 increased rapidly in a time and dose effect. The mRNA expression of aryl hydrocarbon receptor (AhR), cyto- 21 chrome P450 1A1 (CYP1A1), CYP1B1, multidrug resistance protein 1 (MRP1/ABCC1), breast cancer resistance 22 protein (BCRP/ABCG2) and P-glycoprotein (P-gp) were induced especially in the mixtures of BaP and CHR. 23 Heat shock protein 90 (HSP90) and aryl hydrocarbon receptor nuclear translocator (ARNT) mRNA expression 24 was significantly elevated at days 1, 10 and 15. Detoxification enzymes of 7-ethoxyresorufin O-deethylase 25 (EROD), uridine-diphosphate-glucuronyl-transferase (UGT) and sulfotransferase (SULT) were significantly in- 26 duced and then became stable gradually while glutathione-S-transferase (GST) was inhibited in the mixtures 27 of BaP and CHR at days 10 and 15. Superoxide dismutase (SOD), glutathione peroxidase (GPx), catalase (CAT) 28 and glutathione (GSH) were all stimulated especially in the mixtures of BaP and CHR. The levels of DNA strand 29 breaks, lipid peroxidation (LPO) and protein carbonyl (PC) contents showed damage effects exposed BaP and 30 CHR. All the results indicated that BaP and CHR have similar induced effect and a majority of the biomarkers 31 pointed to a more toxic effect when BaP and CHR were mixed. These will provide a solid foundation for the 32 study of PAHs detoxification mechanism in bivalves and valuable information for marine pollution monitoring. 33 © 2016 Published by Elsevier Inc. 34

47 1. Introduction

46 45

Polycyclic aromatic hydrocarbons (PAHs), with two or more fused 48 49 aromatic rings, are mainly derived from petrogenic pollution, anthropogenic combustion, petroleum spillage and industrial and urban runoff 50(Sahu et al., 2009; Lüchmann et al., 2014), and even PAHs can be 04 generated from the direct toasting (Rey-Salgueiro et al., 2008). Due to 5253their persistent, bio-accumulative, toxic and long-range transportable properties, PAHs have drawn wide attention (Wania and Mackay, 541993; Pickering, 1999; Cachot et al., 2006; Yang et al., 2010). In 2008, 5556the sum of benzo[*a*]anthracene (BaA), benzo[*a*]pyrene (BaP), benzo[*b*]fluoranthene (BbF) and chrysene (CHR) was identified as the 57most suitable indicator for PAHs in food (EFSA, 2008). According to 5859previous studies, BaP and CHR contents in surface sediment from 60 Zhanjiang Bay, China, were up to 19.90 and 39.65 ng/g dry weight, 61correspondingly (Huang et al., 2012). Total PAH concentrations ranged 62 from 139.16 to 1717.87 ng/L in surface water, from 226.57 to

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http://dx.doi.org/10.1016/j.cbpc.2016.09.004 1532-0456/© 2016 Published by Elsevier Inc. 1404.85 ng/L dry weight in suspended particulate matter, and from 63 276.26 to 1606.89 ng/g dry weight in sediments, in the Bohai Sea, 64 China (Men et al., 2009). Dissolved and particulate PAHs were up to 65 217 and 1597 ng/L in Northwestern Mediterranean Sea (Guigue et al., 66 2011). All above studies suggested that PAHs pollution had been a 67 serious problem all over the world. Considering unpredictable interac-68 tions of environmental contaminants, studies on the toxic effect and 69 risk assessment of single and mixtures of PAHs are urgently needed. 70

As one of the largest classes of chemicals present in the environment 71 (Van der Oost et al., 2003) PAHs have been considered a health risk for 72 wild life and humans for a long time (IARC, 1973; Ariese et al., 2001). 73 Presently, growing attention has been concentrated on PAH metabolice 74 mechanism in the organism, especially on their toxicity and metabolites 75 (Barhoumi et al., 2011; Cavalieri and Rogan, 1995; Harris et al., 2009; 76 Jacques et al., 2010; Ramesh et al., 2001). The detoxification responses 77 of PAHs are a complex process that involves AhR pathway, phase I 78 (EROD et al), phase II (GST, UGT and SULT et al) xenobiotic- 79 metabolizing enzymes and phase III transporters (Paetzold et al., 80 2009; Pereira et al., 2011; Liu et al., 2014a, 2014b; Cai et al., 2016). 81 CYP1 family plays an important role in metabolism of many toxicants 82 including PAHs (Ghanayem et al., 2000; Matsumoto et al., 2007; Jönsson 83

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et al., 2010) with the catalytic activity of EROD. High induction of CYP1 84 85 transcripts by BaP was observed in rare minnow (Yuan et al., 2013). Moreover, Willett et al. (2006) reported that BaP exposure significantly 05 87 induced CYP1B gene transcription in blood, gonad and liver of laboratory catfish. However, there was limited information on the expression of 88 CYP1B1 mRNA in bivalve exposed to PAHs. The coordinated regulation 89 of xenobiotic-metabolizing enzymes with xenobiotic transporters was 90 91 also regarded as a mechanism of detoxification (Bard, 2000; Leslie 92 et al., 2005). In mammals, arsenic (Liu et al., 2001), arsenite 93 (Kauffmann et al., 2002) and BbF (Ebert et al., 2005) induced the mRNA expression levels of MRP1. Paetzold et al. (2009) reported that 94hepatic mRNA expression levels of MRP1 and BCRP were elevated in 95Tar Pond killifish in the side with higher PAHs level. Though the impor-96 97 tant role of EROD, GST, UGT and SULT were frequently studied in bivalves in PAHs metabolism, however, studies on the exclusion of 98 metabolites were limited, especially the role of MRP1 and BCRP in 99 100 scallon *C* farreri

PAHs could stimulate antioxidant defensive system and lead to 101 oxidative damage through production of ROS during the process of 102detoxification in aquatic organisms (Livingstone et al., 1990; Akcha 103 et al., 2000a; Pan et al., 2009; Bouraoui et al., 2009; Xiu et al., 104 2014). However, there were still few studies concentrating on the 105 106 differences between BaP and CHR, as well as the relationship 107 among detoxification system, antioxidant defensive system, damage effects and bioaccumulation induced by PAHs. It is well known that 108 combining contaminants may lead to synergistic or antagonistic ef-109fects, which were different from individual contaminant (Groten, 110 111 2000; Celander, 2011). Oliveira et al. (2015) reported that the mixtures of BaP, dichlorodiphenyltrichloroethane (DDT) and tributyltin 112 were more toxic to fish Rhamdia quelen than the single. However, 113 there existed an antagonistic effect between BaP and DDT on green 114 115mussel Perna viridis (Song et al., 2016). Although aquatic animals 116had oxidative stress when exposed to BaP or CHR (Xiu et al., 2014; Ren et al., 2014; Ren et al., 2015), the toxic differences of BaP, CHR 117 and their mixtures on scallop C. farreri were not well known. 118

Scallop C. farreri is an important commercial shellfish and is widely 119 120 cultured in the shallow seas of China (Wang et al., 2007). It is also 121 recommended as an indicator in toxicology studies (Pan et al., 2008; Liu, 2009) and environmental monitoring programs (Liu et al., 2012). 06 Furthermore digestive gland is the major site of xenobiotic metabolism 123and biotransformation enzymes in molluscs (Livingstone et al., 1990, 124Livingstone, 1991; Hu et al., 2015a). In the present study, we investigat-07 ed potential toxic effects of BaP, CHR and their mixtures in the digestive 126 gland of scallop C. farreri, including the following endpoints: (a) the ex-127pression levels of detoxification-related genes (AhR, HSP90, ARNT, 128 CYP1A1, CYP1B1, MRP1, BCRP, P-gp), (b) the activities of detoxification 129130enzymes (EROD, GST, UGT, SULT), (c) the effect of antioxidant defense system (SOD, CAT, GPx, GSH), (d) levels of biomolecule damage param-131 eters (F value, LPO levels and PC contents), (e) the bioaccumulation of 132BaP and CHR in digestive gland and soft tissue. The aim of present 133study is to offer the preliminary information on the metabolic pathway, 134135damage effect and bioaccumulation of the single and mixtures of BaP 136and CHR in scallop C. farreri, and provide valuable information for the risk assessments of multiple chemicals. 137

138 **2. Materials and methods**

139 2.1. Chemicals

BaP (CAS#50-32-8) and CHR (CAS#218-01-9) were purchased from
Sigma-Aldrich Chemical Co. (St. Louis, MO, USA). The chemicals for
chemical analysis pretreatment and High Performance Liquid Chromatography (HPLC) detection were of chromatogram grade, obtained
from E. Merck (Darmstadt, Germany). In addition, all the chemicals for
toxic effects analysis were of analytical grade.

2.2. Experimental design

Healthy adult individuals of scallop *C. farreri* ($6.06 \pm 0.7 \text{ cm}$ in shell 147 length of) were obtained from the Shazikou shellfish farm (Yellow Sea, 148 Qingdao, China). All the scallops were acclimated to laboratory 149 condition in aquarium (1 L water per scallop) for seven days in filtered 150 and aerated seawater pumped from Taiping bay before the exposure 151 test. The salinity (30.7 ± 1.1), temperature (19.7 ± 1.0 °C) and pH 152 (8.1 ± 0.8) were measured [average \pm standard deviation (SD)] 153 throughout the experimental periods and following the water 154 exchange. The scallops were fed daily with dried powder of *Spirulina* 155 *platensis* (30 mg/day for each individual). 156

The background concentration of BaP (0.153 ng/L) and CHR 157 (0.139 ng/L) in seawater from Taiping bay was analyzed by HPLC before 158 the experiment. Dimethyl sulfoxide (DMSO) was used as the vehicle of 159 BaP and CHR, and the final concentration of DMSO was 0.001% in all 160 tanks. And the DMSO test had been done in a preliminary experiment 161 with the result that there was no obvious influence of DMSO on scallops. 162 Scallops were divided randomly into seven groups, each of which 163 consisted of three replicate aquarium (n = 100 scallops/aquarium) 164 with 100 L of seawater. One group was used as control (seawater) and 165 each of the remaining six groups was exposed to BaP: 0.1 µg/L 166 (BaP0.1), 1 µg/L (BaP1) and CHR: 0.1 µg/L (CHR0.1), 1 µg/L (CHR1) 167 isolated and in combination (mixtures of BaP0.1 + CHR0.1 or 168 BaP1 + CHR1) for 15 days. The exposure concentrations were based 169 on the concentration of PAHs in the coastal seawater in China, and in 170 order to elicit clearly distinguishable effects, we set two relatively 171 higher concentrations than the environmental concentration (Jin et al., 172 2014; Xiu et al., 2014; Wang et al., 2015). During the experimental 173 period, half of the water was renewed every day, and the concentrations 174 of BaP and CHR were reestablished. 175

Scallops were sampled (three replicates per treatment) at 0, 1, 3, 6, 176 10 and 15 days. Each replicate consisted of fifteen scallops, nine of 177 which were used for BaP and CHR concentration analysis and six of 178 which were used for the measurement of molecular biomarker. The 179 day 0 control samples were selected randomly from each experimental 180 aquarium for no difference between all groups before exposing to contaminants. The digestive gland and soft tissue of scallops were dissected 182 and frozen immediately at -80 °C for subsequent examination.

2.3. Chemical analysis

The analysis of BaP and CHR in tissue was conducted in accordance 185 with the standard method procedures (USEPA, 1996) with some modifications. Freeze-dried scallop tissue sample was Soxhlet extracted with 187 acetone/dichloromethane mixture (1:1) at 70 °C for at least 18 h, and 188 triplicate extracts for each of the scallop tissues were prepared. All 189 extracts were dried using anhydrous sodium sulfate and gently evaporated to 2 mL by a rotary evaporator. The extraction aliquot was 191 added into solid phase extraction column (3 ml PSA, CNW, China) Q8 after the column was preliminarily eluted with 5 mL hexane. Then 193 10 mL mixtures of dichloromethane and hexane (2:3) were eluted 194 thrice to obtain the combined eluate. The eluate was dried with 195 anhydrous sodium sulfate, concentrated by a rotary evaporator, 196 reconstituted with 3 mL acetonitrile and finally filtered through a 197 nylon syringe filter (0.22 µm, 13 mm, ANPEP, China).

The BaP and CHR quantification were analyzed with a high perfor- 199 mance liquid chromatograph (HPLC, Shimadzu, Japan) equipped with 200 two pumps, diode array detector and a reverse-phase C18 column 201 (ZORBAX Eclipse PAH, 4.6×250 mm, 5 µm). The elution temperature 202 was maintained at 34 °C and flow rate was 1 mL/min throughout. The 203 initial mobile phase was 50% acetonitrile for 20 min, followed by 204 20 min linear gradient to 100% acetonitrile and keeping 100% acetoni- 205 trile for 15 min, then decreased to the initial phase in 10 min. 10 µL of 206 concentrated extract were injected for each run. The detection limits 207 for BaP and CHR were 0.01 µg/g d.w. Retention times were defined by 208

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