

A Quick Reference on Metabolic Alkalosis

Daniel S. Foy, MS, DVM^{a,*}, Helio Autran de Morais, DVM, PhD^b

KEYWORDS

• Bicarbonate • Hypochloremia • Base deficit • Alkalosis • Hypoalbuminemia

KEY POINTS

- Metabolic alkaloses are characterized by an increase in bicarbonate concentration, base excess, and pH, and a compensatory increase in carbon dioxide pressure.
- Normal bicarbonate concentration is approximately 19 to 23 mEq/L in dogs and 17 to 21 mEq/L in cats, whereas normal base excess is approximately 0 to –5 in dogs and cats.
- Hypochloremic alkalosis is the most common cause of metabolic alkalosis and is most commonly secondary to vomiting of stomach contents or diuretic administration.

BICARBONATE AND BASE EXCESS: METABOLIC ALKALOSIS—QUICK REFERENCE

- Metabolic alkaloses are characterized by an increase in HCO_3^- concentration and base excess, increase in pH, and a compensatory increase in PCO_2 (Fig. 1).
 - Metabolic alkalosis is identified by an increase in HCO_3^- concentration or base excess.
 - Base excess is the amount of acid needed to return blood pH to normal. A positive base excess is associated with metabolic alkalosis.
 - There is a compensatory hypoventilation that increases PCO_2 and minimizes the change in pH:
 - In dogs, for each 1 mEq/L increase in HCO_3^- , PCO_2 increases by ~0.7 mm Hg.
 - Respiratory compensation for metabolic alkalosis is similar in cats.
- Metabolic alkalosis can result from increase in the strong ion difference (SID; the difference between all strong cations and strong anions in blood; simplified $\text{SID} = [\text{Na}^+] + [\text{K}^+] - [\text{Cl}^-]$) or decrease in nonvolatile weak acids.

Conflict of Interest: No commercial or financial conflicts of interest; no funding source to declare.

^a College of Veterinary Medicine, Midwestern University, 5715 West Utopia Road, Glendale, AZ 85308, USA; ^b Lois Bates Acheson Veterinary Teaching Hospital, Magruder Hall, Oregon State University, 700 Southwest 30th Street, Corvallis, OR 97331, USA

* Corresponding author.

E-mail address: dfoy1@midwestern.edu

Vet Clin Small Anim ■ (2016) ■–■
<http://dx.doi.org/10.1016/j.cvsm.2016.10.007>

vetsmall.theclinics.com

0195-5616/16/© 2016 Elsevier Inc. All rights reserved.

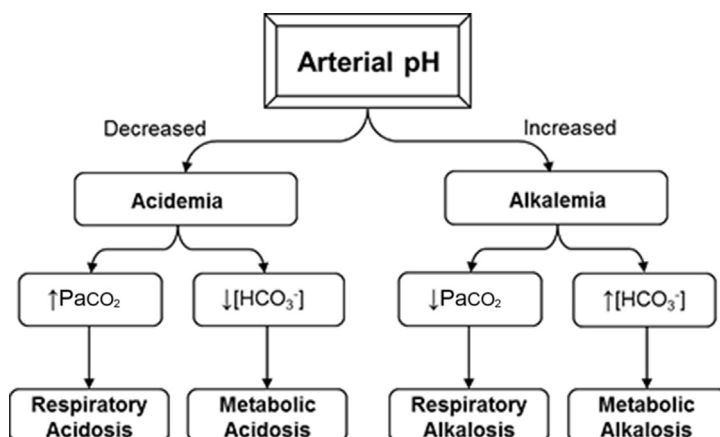


Fig. 1. Algorithm for evaluation of patients with acid-base disorders.

ANALYSIS

- Indications: Measurement of base excess and HCO_3^- concentration is useful in severely ill pets at risk of developing alkalosis (eg, vomiting) or in animals that have a condition known to be associated with metabolic alkalosis or an increase in total carbon dioxide (total CO_2) concentration.
 - A blood gas analysis is necessary to determine if the high total CO_2 is caused by a metabolic alkalosis or a compensation for respiratory acidosis.

Box 1

Principal causes of metabolic alkalosis

SID alkalosis

Concentration alkalosis (recognized by $\uparrow [\text{Na}^+]$)

Pure water loss

Water deprivation

Hypotonic fluid loss

Vomiting

Diarrhea

Hypochloremic alkalosis

Vomiting of stomach content^a

Diuretic administration (loop diuretics and thiazides)^a

NaHCO_3 administration

Chloride-resistant hypochloremic alkalosis

Hyperadrenocorticism

Hyperaldosteronism

Miscellaneous

Marked potassium deficiency

Nonvolatile ion buffer alkalosis

Hypoalbuminemic alkalosis

Liver failure

Protein-losing nephropathy

Protein-losing enteropathy

^a Most important causes in small animal practice.

Download English Version:

<https://daneshyari.com/en/article/5544615>

Download Persian Version:

<https://daneshyari.com/article/5544615>

[Daneshyari.com](https://daneshyari.com)