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On the minimum ignition energy (MIE) for propane/air

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1. Introduction

It seems that for more than half a century the classical MIE values provided by Lewis and von Elbe [1] for a wide range of mixtures of combustible gases and air have been widely used as standard reference values. For propane/air the value is 0.25 mJ. The numerous U-shaped curves for MIE as a function of mixing ratio of combustible gas/air published by these workers have gained wide general acceptance.

The present investigation came about as a result of an attempt to develop an electric spark generator for determination of MIEs of very ignition-sensitive dust clouds in air in the <1 mJ range. Because the experimental dust clouds used were transient, produced by a short blast of air, the appearance of the electric spark had to be synchronized with the appearance of the dust cloud. Therefore, a means of synchronization had to be incorporated into the spark generation circuit. The details of the first version of the generator developed for this purpose were published by Randeberg et al. [2]. The first MIE results were presented by Randeberg and Eckhoff [3]. However, as discussed by Eckhoff et al. [4], it appeared that Randeberg et al. [2] significantly underestimated the real spark energies produced by their circuit because of a previously undetected subtle additional energy supply to the spark channel before onset of the ignition process. In view of this finding it was necessary to redesign the spark generator in order to minimize and control this additional energy supply to the extent practically possible. In order

ABSTRACT

A copy of the standard ASTM spark generator for determination of MIEs of gases and vapours was built and measurements to determine MIE of propane/air at normal atmospheric conditions were performed. However, the ASTM standard does not prescribe any statistical procedure for deriving MIE values from primary test data. We therefore adopted the "highest-possible-border-line" procedure proposed by Moorhouse et al. in 1974, and obtained a MIE of 0.48 mJ, which is very close to the 0.46 mJ found by these workers, as opposed to the classical Lewis and von Elbe value of only 0.25 mJ. One possible reason for the discrepancy could be the very low ignition probability of only 1% used by Lewis and von Elbe as their MIE criterion. However, when applying both linear and logistic regression analysis to our experimental data, the spark energies yielding 1% probability of ignition were found to be 0.40 ± 0.06 and 0.45 ± 0.08 mJ, respectively, which are both significantly higher than 0.25 mJ. This may indicate that the classical MIE values for gases and vapours published by Lewis and von Elbe (1961) are perhaps unnecessarily conservative.

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to calibrate the re-designed circuit for generation of synchronized sparks, using premixed propane/air, it was decided to first measure MIE of premixed propane/air by means of the spark generator prescribed in the ASTM [5] standard, and then subsequently conduct similar experiments using the synchronized spark generator, and compare the results.

However, it then appeared that the measurements by means of the ASTM standard spark generator produced an MIE for premixed propane/air that differed significantly from the classical US Bureau of Mines value. This was considered of sufficient interest in itself to justify a closer investigation.

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2. The investigation of Moorhouse et al.

Moorhouse et al. [6] determined the MIEs for C_1 to C_7 hydrocarbon/air mixtures. On the whole their values were significantly higher than those reported by Lewis and von Elbe [1]. Because we were already working with propane/air in other contexts, we selected this mixture for a closer study. The MIE for propane/air at normal atmospheric pressure and temperature found by Moorhouse et al. is 0.46 mJ, whereas the value of Lewis and von Elbe [1] is 0.25 mJ. Moorhouse et al. pointed out that Lewis and von Elbe had adopted a very low ignition probability of only 1% as their MIE criterion, and suggested that this could be the reason for the discrepancy. The alternative approach adopted by Moorhouse et al. was to draw the highest possible border line through the array of

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Fig. 1. Block diagram of the expanding-plate-capacitor electric spark generator used by Moorhouse et al. [6].

experimental ignition/no ignition points versus propane/air mixing ratio, below which there were no ignitions ("a boundary below which no ignitions were observed"). The minimum value of this, mostly U-shaped, border line was taken as the MIE.

Moorhouse et al. obtained their electric sparks by means of the variable-air-capacitor circuit developed by Cheng [7], which differs from the ASTM standard circuit used in the present investigation. Their circuit is illustrated in Fig. 1.

This is basically a simple RLC circuit with $L \approx 1 \,\mu\text{H}$. The external circuit resistance was negligible compared to the average spark resistance. The capacitances were in the range 19–64 pF. and typical spark signatures were damped oscillations of about 10 MHz and 1 μ s duration. The capacitor consisted of two 100 mm \times 200 mm metal plates, one fixed, the other movable. The fixed plate was connected to the high-voltage electrode in the explosion vessel whereas the movable plate was connected to the other, earthed electrode. When the fixed plate had been charged from a highvoltage source, the spring loaded movable plate was released to become displaced rapidly outwards from its closest distance from the fixed plate (about 1 mm). This caused a decrease of the capacitance and a corresponding increase of the voltage across the capacitor, and hence across the spark gap. A discharge between the electrodes occurred as soon as the voltage had risen to the breakdown voltage of the spark gap.

During plate separation the voltage *U* across the plates will increase and the capacitance *C* of the plate system decrease according to

$$U_1 C_1 = U_0 C_0 \to U_1 = \frac{U_0 C_0}{C_1} \tag{1}$$

where the index 0 refers to the situation before plate separation and the index 1 to the situation at spark breakdown at some point during separation. By measuring the position of the movable capacitor plate at the moment of spark-over by a displacement transducer, the capacitance C_1 at spark-over can be calculated by simple basic theory. Then, if the residual energy left on the capacitor after sparkover is negligible in comparison to the spark energy, the latter equals

$$E = \frac{1}{2}C_1 U_1^2$$
 (2)

which, by inserting Eq. (1) in (2), contains only known quantities as follows:

$$E = \frac{1}{2} \frac{U_0^2 C_0^2}{C_1} \tag{3}$$



Fig. 2. Electric-spark-ignition diagram for n-pentane at $22 \degree C$ and 1 bar(g). Electrode separation: 2.2 mm; (\bullet) non-ignition; (\bigcirc) ignition. From Moorhouse et al. [6].

Moorhouse et al. determined their MIEs by varying the fuel/air ratio and the spark energy over wide ranges until a borderline between ignition and no-ignition could been identified. Fig. 2 illustrates the method. In the case shown, with pentane/air at atmospheric pressure, the minimum of the U-shaped curve is about 0.5 mJ, and hence this was taken as the MIE of pentane/air at atmospheric pressure and ambient temperature. For propane/air Moorhouse et al. [6] found the value 0.46 mJ, when using the same data reduction procedure as illustrated in Fig. 2.

3. Experimental apparatus and procedures

The block diagram of the standard ASTM [5] electric spark discharge circuit used in the present investigation is shown in Fig. 3. An isolating resistor of at least $1 T\Omega$ is required to ensure a sufficiently large time constant for charging of the comparatively small energy storage capacitors. The spark gap voltage at spark-over was measured by means of an electrostatic voltmeter. With a simple electrostatic voltmeter, a separate divider resistor of at least $100 T\Omega$ is required. In our case the voltmeter contained an integrated field mill representing an internal DC resistance of even more than $100 T\Omega$. Hence, our external divider resistor was only $330 k\Omega$. Our voltmeter contributed a small capacitance of about 2 pF in parallel with the main energy storage capacitor, which had to be accounted for when calculating the spark energy.

A cross-section of our explosion vessel is shown in Fig. 4. The vessel is cylindrical and made of hard transparent plastic. The internal diameter is 40 mm and the height 156 mm, corresponding to a net internal volume of 200 cm³. A locking ring for fixing a paper cover is placed at the top of the vessel. Two 1.6 mm flat-ended tungsten electrodes fitted with glass flanges are located at the centre of the tube. One electrode is fixed, the other adjustable, allowing adjustment of the gap distance between the electrodes. The circular flanges of borosilicate glass have a diameter of 15 mm and a thickness of 3.0 mm. The flanges are fastened to the electrodes with Araldite.





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