



Original Research Article (Experimental)

## Evaluation for substitution of stem bark with small branches of *Myrica esculenta* for medicinal use – A comparative phytochemical study

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## ARTICLE INFO

## Article history:

Received 10 May 2016

Received in revised form

4 August 2016

Accepted 20 August 2016

Available online 25 November 2016

## Keywords:

*Myrica esculenta*

Plant part substitution

## ABSTRACT

**Background:** Over exploitation of many traditional medicinal plants like *Myrica esculenta* has become a threat and in the near future, many medicinal plants may be unavailable for use of industry.

**Objective:** Present study outlines the concept of plant part substitution. Stem bark and small branches of *M. esculenta* are compared on the basis of physicochemical analysis, phytochemical analysis, total phenolic contents, total flavonoid contents and high performance thin layer chromatography (HPTLC) to evaluate the possibilities of using small branches in place of stem bark.

**Material and methods:** Physicochemical parameters and preliminary phytochemical screening were carried out using standard methods. Total phenolic and total flavonoid contents were estimated spectrophotometrically using Folin-Ciocalteu and aluminum chloride method, respectively. CAMAG HPTLC system equipped with semi-automatic applicator was used for HPTLC profiling. *n*-Hexane, ethyl acetate and ethanol extracts of stem bark and small branches were developed in suitable mobile phase using standard procedures and visualized in UV 254 and 366 nm and in white light after derivatization within anisaldehyde-sulphuric acid reagent.

**Results:** Phytochemical analysis and HPTLC profile of different extracts showed the presence of almost similar phytochemicals in both stem bark and small branches.

**Conclusion:** Similarities in phytochemical analysis and HPTLC profile of various extracts suggests that small branches may be used in place of stem bark. The study provides the base for further study to use small branches as a substitute of stem bark of *M. esculenta*.

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### 1. Introduction

The importance of medicinal plants has been realized and well documented by physician and scientists since ancient time. Majority of the population in developing countries depend on traditional system of medicine for their primary health care. Over-harvesting of many traditional medicinal plants indirectly poses a risk to the society and has placed many medicinal species at risk of extinction. Commercial exploitation has also sometimes led to traditional medicines becoming unavailable to the indigenous peoples who have relied on them for centuries. In Indian system of medicine the most commonly used medicinal plants are slow-

growing trees, with bark and underground parts being the parts mainly utilized. Collection of underground parts, stem or bark of medicinal plants leads to mass scale uprooting from their natural habitat. This is leading to depletion of plant resources in near future. Due to which plant may be difficult to use in traditional system of medicine in near future. Substitution of the plant or plant parts is therefore the need of the hour for preventing medicinal plants from becoming red listed. It will provide greater scope for practitioners of traditional medicine to utilize herbs that are easily available, cost-effective and most appropriate for the clinical condition. There are a number of possible approaches to solve this problem like establishing conservation areas, enforcing laws against collecting underground parts and bark, large-scale cultivation and use of alternative parts of the same plant such as aerial parts in place of underground parts. Substitution with the part of the same plant is likely to be much better accepted by the patients of traditional healer. Possibilities of substitution of aerial parts with

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Peer review under responsibility of Transdisciplinary University, Bangalore.

underground parts as a strategy for conservation of medicinal plants has been studied by researchers in *Eucomis autumnalis*, *Siphonochilus aethiopicus*, *Ocotea bullata*, *Warburgia salutaris* [1] and *Aegle marmelos* [2].

*Myrica esculenta* (Family: Myricaceae) commonly called Katphala is a medicinal plant widely used in Ayurveda. As per the Ayurvedic literature, stem bark of this plant is used in *gulma* (abdominal tumors), *jvara* (fever), *arsa* (piles), *grahani* (irregular bowel function), *pandu roga* (anemia), *hrillasa* (nausea), *mukha roga* (oral disorders), *kasa* (cough), *svasa* (dyspnea), *agnimandhya* (indigestion), *aruchi* (anorexia) and *kantharoga* (ears, nose, and throat disorders) [3]. Stem bark is also reported for various pharmacological activities like radical-scavenging [4], antioxidant [4,5], anti-diabetic [6], anxiolytic [7], antibacterial [8], anti-helminthic [9], anti-allergic [10], anti-inflammatory [11,12], antimicrobial [12], mast cell stabilizing [13], anti-asthmatic [14].

The stem bark mainly contains gallic acid, castalagin [15], myricanol, myricanone [16–18], epigallocatechin 3-*O*-gallate, epigallocatechin-(4 $\beta$ →8)-epigallocatechin 3-*O*-gallate, 3-*O*-galloylepigallocatechin-(4 $\beta$ →8)-epigallocatechin-3-*O*-gallate, proanthocyanidin, catechine, delphinidine chloride [19], myriconol [20], quercetin,  $\beta$ -sitosterol, taraxerol and triterpene diol [21]. Removal of stem bark from trunk damages the plant. Due to this, availability of the plant may be difficult in near future for use in Indian system of medicine. Present study is carried out in *M. esculenta* to evaluate the possibilities of using small branches in place of stem bark, which will help sustainable utilization.

## 2. Material and methods

### 2.1. Plant material

Stem bark and small branches of *M. esculenta* were collected from Regional Research Institute of Himalayan Folra, Thapala, Ganiadholi, Ranikhet, Distt. Almora (Uttarakhand). Plant material was identified and authenticated by botanist of the Institute, a voucher Specimen (Accession no. 2115) was deposited in Institute.

### 2.2. Instrumentation

A CAMAG HPTLC system (Muttentz, Switzerland) equipped with a semi automatic TLC applicator Linomat IV, twin trough plate development chamber, Win CATS software version 1.4.2. and Hamilton (Reno, Nevada, USA) Syringe (100  $\mu$ l).

### 2.3. Material and reagents

All chemicals, reagents and solvents used during the experimentation were of analytical grade and HPTLC plates were purchased from E. Merck Pvt. Ltd. (Mumbai, India).

### 2.4. Physicochemical parameters

Stem bark and small branches were studied for various physicochemical standards like foreign matter, loss on drying at 105 °C, total ash, acid-insoluble ash, alcohol soluble extractive, water-soluble extractive and pH of 10% aqueous solution using standard methods [22,23].

### 2.5. Preliminary phytochemical screening

*n*-Hexane, ethyl acetate and ethanol extracts of both stem bark and small branches were screened for the presence of phenols, tannins, carbohydrates, saponins, alkaloids, proteins, flavonoids,

steroids, furanoids, coumarins, quinone and triterpenoids by the standards methods of Harborne [24] and Kokate et al. [25].

### 2.6. Estimation of total phenolic and flavonoid content

Five grams of each of shade-dried plant material was pulverized into coarse powder and subjected to ethanolic extraction using Soxhlet apparatus. Extracts were concentrated to dryness. Dried residues were then dissolved in 100 ml of 95% ethanol. Extracts were used for total phenolic and flavonoid assay.

Total phenolics content was determined by using Folin-Ciocalteu assay [26]. An aliquot (1 ml) of extracts or standard solution of gallic acid (20, 40, 60, 80 and 100  $\mu$ g/ml) was added to a 25 ml volumetric flask, containing 9 ml of distilled water. A reagent blank was prepared using distilled water. One milliliter of Folin-Ciocalteu phenol reagent was added to the mixture and shaken. After 5 min, 10 ml of 7% Na<sub>2</sub>CO<sub>3</sub> solution was added to the mixture. Volume was then made up to the mark. After incubation for 90 min at room temperature, absorbance against reagent blank was determined at 550 nm with an UV/Vis spectrophotometer. Total phenolics content was expressed as mg gallic acid equivalents (GAE).

Total flavonoid content was measured by aluminum chloride colorimetric assay [27]. An aliquot (1 ml) of extracts or standard solutions of quercetin (20, 40, 60, 80 and 100  $\mu$ g/ml) was added to a 10 ml volumetric flask containing 4 ml of distilled water. To the flask, 0.3 ml of 5% NaNO<sub>2</sub> was added and after 5 min, 0.3 ml of 10% AlCl<sub>3</sub> was added. After 5 min, 2 ml of 1M NaOH was added and volume was made up to 10 ml with distilled water. Solution was mixed and absorbance was measured against blank at 510 nm. Total flavonoid content was expressed as mg quercetin equivalents (QUE).

### 2.7. HPTLC profiles

HPTLC studies were carried out by following methods of Sethi [28], Stahl [29] and Wagner et al. [30]. Stem bark and small branches were powdered coarsely. Ten gram powdered samples of each of stem bark and small branches were accurately weighed and exhaustively extracted by *n*-hexane, ethyl acetate and ethanol (each 100 ml) separately using Soxhlet apparatus. Extracts were filtered and concentrated under reduced pressure and made up to 10 ml in standard flasks separately.

Mobile phases used for developing the *n*-hexane, ethyl acetate and ethanol extract were toluene: ethyl acetate (5:5 v/v), toluene: ethyl acetate (7:3 v/v) and toluene: ethyl acetate: formic acid (5:5:0.5 v/v), respectively.

Samples were spotted in the form of bands of width 10 mm with a 100  $\mu$ l Hamilton syringe on aluminum TLC plates pre-coated with Silica gel 60 F<sub>254</sub> of 0.2 mm thickness with the help of TLC semi-automatic applicator Linomat IV attached to CAMAG HPTLC system, which was programmed through Win CATS software version 1.4.2. 10  $\mu$ l of each extracts of stem bark and small branches were applied in two tracks as 10 mm bands at a spraying rate of 10 s/ $\mu$ l. Track 1 was stem bark and track 2 was small branches for each of extracts applied.

Development of plate upto a migration distance of 80 mm was performed at 27  $\pm$  2 °C with mobile phase for each extracts in a CAMAG HPTLC chamber previously saturated for 30 min. After development the plate was dried at 60 °C in an oven for 5 min and visualized under wavelength 254 nm and 366 nm for ultraviolet detection. Developed plate was then dipped in anisaldehyde sulphuric acid reagent for derivatization and dried at 105 °C in hot air oven till color of band appears and visualized under white light. Images were captured by keeping plates in photodocumentation chamber and R<sub>f</sub> values were recorded by Win CATS software.

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