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Determination of the Henry's law constants of low-volatility compounds via the measured air-phase transfer coefficients

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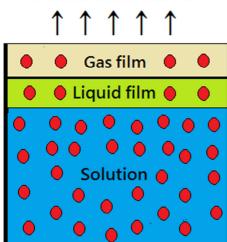
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Volatilization into air



Henry's law constant (H)

 $H = K_L/k_G$ for low-H solutes

 K_L = Volatilization coef.

 $k_{\rm G}$ = Gas-phase transfer coef.

 $k_{\rm G} = \beta (\rm RT/2\pi M)^{1/2}$

where ß is a constant for all vapors at a given level of air dynamics

● low-*H* solutes (*H* <10⁻³)

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