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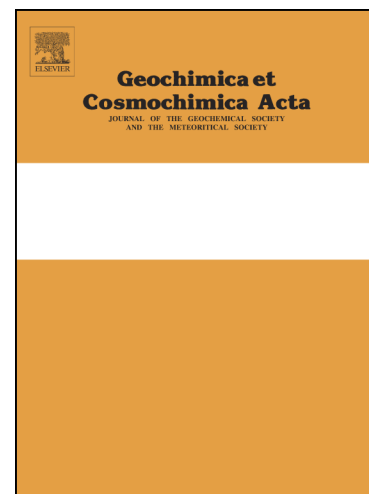
Fe(III):S(-II) Concentration Ratio Controls the Pathway and the Kinetics of Pyrite Formation during Sulfidation of Ferric Hydroxides

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Fe(III):S(-II) Concentration Ratio Controls the Pathway and the Kinetics of Pyrite Formation during Sulfidation of Ferric Hydroxides

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Key words

rapid pyrite formation, pathway, polysulfides, reactive ferrous iron, iron-sulfur interaction, ferric hydroxides, Mössbauer spectroscopy

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