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2 Effect of lubricant sulfur on the morphology and elemental 2 composition of diesel exhaust particles

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ABSTRACT

This work investigates the effects of lubricant sulfur contents on the morphology, nanostructure, 14 size distribution and elemental composition of diesel exhaust particle on a light-duty diesel 15 engine. Three kinds of lubricant (LS-oil, MS-oil and HS-oil, all of which have different sulfur 16 contents: 0.182%, 0.583% and 1.06%, respectively) were used in this study. The morphologies and 17 nanostructures of exhaust particles were analyzed using high-resolution transmission electron 18 microscopy (TEM). Size distributions of primary particles were determined through advanced 19 image-processing software. Elemental compositions of exhaust particles were obtained through 20 X-ray energy dispersive spectroscopy (EDS). Results show that as lubricant sulfur contents 21 increase, the macroscopic structure of diesel exhaust particles turn from chain-like to a more 22 complex agglomerate. The inner cores of the core-shell structure belonging to these primary 23 particles change little; the shell thickness decreases, and the spacing of carbon layer gradually 24 descends, and amorphous materials that attached onto outer carbon layer of primary particles 25 increase. Size distributions of primary particles present a unimodal and normal distribution, and 26 higher sulfur contents lead to larger size primary particles. The sulfur content in lubricants 27 directly affects the chemical composition in the particles. The content of C (carbon) decreases as 28 sulfur increases in the lubricants, while the contents of O (oxygen), S (sulfur) and trace elements 29 (including S, Si (silicon), Fe (ferrum), P (phosphorus), Ca (calcium), Zn (zinc), Mg (magnesium), 30 Cl (chlorine) and Ni (nickel)) all increase in particles. 31

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46 Introduction

48 Due to their negative impact on health and the environment, 49diesel engine particle emissions have received more and more Q4 attention (Huang et al., 2012; Ris, 2007; Tan et al., 2014). The diesel exhaust particles originate from the combustion of fuels and 51lubricant entering the combustion chamber (Brandenberger et al., 522005; Kleeman et al., 2008; Fraser et al., 2003). Diesel engine 53 exhaust particles consist dominantly of soot particles as a result 54of high temperatures and lack of oxygen in the cylinder, soluble 55organic fraction (SOF) composed of unburned and partially 56burned fuels and lubricant, sulfates formed by the oxidation of 57

sulfur in fuels and lubricant (Tan et al., 2009; Vaaraslahti et al., **Q5Q6** 2005), and ash that includes trace metal elements. 59

The formation and oxidation of particles is a dynamic 60 process in cylinder combustion, cooling of the exhaust and the 61 after-treatment purification in diesel engines (McEnally et al., 62 2006; Shi and Harrison, 1999). The morphology, nanostructure, 63 size distribution and elemental composition of diesel exhaust 64 particles all directly affect oxidation activity (Vander Wal et al., Q7 2003; Boehman et al., 2005; Vander Wal et al., 2007). Vander Q8 Q9 Wal et al. (2003) claimed that the increased oxidation reactivity 67 of particles demonstrated to be related with the higher level 68 of tortuosity of carbon lattice layer and they refer it as a 69

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predominant parameter in determining soot reactivity. Lu et al.
(2012) collected exhaust particles from a medium-duty diesel
engine and suggested that particles between 100 and 180 nm
and 320–560 nm show a more orderly nanostructure; however,
those particles less than 56 nm and within 1000–1800 nm have
a faster oxidation rate.

Many studies have been performed focusing on the effects of 76 fuel properties, lubricant properties, operating conditions, fuel 77 78 injection parameters and after-treatment technologies on the 79 characteristics of diesel exhaust particles. Lapuerta et al. (2012) investigated the effects of fuel on soot oxidation reactivity. 80 Results showed that the biodiesel soot displayed higher reactiv-81 ity due to its smaller primary particles and higher specific active 82 surface, regardless of a higher degree of graphitization. Soewono 010 et al. (2011) studied the morphology and nanostructure of 84 particles in a light-duty diesel engine using ultra-low sulfur fuel 85 and B20 biodiesel. Based on the TEM images, the result shows 86 that the fractal dimensions of the particles measure between 87 1.70 and 1.85, and fractal dimensions are independent of 88 fuel type. It was also illustrated that the B20 particle exhibited a 011 greater structural disorder through a Raman spectroscopy 90 analysis. Virtanen et al. (2004) evaluated the impact of engine 012 load on soot particle size distributions and morphology. The 92 93 number distribution follows the lognormal distribution, and the width of the distribution increased with the load. Simultaneous-94 95 ly the fractal dimension of soot particles varies from 2.6 to 2.8 in 96 dependence of engine load. Liati et al. (2013a) collected soot from 97 different sites of the engine's exhaust after-treatment system to investigate reactivity in terms of the morphology and structure 98 of primary soot particles. Results showed that primary soot 99 particles that passed through the diesel particle filter (DPF) and 100 entered the atmosphere showed a higher degree of graphitiza-101 tion than those entering the DPF, leading to lower reactivity. 102

103 Considering that the exhaust particles from diesel engines originate from combustion of fuels and lubricants entering the 104 combustion chamber, fuel and lubricant properties are critical 105factors affecting particle emissions. With fuel properties improv-106 ing (ultra-low sulfur content) and diesel particle emission 107 regulations tightening (Johnson, 2015), lubricant plays an increas-108 ingly important role in contributing to the particle emission 109lubricant (Miller et al., 2007). Many investigations have been 110 111 performed on the contribution of lubricant on particle emissions. 112Kyotö et al. (2002) suggested that lubricant can increase the particle number emissions and can provide low particle mass 113emissions. Storey et al. (2015) showed that the contribution of 114 lube oil to the total mass of the PM is on the order of 1%. Plumley 013 (2005) compared the PM emission form mineral-based motor oils 116and full synthetic (PAO) base oil on a three-cylinder direct 117 injection engine. Results showed that the specific particulate 118 emissions of synthetic oil were 19-24% smaller than those of the 119 120mineral oils. Wang et al. (2014) used neat fuel and blended fuel containing oil pour point depressant (PPD) additive to study the 121 122 effects of lubricant additives on particle emission. Results show that the layer fringe length decreases from 1.191 nm to 1.064 nm, 123124 while both the separation distance and tortuosity increase.

The effects of lubricant properties on diesel particle emissions become more important; this is particularly true of lubricant sulfur content (Ronkkö et al., 2013). Although fuel's sulfur content is as low as 2 ppm, there is a large quantity of sulfur-associated q14 nanoparticles in diesel engine exhaust (Vaaraslahti et al., 2004). Sulfur can be converted into sulfuric acid aerosols that are an 130 important component of particle through combustion (Cornelius 131 et al., 1993). He et al. (2015) showed that sulfuric acid oxidized 132 from SO₂ made a negligible contribution to the growth of >10 nm 133 new particles. Tornehed et al. (2012) summarized that the **Q15** sulfur-associated particulate emissions increase by a maximum 135 of approximately 0.15 g for every gram of sulfur in the lubricant. 136

This article will study the effects of lubricant sulfur content 137 on the morphology, nanostructure, size distribution and ele-138 mental composition of exhaust particles from diesel engines. 139 Transmission electron microscopy (TEM) was used to observe 140 the morphology and nanostructure of diesel exhaust particles, 141 and advanced image-processing software was applied to analyze 142 the size distributions of the primary particles. Elemental 143 compositions of exhaust particles were obtained through X-ray 144 energy dispersive spectroscopy (EDS). This study can provide 145 useful information regarding the future of lubricant formulation. 146

1. Materials and methods

1.1. Test engine, fuel and lubricant

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This study used a light-duty, direct injection and four-stroke 150 diesel engine with a rated power of 24.6 kW at 2400 r/min and 151 a maximum torque of 114 N·m at 1600 r/min. The engine has 152 two cylinders with a compression ratio of 17.1, and a 153 displacement volume of 2.22 L. The engine was manufactured 154 by Fujian Lijia Co., Ltd. in China with a model number of 155 SL2110ABT and met China IV emission standards. To mini-156 mize any fuel effects we used an ultra-low sulfur fuel with a 157 sulfur content of less than 10 ppm. The primary specifications 158 of the test fuel are shown in Table 1.

Three kinds of lubricant with different sulfur contents were 160 used in this study. The first lubricant is Castrol oil, which has 161 the lowest sulfur content (0.182%) of the three lubricants, meets 162 the ACEA (European Automobile Manufacturers' Association) 163 C1 grade oil standard (The standard is global strictest lubricant 164 standard for light-duty diesel engines). It is suitable for EURO VI 165 light-duty diesel engines to ensure high DPF performance as a 166 catalyst-compatible and stay-in-grade oil. Now, China has a 167 mandatory regulation for diesel lubricants, which does not limit 168 the sulfur content in lubricants. Therefore, the sulfur content in 169 China's current lubricant is usually far higher than what is found 170 in the ACEA C1 grade oil standard. Based on the preceding ACEA 171 C1 lubricant (called as LS-oil), other two sulfur content lubricants 172 were obtained by adding sulfurous additives (T-323/Amino 173

Table 1 – Test fuel specifications.				t1.1
Property	Unit	Method	Parameters	$^{t1.3}_{t1.4}$
Density @ 20 °C	kg/m ³	ASTM D4052	821.9	t1.5
Cetane number		ASTM D613	52.3	t1.6
Sulfur	ppm, m/m	ASTM D5453	<1	t1.7
Flash point	°C	ASTM D93	92.0	t1.8
Distillation at 90% volume	°C	ASTM D86	323.1	t1.9
Kinematic viscosity @ 40°C	mm²/sec	ASTM D445	4.54	t1.10
Lower heating value	MJ/kg	ASTM D240	43.96	t1.11

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