



## Valorization of apple tree wood residues by polyphenols extraction: Comparison between conventional and microwave-assisted extraction



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### ABSTRACT

For the first time, the characterization of antioxidant activity and phenolic profile of apple tree (*Malus domestica*) bark, core and roots was carried out. Phenolic compounds were extracted from the Belgium apple tree wood residues collected at two seasons, namely summer 2015 and winter 2016, using conventional (CE) and microwave-assisted extraction (MAE) techniques. For each extraction technique, the influence of the most important operational parameters, namely solvent composition, extraction time and temperature, on the total phenolic and flavonoid content, and antioxidant activity by the 2,2-diphenyl-1-picrylhydrazyl radical scavenging activity (DPPH-RSA) and ferric reducing activity power (FRAP) assays were optimized. The phenolic profile from the obtained extracts was also characterized by high-performance liquid chromatography with photodiode array detection (HPLC-PDA). Optimum conditions were: 20 mL ethanol:water 60:40 v/v, 20 min, 100 °C, sample weight 0.1 g for MAE and 20 mL ethanol:water 50:50 v/v, 2 h, 55 °C, sample weight 0.5 g for CE. Root extracts obtained by MAE (the most efficient technique) presented the highest phenolic ( $47.7 \pm 0.9$  mg gallic acid equivalents/g dry weight) and flavonoid ( $17.1 \pm 0.8$  mg epicatechin equivalents/g dry weight) content, and antioxidant activity ( $28.4 \pm 2.0$  mg trolox equivalents/g dry weight and  $36.1 \pm 2.7$  mg ascorbic acid equivalents/g dry weight for DPPH-RSA and FRAP assays, respectively), followed by bark and core wood extracts. HPLC-PDA analysis revealed that phloridzin was the main contributor to the phenolic composition representing 52%–87% of the total amount of phenolic compounds quantified, while phenolic acids represents less than 10%. This study reveals the potential of apple tree wood residues valorization through the recovery of phenolic compounds for food, pharmaceutical and cosmetic applications.

### 1. Introduction

Polyphenols are one of the main groups of secondary plant metabolites, essentially for their normal growth and defense against infection and injury, and their health benefits have been extensively described (El Gharras, 2009; Ghitescu et al., 2015; Stevanovic et al., 2009). In the last few years, search of inexpensive and renewable sources of polyphenols has been attracting researchers interest. For that reason, the number of publications concerning the extraction of these compounds from biomass has been increasing (Bouras et al., 2015; Ghitescu et al., 2015; Hofmann et al., 2015; Lazar et al., 2016).

Every year, Belgium apple farmers renew 6% of the apple plantation, which reflects in the annual production of 30,000 ton of woods residues (FAOSTAT, 2015). Traditionally, these apple tree residues are used in low added value applications, such as firewood or dispersed (Dedrie et al., 2015; Ghitescu et al., 2015). Still, in last few years, tree materials, such as bark, have been emerging as possible sources of valuable compounds (Table 1) (Ghitescu et al., 2015; Hofmann et al., 2015). To this regard, the recovery of phenolic compounds from these wood wastes is gaining considerable attention, especially ascribable to the antioxidant properties that these compounds exert (Kammerer et al., 2014; Stevanovic et al., 2009).

**Abbreviations:** AA, ascorbic acid; CE, conventional extraction; CV, coefficient variation; DPPH-RSA, 2,2-diphenyl-1-picrylhydrazyl radical scavenging activity; DW, dry weight; FRAP, ferric reduction activity power; GA, gallic acid; GAE, gallic acid equivalents; HPLC-PDA, high performance liquid chromatography with photodiode array detection; MAE, microwave assisted extraction; RSM, response surface methodology; TE, trolox equivalents; TPC, total phenolic content; TPTZ, 2,4,6-tris(2-pyridyl)-s-triazine; Trolox, 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid; UPLC-DAD, ultra-performance liquid chromatography–diode array detection

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**Table 1**  
Summary of the published reports on the extraction of phenolic compounds from tree wood residues.

Sample	Extraction technique and conditions	TPC <sup>a</sup>	TPC units	Reference
Birch ( <i>Betula pendula</i> ) and pine ( <i>Pinus sylvestris</i> ) bark varieties Port Orford cedar ( <i>Chamaecyparis lawsoniana</i> (A. Murr.) Parl.) bark	Stirring: 500 mg, 2 × 10 mL of methanol:water 80:20 v/v, 1 min SE <sup>d</sup> : 100 g, 1 L of methanol	2.0 ± 0.1–76.0 ± 2.9 77.7–88.8	mg GAE <sup>b</sup> /g DW <sup>c</sup> mg GAE/g DW	Kähkönen et al. (1999) Gao et al. (2007)
Pine ( <i>Pinus rigida</i> × <i>taeda</i> and <i>Pinus koraiensis</i> ) bark Bark of <i>Phyllanthus emblica</i> L. Oak ( <i>Prunus avium</i> and <i>Quercus robur</i> ) wood from different countries	10 g, 100 mL boiling water, 1 h MAE: 3 g, 75% aqueous ethanol, 25 min, 45 °C 500 mg of sawdust, 30 mL methanol, 30 min, RT <sup>f</sup>	111–862 19.8 6.95 ± 0.37– 101.2 ± 1.2 412 ± 0 479 ± 49 388 ± 7 622 ± 40	mg CE <sup>g</sup> /g extract % µg GAE/g oak wood mg CE/g bark	Ku et al. (2007) Yang et al. (2009) Alañón et al. (2011) Aspé and Fernández (2011)
<i>Pinus radiata</i> bark	Maceration: 20 g, 200 mL acetone:water 70:30 v/v, 40 °C MAE <sup>h</sup> : 20 g, 200 mL acetone:water 70:30 v/v, 900 W, 2450 MHz UAE <sup>h</sup> : 20 g, 200 mL acetone:water 70:30 v/v, 35 KHz, 85 W, RT SE: 20 g, 200 mL acetone:water 70:30 v/v, 82 °C 100 g, 2 L ethanol:water 3:1 w/w, 120 min, 120 °C	0.55 ± 0.01 43.2 ± 1.2 3.7 ± 0.6 11.7 ± 0.5 16.50 ± 0.07 54.2 ± 19.4	g GAE/g extract mg GAE/g mg GAE/g bark mg GAE/g bark	Boclandro et al. (2012) Lamoumier et al. (2012) Bouras et al. (2015)
<i>Pinus radiata</i> bark Bark of <i>Maclura tinctoria</i> (L.) <i>Quercus</i> ( <i>Q. robur</i> L.) bark	Stirring: acetone:water 70:30 v/v, 30 min, RT + 10 min at 85 °C Maceration: 5 g, 100 mL water, 120 min, 25 °C Heat reflux: 5 g, 100 mL water, 120 min, 100 °C MAE: 5 g, 100 mL water, 120 min, 100 °C, 400 W Maceration: 1 g, 9 mL of water + 1% NaOH + 0.25% Na <sub>2</sub> SO <sub>3</sub> + 0.25% NaHSO <sub>3</sub> , 2 h, 70 °C			Chupin et al. (2015)
French maritime pine bark ( <i>Pinus pinaster</i> )	MAE: 1 g, 10 mL ethanol:water 80:20 v/v, 3 min, 100 W Hydrodistillation: 100 g, 1 L water, 180 min Solvent free MAE: 100 g, 92.4 min, 803.5 W	28.3 ± 2.9 14.3 139.2		Mellouk et al. (2015)
Oak bark <i>Quercus robur</i> L. and <i>Quercus petraea</i> Bark of <i>Solidago canadensis</i> L.	SE: 1 g, 70 mL with 2 successive extractions with water and ethanol, 6 h Stirring: 1 g, 20 mL 50% aqueous ethanol, 25 °C, 30 min UAE: 5 g, 100 mL 50% aqueous ethanol, 30 min HPE: 1 g, 20 mL 50% aqueous ethanol, 25 °C, 500 MPa, 10 min UAE: 10 g, 100 mL ethanol:water 70:30 v/v, 54 °C, 60 min, 35 KHz, 320 W Stirring: 0.15 g, 15 mL ethanol:water 80:20 v/v, 5 h, RT Sonication: 0.15 g, 15 mL ethanol:water 80:20 v/v, 10 min, RT MAE: 0.15 g, 15 mL ethanol:water 80:20 v/v, 20 min, 120 °C Stirring: 1.0 g, 20 mL methanol:water 80:20 v/v, 2 h, 20 °C Infusion: 1.0 g, 20 mL methanol:water 80:20 v/v, 2 h, 60 °C MAE: 1.0 g, 20 mL methanol:water 80:20 v/v, 5 min, 150 W MAE: 1.0 g, 20 mL ethanol, 5 min, 150 W MAE: 1.0 g, 20 mL water, 5 min, 150 W UAE: 5 g, 50 mL ethanol:water 70:30 v/v, 60 °C, 35 KHz, 320 W	5.0 ± 0.3–13.4 ± 0.2 0.601 ± 0.041–1.584 ± 0.020	mg GAE/g DW mg GAE/g	Dedrie et al. (2015) Deng et al. (2015)
Spruce bark Beech bark ( <i>Fagus sylvatica</i> L.)		13.2 48.3 ± 1.2 49.9 ± 1.1 65.2 ± 5.6 353.6 ± 0.5 258.4 ± 3.6 441.6 ± 0.3 279.7 ± 8.1 298.8 ± 0.6 43.1	mg GAE/g mg QE/g DW	Ghitescu et al. (2015) Hofmann et al. (2015)
Moroccan <i>Acacia mollissima</i> bark			mg GAE/g bark	Naima et al. (2015)
Spruce bark ( <i>Picea abies</i> )			mg GAE/g	Lazar et al. (2016)

<sup>a</sup> TPC: total phenolic content.

<sup>b</sup> GAE: gallic acid equivalents.

<sup>c</sup> DW: dry weight.

<sup>d</sup> SE: Soxhlet extraction.

<sup>e</sup> CE: catechin equivalents.

<sup>f</sup> RT: room temperature.

<sup>g</sup> MAE: microwave assisted extraction.

<sup>h</sup> UAE: ultrasound assisted extraction.

<sup>i</sup> HPE: high hydrostatic pressure-assisted extraction.

<sup>j</sup> QE: quercetin equivalents.

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