



# N<sub>2</sub>O micro-profiles in biofilm from a one-stage autotrophic nitrogen removal system by microelectrode



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## HIGHLIGHTS

- N<sub>2</sub>O production from biofilm in a one-stage completely autotrophic nitrogen removal system was investigated using microelectrode technique.
- The pathways of nitrogen transformation and N<sub>2</sub>O production was characterized by concentration micro-profiles of dissolve oxygen, nitrogen compounds, and N<sub>2</sub>O in the biofilm.
- NH<sub>2</sub>OH oxidation, AOB denitrification, and HD were the pathways for N<sub>2</sub>O production from the biofilm.
- NO<sub>2</sub><sup>-</sup> played a key role in N<sub>2</sub>O production from the biofilm.

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## ABSTRACT

Emission of nitrous oxide (N<sub>2</sub>O), a greenhouse gas, is of growing concern in biological wastewater treatment. N<sub>2</sub>O emission from biofilm in a one-stage completely autotrophic nitrogen removal system was investigated using microelectrodes in this study. It is indicated that the pathways of nitrogen transformation in biofilm mainly included partial nitrification and anaerobic ammonium oxidation (anammox), also included nitrification and heterotrophic denitrification (HD). Ammonium-oxidizing bacteria (AOB) denitrification and HD were the main pathways resulting in N<sub>2</sub>O production in the biofilm, and hydroxylamine (NH<sub>2</sub>OH) oxidation was a subordinate pathway. In addition, the amount of N<sub>2</sub>O emission in test in which both NH<sub>4</sub><sup>+</sup> and NO<sub>2</sub><sup>-</sup> were added (NH<sub>4</sub><sup>+</sup>-N: NO<sub>2</sub><sup>-</sup>-N = 1:1) was about 2 times greater than that in test with NH<sub>4</sub><sup>+</sup> addition only. This result expressed that NO<sub>2</sub><sup>-</sup> is an important factor affecting N<sub>2</sub>O production in the biofilm. In conclusion, the present study provides a theoretical support for reducing N<sub>2</sub>O production in one-stage completely autotrophic nitrogen removal system.

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## 1. Introduction

Methods for reducing the production of nitrous oxide (N<sub>2</sub>O) have become a research focus, because N<sub>2</sub>O has a more than 300-fold stronger effect on global warming than carbon dioxide (CO<sub>2</sub>) and is contributing to ozone layer destruction (Bates et al., 2008). Researchers have found that wastewater treatment plants (WWTPs) are an important source of N<sub>2</sub>O emission, especially those employing biological nitrogen removal processes that involve

nitrification and denitrification (Kampschreur et al., 2009; Sun et al., 2015). To minimize N<sub>2</sub>O emission, it is necessary to identify the main biological pathway of N<sub>2</sub>O emission from WWTPs.

The one-stage completely autotrophic nitrogen removal process combines partial nitrification (PN) and anaerobic ammonium oxidation (anammox) in one reactor. It has remarkable potential for treating wastewater with high-strength ammonium and a low C/N ratio. Because this process requires less aeration and no external organic carbon addition, it is increasingly used in WWTPs (Zhang et al., 2014). However, it is reported that N<sub>2</sub>O emission is considerable from one-stage completely autotrophic nitrogen removal process, because the process is oxygen-limited, lacks carbon sources, and involves multiple nitrogen removal pathways. 2.3% of the nitrogen load that has been found in the full-scale two-reactor nitrification-anammox process is emitted as N<sub>2</sub>O (Kampschreur

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et al., 2008). In a one-stage granular sludge reactor using partial nitrification-anammox, 2.0% of total incoming nitrogen was found to be emitted as  $\text{N}_2\text{O}$ -N (Castro-Barros et al., 2015). In a lab-scale two-reactor PN and anammox process, the average amounts of the influent nitrogen load emitted as  $\text{N}_2\text{O}$  from the PN and anammox reactors were  $4.0 \pm 1.5\%$  and  $0.1 \pm 0.07\%$ , respectively (Okabe et al., 2011).

In the present study, the production of  $\text{N}_2\text{O}$  from biological nitrogen removal processes was studied according to three main pathways: hydroxylamine ( $\text{NH}_2\text{OH}$ ) oxidation (Xiao et al., 2014), ammonium-oxidizing bacteria (AOB) denitrification (Sabba et al., 2015), and heterotrophic denitrification (HD) (Wang et al., 2014).  $\text{N}_2\text{O}$  is a by-product of incomplete  $\text{NH}_2\text{OH}$  oxidation during nitrification, and thus,  $\text{NH}_2\text{OH}$  oxidation plays a role in  $\text{N}_2\text{O}$  production during nitrification (Schreiber et al., 2012; Sabba et al., 2015). AOB denitrification, which uses nitrite ( $\text{NO}_2^-$ ) as a terminal electron acceptor, generates  $\text{N}_2\text{O}$  as final product under oxygen-limited conditions (Ni et al., 2014). Excessive ammonia loading, accumulation of  $\text{NO}_2^-$ , and limited oxygen lead to  $\text{N}_2\text{O}$  production (Chandran et al., 2011). In addition,  $\text{N}_2\text{O}$  is an indispensable intermediate product in HD. An insufficient biodegradable C/N ratio and a higher consumption rate of  $\text{NO}_2^-$  have been shown to result in accumulation of  $\text{N}_2\text{O}$  by HD via the  $\text{NO}_2^-$  pathway (Scaglione et al., 2013; Gabarró et al., 2014). Law et al. (2012) reported that AOB denitrification is the main contributor to  $\text{N}_2\text{O}$  production rather than HD in an aerated environment, whereas Ishii et al. (2014) concluded that HD is a major pathway for  $\text{N}_2\text{O}$  production and that the contribution of HD to  $\text{N}_2\text{O}$  production was relatively greater (20–30%) than that of  $\text{NH}_2\text{OH}$  oxidation and AOB denitrification in a partial-nitrification aerobic granule reactor. However, the source of and factors affecting  $\text{N}_2\text{O}$  production are not clear in the biofilm of the one-stage completely autotrophic nitrogen removal process.

The microelectrode technique, which offers high spatial and temporal resolution, can selectively measure substrate concentrations (Hinzman et al., 2015), and it is also used to determine micro-profiles of the biofilm non-destructively, which is beneficial for analyzing nitrogen compound transformation and  $\text{N}_2\text{O}$  production from biofilm.

This study investigated  $\text{N}_2\text{O}$  production from the biofilm in the one-stage completely autotrophic nitrogen removal process via microelectrode measurement. Several batch experiments were carried out under low oxygen conditions, and the tested biofilm was sampled from a sequencing batch biofilm reactor (SBBR). The transformation pathways of nitrogen compounds in the biofilm were analyzed according to concentration micro-profiles of DO,  $\text{NH}_4^+$ ,  $\text{NO}_2^-$ , and  $\text{NO}_3^-$ . In addition, potential pathways and influencing factors of  $\text{N}_2\text{O}$  production from the biofilm were deduced from the  $\text{N}_2\text{O}$  concentration micro-profiles and the transformation pathways of nitrogen compounds.

## 2. Materials and methods

### 2.1. Operation of SBBR

The one-stage completely autotrophic nitrogen removal process

was carried out in this sequencing batch biofilm reactor (SBBR) with an effective volume of 15 L (Fig. S1). Polyacrylonitrile-activity carbon fiber (PAN-ACF) was used as the biofilm support material. Water-bath heating outside the reactor was used to keep the constant temperature of the SBBR. Magnetic stirrers were used to mix the wastewater. The SBBR was operated at steady-state with continuous aeration. The concentration of DO is  $2.00 \pm 0.20 \text{ mg L}^{-1}$ . Inorganic synthetic wastewater was supplied to the SBBR as the nutrient source for the biofilm and consisted of  $\text{NH}_4\text{HCO}_3$  ( $200 \text{ mg L}^{-1} \text{ NH}_4^+\text{-N}$ ),  $\text{KH}_2\text{PO}_4$  ( $20 \text{ mg L}^{-1} \text{ TP}$ ),  $\text{NaHCO}_3$  ( $330 \text{ mg L}^{-1}$ ), pH adjusted to  $8.0 \pm 0.2$ , and  $2 \text{ mL L}^{-1}$  of a trace element solution that was described previously (Chen et al., 2012). The temperature was kept at  $30 \pm 2^\circ\text{C}$ , and the hydraulic retention time was 24 h. One period of the SBBR contains the influent process about 5 min, reaction process about 23 h, sedimentation process about 50 min, and effluent process about 5 min.

### 2.2. Tests design

Three tests were carried out as described in Table 1 to investigate the nitrogen transformation pathways in the biofilm and elucidate the main pathways and factors responsible for  $\text{N}_2\text{O}$  production from the biofilm. A schematic diagram of the tests system for measuring substrate concentration in the biofilm is shown in Fig. 1. It mainly contains an oxygen and nitrogen gas mixing system, a special 500 mL measurement vessel, and magnetic stirrers, and oxygen meter (YSI), and microelectrodes. Test biofilm samples which were taken from the SBBR, washed with ultrapure water. The washed biofilm was weighted about 20 g (wet weight), and cultured in the special measurement vessel without substrates or aeration approximately 10 h to remove residual  $\text{NO}_2^-$  and  $\text{NO}_3^-$  completely before tests. The concentrations of  $\text{NaHCO}_3$  and trace elements in the tests were the same as those in the SBBR, and temperature was kept at  $30 \pm 2^\circ\text{C}$ . The measurement vessels were flushed with  $\text{O}_2$  and  $\text{N}_2$  mixing gas throughout the tests to maintain the low oxygen condition, and magnetic stirring was used to mix the wastewater. Micro-profiles of DO,  $\text{N}_2\text{O}$ ,  $\text{NH}_4^+$ ,  $\text{NO}_2^-$ , and  $\text{NO}_3^-$  concentrations at 4 h (after substrate addition) were acquired using microelectrodes within the biofilm over the range from  $-200 \mu\text{m}$  to  $750 \mu\text{m}$  (the biofilm surface was at the depth of  $0 \mu\text{m}$ ).

### 2.3. Chemical analyses of the SBBR

To monitor the performance of the SBBR, influent and effluent samples were collected at an interval of 2 h.  $\text{NH}_4^+\text{-N}$ ,  $\text{NO}_2^-\text{-N}$ ,  $\text{NO}_3^-\text{-N}$ , and TN concentrations were measured using standard methods (Nepa, 2002). The DO concentration and temperature (T) were measured using a portable digital dissolved oxygen meter (YSI, Professional ODO™, YSI Co., USA).

### 2.4. DNA analyses

Sludge samples were collected from the biofilm which was taken from different depth in the SBBR. DNA was extracted by 3S DNA centrifugal Kit for environmental samples (Shenergy Biocolor Biotech, Shanghai, China). Illumina Miseq PE300 sequencing was

**Table 1**  
The design of tests.

Test number	$\text{NH}_4^+$ ( $\text{mg L}^{-1}$ )	$\text{NO}_2^-$ ( $\text{mg L}^{-1}$ )	Potential main nitrogen transformation pathways	Purpose
A	100	0	PN, anammox	Analyze $\text{N}_2\text{O}$ production pathways
B	0	100	nitrification, HD	
C	100	100	PN, anammox, nitrification, HD	

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