Accepted Manuscript

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Authors: Mei Li, Gisela Buschle-Diller

PII: S0141-8130(16)32234-6

DOI: http://dx.doi.org/doi:10.1016/j.ijbiomac.2017.03.091

Reference: BIOMAC 7257

To appear in: International Journal of Biological Macromolecules

Received date: 4-11-2016 Accepted date: 16-3-2017

Please cite this article as: Mei Li, Gisela Buschle-Diller, Pectin-blended anionic polysaccharide films for cationic contaminant sorption from water, International Journal of Biological Macromoleculeshttp://dx.doi.org/10.1016/j.ijbiomac.2017.03.091

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Pectin-blended anionic polysaccharide films for cationic contaminant sorption from water

Mei Li¹, Gisela Buschle-Diller^{1*}

¹Department of Biosystems Engineering, Auburn University, AL 36849, USA

Correspondence to: Gisela Buschle-Diller (Email: buschgi@auburn.edu, Tel: 334-844-5468, Fax: 334-844-4068)

Abstract

Substantial amounts of industrial, agricultural, medical and domestic water worldwide are polluted by

different types of contaminants. Removing these contaminants from effluent by adsorbent materials

made from abundant, inexpensive polysaccharides is a feasible and promising approach to deal with

this problem. In the present study, pectin blended with four other types of anionic polysaccharides,

including alginate, carrageenan, xylan and xanthan, were crosslinked with zinc acetate and formed into

thin films. In addition, a negatively charged polyelectrolyte, poly(4-styrenesulfonic acid-co-maleic

acid) sodium salt (PSSMA) was coated on the film surface with the goal of increasing the capture of

cationic contaminants. The average film thickness was measured by a digital micrometer. Surface

morphologies and element analysis were obtained by energy dispersive spectroscopy connected with

scanning electron microscopy. The swelling ratio and the mechanical properties of the films were

investigated in relationship to their composition and PSSMA coating. The sorption of model cationic

pollutants clearly improved for coated films and showed to be predominantly based on the interaction

of positively and negatively charged groups between film/coating and contaminants.

Keywords: pectin, film, decontamination, antiseptic, sorption, anionic polysaccharide.

1. Introduction

Wastewater discharged from industrial, agricultural, medical and domestic activities is a major

contributor to water pollution worldwide. Heavy metals, dyes, pesticides, and fertilizers are among the

most persistent hazardous contaminants [1,2]. In recent years, pharmaceuticals and personal care

product residues have also been detected in water bodies throughout the world [3]. Chemicals, such as

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